TEXAS WATER DEVELOPMENT BOARD

REPORT 129

RECONNAISSANCE OF THE CHEMICAL QUALITY OF SURFACE WATERS OF THE RED RIVER BASIN, TEXAS

Ву

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Prepared by the U.S. Geological Survey in cooperation with the Texas Water Development Board

TEXAS WATER DEVELOPMENT BOARD

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TABLE OF CONTENTS

	Pag
ABSTRACT	1
INTRODUCTION	3
RED RIVER DRAINAGE BASIN	3
General Description	3
Population and Municipalities	5
Agricultural and Industrial Development	9
Development of Surface-Water Resources	9
CHEMICAL QUALITY OF THE WATER	9
Chemical-Quality Records	9
Streamflow Records	10
Environmental Factors and Their Effects on the Chemical Quality of the Water	10
Geology	10
Streamflow	12
Activities of Man	12
Relation of Quality of Water to Use	17
Domestic Use	17
Industrial Use	19
Irrigation	19
	111273
Geographic Variations in Water Quality	19
Dissolved Solids	21
Chloride	22
Hardness	22
Other Constituents	22
Water Quality in Reservoirs	22
Buffalo Lake	22

TABLE OF CONTENTS (Cont'd.)

		raye
	Bivins Lake	22
	Baylor Creek Reservoir	23
	Greenbelt Reservoir	23
	Lake Kemp and Diversion Lake	23
	Santa Rosa Lake	23
	North Fork Buffalo Creek Reservoir	23
	Lake Wichita	23
	Lake Kickapoo	23
	Lake Arrowhead	23
	Farmers Creek Lake	23
	Hubert H. Moss Lake	23
	Lake Texoma	23
	Lake Randall	25
	Brushy Creek Reservoir and Coffee Mill Creek Lake	25
	Pat Mayse Reservoir	25
	Lake Crook	29
Wate	r Quality at Potential Reservoir Sites	29
	Mackenzie	29
	Buck Creek	29
	Lelia Lake Creek	29
	Dozier Creek	29
	Lower McClellan Creek	29
	Sweetwater Creek	29
	Ringgold	29
	Timber Creek and Bois d'Arc Creek	29
	Big Pine	29
	Pecan Bayou	29
	Barkman Creek	29

TABLE OF CONTENTS (Cont'd.)

		Pag
	Present and Future Water-Quality Problems	30
SEL	ECTED REFERENCES	31
	TABLES	
	TABLES	
1.	Reservoirs in the Red River Basin in Texas Having a Capacity of 5,000 Acre-Feet or More	11
2.	Source and Significance of Dissolved-Mineral Constituents and Properties of Water	18
3.	Water-Quality Tolerances for Industrial Applications	20
4.	Index of Surface-Water Records in the Red River Basin, Texas	34
5.	Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas	38
6.	Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas	45
7.	Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma	62
	FIGURES	
1.	Index Map Showing River Basins in Texas and Physiographic Sections of the Red River Basin	4
2.	Map and Graphs Showing Precipitation and Runoff	7
3.	Geologic Map and Chemical Composition of Low-Flow of Streams	13
4.	Map Showing Location of Natural Brine Emissions and Areas of Petroleum Production	15
5.	Diagram for Classification of Irrigation Waters	21
6.	Duration Curve of Dissolved Solids for Red River Near Gainesville, Texas, 1953-63	21
7.	Graph Showing Dissolved-Solids Content and Quantity of Water in Lake Texoma, 1945-67	24
8.	Map Showing Sampling Sites in Lake Texoma	26
9.	Vertical Profiles of Specific Conductance, Dissolved Oxygen, and Temperature of Lake Texoma	27
10.	Longitudinal Profiles of Lake Texoma Showing Water Quality, July 25-27, 1967	28

TABLE OF CONTENTS (Cont'd.)

		Page
11.	Map Showing Location of Streamflow and Chemical-Quality Data-Collection Sites, Major Existing Reservoirs, and Potential Reservoir Sites in Texas	69
12.	Map Showing Dissolved-Solids Concentrations of Surface Water	71
13.	Map Showing Chloride Concentrations of Surface Water	73
14.	Map Showing Hardness of Surface Water	75

RECONNAISSANCE OF THE CHEMICAL QUALITY OF SURFACE WATERS OF THE RED RIVER BASIN, TEXAS

ABSTRACT

The Red River, from its point of origin in eastern New Mexico to the northeast corner of Texas, drains an area of about 48,000 square miles. The total area in Texas is 24,500 square miles. From west to east the topography changes from the nearly flat surface of the High Plains, to a gently eastward-sloping plain dissected by prominent systems of drainage in the Osage Plains, to the low relief and gently gulfward slope of the West Gulf Coastal Plain.

The climate of the basin ranges from semiarid to humid; mean annual precipitation is less than 18 inches in the far western part and more than 46 inches in the extreme eastern part. Runoff increases from about 50 acre-feet per square mile at the 100th meridian to more than 800 acre-feet per square mile in the northeast corner of the State.

The dissolved-mineral content and chemical character of waters in the Red River basin vary widely from place to place and from time to time. Geologic factors, runoff and streamflow characteristics, and activities of man largely determine the nature and amount of dissolved material transported by the Red River and its tributaries. In the semiarid western part of the basin, base flow is usually nonexistent. However,

numerous seeps and springs in Permian rocks that crop out in this part of the basin account for much of the salt load in the Red River above Lake Texoma. The water quality of the main stem has been further degraded by oil-field brines. The eastern part of the basin is in an area of high rainfall and well-leached rocks and soils. Ground-water effluent is generally low in dissolved minerals, and the dissolved-solids content of streamflow varies only slightly with discharge.

The highly mineralized waters from salt sources in the western part of the basin cause the water of the Red River to be undesirable for public supply throughout most of its reach in Texas. Storage of good-quality water in existing and proposed reservoirs on tributaries to the Red River will increase degradation of water quality in the main stem, especially above Lake Texoma. Even if releases are made from tributary impoundments in the western part of the basin, evaporation during impoundment and waste water from various uses of the reservoir waters will degrade the tributary waters entering the main stem. For any plan to be effective in the improvement of water quality of Red River throughout its reach in Texas, large amounts of natural brine must be prevented from entering water courses of the basin.

RECONNAISSANCE OF THE CHEMICAL QUALITY OF SURFACE WATERS OF THE RED RIVER BASIN, TEXAS

INTRODUCTION

The investigation of the chemical quality of the surface waters of the Red River basin, Texas, is part of a statewide reconnaissance. Each major river basin in the State is being studied and a report is being prepared to present the results of the study and to summarize the available chemical-quality data. Reports that have been published are included in the list of references.

The purpose of this report is to summarize information on the quality of surface water in the Red River basin, and to present it in a form that will aid in the proper development, control, and use of water resources of the area. In the study, the following items were considered: the nature and amounts of mineral constituents in solution; the geologic, hydrologic, and cultural influences that determine water quality; the amount and probable source of the salt transported by streams; and the suitability of the water for domestic, industrial, and agricultural uses. Data for the Oklahoma part of the Red River basin are included to show the effect of runoff from Oklahoma on the chemical quality of water in the mainstem Red River.

A network of daily chemical-quality stations on principal streams in Texas is operated by the U.S. Geological Survey in cooperation with the Texas Water Development Board and with federal and local agencies. This network has not been adequate to inventory completely the chemical quality of the surface waters of the State. To supplement the information being obtained by the network, a cooperative statewide reconnaissance by the U.S. Geological Survey and Texas Water Development Board was begun in September 1961. In this reconnaissance, samples for chemical analyses have been collected periodically at numerous sites throughout the State so that some quality-of-water information would be available for locations where water-development projects are likely. These data aid in the delineation of areas having water-quality problems and in the identification of probable sources of pollution, and thus indicate areas where more detailed investigations are needed.

During the period September 1961 to September 1967, water-quality data were collected on the principal streams, on the major reservoirs, at a number of potential reservoir sites, and on many tributaries in the basin.

Quality-of-water information for the Oklahoma part of the Red River basin was collected by the U.S. Geological Survey in cooperation with the Oklahoma Water Resources Board. Water-quality data in Texas and Oklahoma have been collected also by the U.S. Public Health Service and the Federal Water Quality Administration.

Agencies that have cooperated in the collection of water-quality and streamflow data include the U.S. Army Corps of Engineers, the Texas State Department of Health, and the city of Wichita Falls.

RED RIVER DRAINAGE BASIN

General Description

The Red River basin in Texas is bounded on the north by the Canadian River basin and on the south by the Brazos, Trinity, and Sulphur River basins. (See Figure 1).

The headwater stream in the Red River basin, Tierra Blanca Creek, rises in the High Plains of eastern New Mexico about 40 miles west of the Texas-New Mexico boundary at an elevation of about 4,800 feet above mean sea level. Tierra Blanca Creek flows eastward across the Texas High Plains and becomes the Prairie Dog Town Fork Red River in eastern Randall County. The Prairie Dog Town Fork Red River flows eastward to the southeast corner of the Texas Panhandle where it becomes the Red River. The Red River flows eastward as the Texas-Oklahoma boundary, then becomes the Texas-Arkansas boundary for about 30 miles before leaving the State. At the northeast corner of Texas, the streambed elevation is about 250 feet.

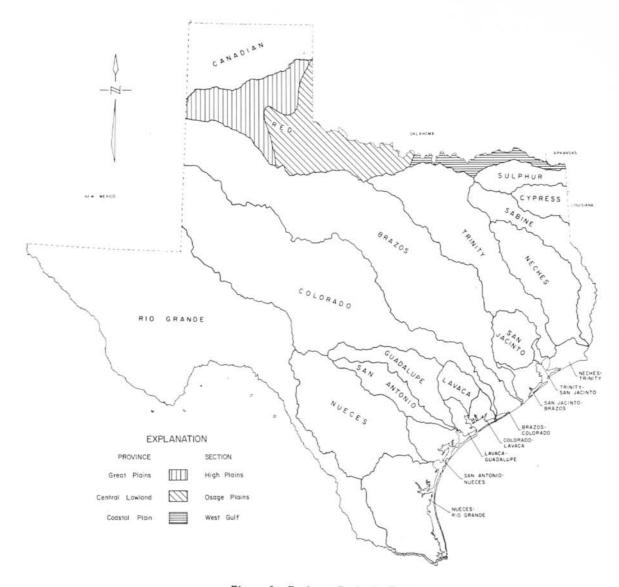


Figure 1.—Drainage Basins in Texas

The Red River has many tributaries in Oklahoma and Texas. The Washita River, Sweetwater Creek, and the North and Salt Forks Red River rise in the Texas Panhandle and flow into Oklahoma before joining the Red River from the north. The major all-Texas tributaries are the Pease, Wichita, and Little Wichita Rivers. Downstream from the Little Wichita River, the Texas part of the basin is narrow and is drained by numerous small streams. The major all-Oklahoma tributaries are Muddy Boggy Creek and the Kiamichi River.

The total area drained by the Red River usptream from the northeast corner of Texas is approximately 48,000 square miles, of which about 5,900 square miles is considered as noncontributing to streamflow. The total area in Texas draining to the Red River is approximately 24,500 square miles of which about 5,300 square miles is considered noncontributing.

The Red River basin in Texas is in three physiographic sections—the High Plains section of the Great Plains province, the Osage Plains section of the Central Lowlands province, and the West Gulf Coastal Plain section of the Coastal Plain province. The physiographic sections are shown on Figure 1.

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The High Plains section within the Red River basin is characterized by a nearly flat surface sloping gently southeastward about 10 feet per mile. Among the few and generally insignificant features of relief are saucerlike depressions, ranging in diameter from several tens of feet to about 1 mile, and ranging in depth from a few inches to about 60 feet. The eastern margin of the High Plains is marked by a prominent escarpment or "break of the plains."

The Osage Plains section within the Red River basin adjoins the High Plains section and has as its eastern boundary the western margin of the gulfward-dipping Cretaceous rocks of the West Gulf Coastal Plain, which extends diagonally from northeast to southwest across Montague County. The Osage Plains section generally is a gentle eastward-sloping plain dissected by prominent systems of drainage. The valleys are wide and bounded by abrupt escarpments, and the streams flow in broad, shallow channels. Much of the surface area has a definite reddish color.

The West Gulf Coastal Plain section extends from the edge of the Osage Plains section eastward throughout the remainder of the report area. Low relief and a gentle gulfward slope of the land surface characterizes this section. Local topographic features are irregular, rolling, and hilly uplands, and flat flood plains and terraces. The streams have wide, nearly flat flood plains bounded by a series of terraces, which may be more than 100 feet higher than the stream channels.

The climate of the basin ranges from semiarid to humid (Thornthwaite, 1952, p. 32). Thornthwaite's classification, which is based on a moisture index, compares potential evapotranspiration with precipitation. Where precipitation is exactly the same as potential evapotranspiration and water is available just as needed, water is neither deficient nor in excess, and the climate is neither moist or dry. As water deficiency becomes larger with respect to potential evapotranspiration, the climate becomes more arid; conversely, as water surplus becomes larger, the climate becomes more humid.

East of a north-south line near the Cooke-Montague County line, the basin has surplus moisture and is characterized by a moist subhumid to humid climate. West of this line the area is deficient in moisture and has a dry subhumid to semiarid climate.

Precipitation ranges from an annual mean of less than 18 inches in the far western part of the basin to more than 46 inches in the extreme eastern part.

Figure 2 shows the average monthly precipitation at Amarillo, Wichita Falls, and Sherman, Texas, and McAlester, Oklahoma; it also shows the annual precipitation for 1937-65 at Altus, Oklahoma. In general, precipitation is greatest during the spring and summer months and least during the winter. However, precipitation is more evenly distributed throughout the year in the eastern part of the basin than in the western part.

Runoff is that part of precipitation that appears in surface streams. It is the same as streamflow unaffected by artificial diversions, storage, or other works of man in or on stream channels (Langbein and Iseri, 1960, p. 17). However, the terms are not synonymous for regulated flow. The Red River is regulated by Lake Texoma, and some of the tributary streams are regulated by reservoirs,

floodwater-retarding structures, and farm ponds. However, many streams in the Red River basin are not regulated by reservoirs of appreciable size.

The 28-year record of the Salt Fork Red River at Mangum, Oklahoma, is the only long-term record of flow from west of the 100th meridian. Runoff varies widely from year to year, and at the Mangum station has varied from a maximum of 200,400 acre-feet in 1941 to a minimum of 8,930 acre-feet in 1940. Annual average runoff is 0.9 inch (50 acre-feet per square mile) at this station. Runoff at Mangum is indicative of runoff from the area west of the 100th meridian.

Runoff in the Red River basin in Texas increases more or less uniformly from west to east, and averages more than 15 inches per year (800 acre-feet per square mile) at the northeast corner of the State. For the period 1944-65, the average runoff was 15.8 inches per year at the U.S. Geological Survey stream-gaging station Boggy Creek near Daingerfield in nearby Cypress Creek basin. Average annual runoff in inches per year, as computed from streamflow records for the period 1938-66, is given for seven stations on Figure 2. Also shown on Figure 2 is annual runoff expressed as mean discharge in cubic feet per second and inches per year for the gaging stations Salt Fork Red River at Mangum, Oklahoma; Red River near Gainesville, Texas; and Red River at Index, Arkansas.

Population and Municipalities

The Red River basin in Texas constitutes about 9 percent of the area of the State and has about 4 percent of the population. Much of the land is sparsely populated. Population changes within the basin reflect the national trend of rural area decline and urban area increase. As in other areas of the country, these changes are due to the reduction of farm employment opportunities resulting from the development of mechanized, large-scale agricultural methods, and the consequent exodus of surplus farm labor to cities, as well as other migration and social factors. The larger cities have continued to grow while the population of small towns has remained fairly constant. The cities and towns having populations over 2,500 are listed in the following table.

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CITY	*POPULA- TION	CITY	*POPULA- TION
Amarillo1/	164,770	Bonham	7,600
Wichita Falls	113,800	Tulia	6,690
Sherman	27,100	Childress	6,420
Paris	24,000	Iowa Park	5,410
Denison	23,400	Shamrock	3,420
Vernon	13,980	Nocona	3,360
Hereford	12,570	Henrietta	3,200
Burkburnett	8,490	Whitesboro	2,980

^{* 1967} population estimates (Dallas Morning News, 1967).

^{1/}Amarillo is partly in the Canadian River basin.

Agricultural and Industrial Development

Agriculture has contributed substantially to the economic growth of the Red River basin. Farming, livestock raising, and dairying are successful because of the fertile soils and generally favorable climate. The availability of ground water for irrigation and the advent of mechanized farm equipment have been largely responsible for the success of farming in the drier western part of the basin. In the eastern part, where rainfall is greater, supplemental irrigation insures good crop yields. Cotton, grain sorghums, and wheat are the principal crops in the western part, and cotton, corn, and vegetables predominate in the eastern part.

The processing of local farm products is one of the major industries in the basin; processing plants are located close to areas of agricultural production. Oil and gas production constitutes another substantial income-producing segment of the economy. Much of the industrial development of the basin is related to the production of oil and gas. The development of irrigation in places has been greatly facilitated by the abundant supply of natural gas for power. Industries in the area that depend on the production of oil and gas include synthetic rubber, carbon black, oil refining, petrochemical, and pipeline equipment. Lumber mills, plants related to timber production, power plants, machinery, and furniture manufacturers are also located in the basin.

Development of Surface-Water Resources

The only reservoir on the Red River is Lake Texoma, which was built for flood control and hydroelectric power generation. Because of the poor quality of the water of the main stem, most of the water development projects in the basin are on tributary streams. As of December 31, 1967, nineteen reservoirs in the Texas part of the basin had capacities of 5,000 acre-feet or more. The capacity, ownership, and use of these reservoirs are listed in Table 1; the locations are shown on Figure 11.

CHEMICAL QUALITY OF THE WATER

Chemical-Quality Records

Although the U.S. Geological Survey has collected chemical-quality records in the Red River basin, Texas, since 1942, very few long-term daily records are available. In 1942, a daily sampling station was established on the Pease River near Crowell, but it was discontinued in 1943. Daily chemical-quality records of more than 10 years are available for the stations at Red River near Gainesville and Red River at Denison Dam. Since 1942, the U.S. Geological Survey has collected daily

chemical-quality data for varying periods at 12 stations either on the main stem or on Texas tributaries. In addition, miscellaneous chemical-quality data are available for numerous sites.

The periods of record at all data-collection sites in Texas are given in Table 4 and the locations are shown on Figure 11. The chemical-quality data for the daily stations are summarized in Table 5, and the complete records are published in an annual series of U.S. Geological Survey Water-Supply Papers and in reports of the Texas Water Development Board and predecessor agencies. Results of all the miscellaneous analyses are given in Table 6. Chemical analyses from selected stations in Oklahoma are given in Table 7. Complete records of all chemical-quality data available for surface water in Oklahoma are published in the annual series of U.S. Geological Survey Water-Supply Papers and in reports of the Oklahoma Water Resources Board. See list of references.

Chemical-quality records, including continuous specific conductance data, were collected by the U.S. Public Health Service (1964) at 27 sites in the Red River basin in Texas and Oklahoma during 1961-62. Public Health Service sampling sites in the Red River basin in Texas are identified in Table 4.

The Texas State Department of Health has made available to the Geological Survey the data collected in its former statewide stream-sampling program. The former data-collection sites are listed in the following table. Some of them are at U.S. Geological Survey stream-gaging stations. The numbers refer to sites shown on Figure 11.

SITE NO.	FORMER TEXAS STATE DEPART-
	MENT OF HEALTH DATA-
	COLLECTION SITES

- 4 Palo Duro Creek at Park Road 5 near Canyon, Texas.
- 13 Prairie Dog Town Fork Red River at State Highway 70 near Brice, Texas.
- 22 Prairie Dog Town Fork Red River at U.S. Highway 83 near Childress, Texas.
- 24 Red River at State Highway 283 near Quanah, Texas.
- Red River at U.S. Highway 183 near Oklaunion, Texas.
- 68 Red River at U.S. Highway 281 near Burkburnett, Texas.
- Red River at State Highway 79 near Byers, Texas.

SITE NO. FORMER TEXAS STATE DEPART-MENT OF HEALTH DATA-COLLECTION SITES

- 97 Red River at U.S. Highway 81 near Terral, Oklahoma.
- 99 Red River near Gainesville, Texas.
- Red River near Denison, Texas.
- Red River at State Highway 78 near Bonham, Texas.
- 104 Red River at U.S. Highway 271 near Arthur City, Texas.
- Red River at State Highway 37 at Albion, Texas.
- Red River at U.S. Highway 59 near Texarkana, Texas.

Streamflow Records

Streamflow records in the Red River basin date from the 1890's, when the U.S. Weather Bureau began collecting gage-height records on the Red River at Arthur City in 1891 and on the Red River near Colbert, Oklahoma (near Denison, Texas) in 1892. The first Geological Survey gaging station was established on the Wichita River at Wichita Falls in 1900. Discharge records are available for more than 50 stations on the Red River and its tributaries in Texas; 11 stations have 15 years or more of record and several others have more than 5 years of record.

In 1966 the Geological Survey operated 26 streamflow stations, five reservoir content stations, and five partial-record stations in the Red River basin, Texas. During this reconnaissance, discharge measurements were made at other sites where water samples were collected for chemical analyses.

Records of discharge, stage of streams, and contents and stages of reservoirs from 1900 to 1960 have been published in the annual series of the U.S. Geological Survey Water-Supply Papers. Beginning with the 1961 water year, streamflow records have been released by the Geological Survey in annual reports for each state (U.S. Geological Survey, 1961, 1962, 1963, 1964b, 1965, 1966). Summaries of discharge records giving monthly and annual totals have been published (U.S. Geological Survey 1955, 1964a; Texas Board of Water Engineers, 1958).

Environmental Factors and Their Effects on the Chemical Quality of the Water

Water from natural sources contains mineral constituents dissolved from the rocks and soils of the

earth's crust. The kind and quantities of dissolved minerals in surface water depend upon a number of environmental factors, some of the most important of which are geology, streamflow characteristics, and the activities of man.

Geology

The amounts and kinds of minerals dissolved in water that drains from areas where municipal and industrial influences are small depend principally on the chemical composition and physical structure of the rocks and soils traversed by the water. The length of time the water is in contact with the soil and rocks is also important. The amount of minerals in the soils and rocks available for solution is decreased by leaching; therefore, in areas of high rainfall, rocks that originally contained large quantities of readily soluble minerals have been leached by circulating water until the mantle rock and residual soil contain relatively small amounts of readily soluble materials. These rocks usually yield water of low mineralization. However, in arid or semiarid regions most soils, and the rocks from which they originated, are incompletely leached and still contain large amounts of readily soluble material.

In the semiarid western part of the Red River basin, some rocks and soils contain large quantities of halite, gypsum, limestone, and dolomite. Water of streams draining these areas usually is highly mineralized. In the eastern part of the basin, where precipitation is more abundant, the well-leached rocks usually yield waters of low mineralization.

The geology of the Red River basin, Texas, has been described by Baker and others (1963, p. 18-26). Rocks exposed in the Texas part of the basin consist of a thick series of sedimentary strata that range in age from Pennsylvanian to Quaternary. The outcrop areas of the geologic units are shown on Figure 3.

Chemical analyses of selected low-flow samples are represented diagrammatically (Stiff, 1951) on Figure 3 to relate chemical composition of surface waters to geology. The shape of the diagram indicates the relative concentrations of the principal chemical constituents of the water (in milliequivalents per liter) and the size of the diagram indicates roughly the relative degree of mineralization.

The headwater stream of the Red River rises in the Ogallala Formation of Tertiary age. The Ogallala consists of clay, silt, sand, gravel, and caliche. Some of the sand, gravel, and silt are unconsolidated; but some cementation occurs, chiefly by calcium carbonate. The principal chemical constituents in water from the Ogallala are sodium, calcium, magnesium, and bicarbonate. Base flow is generally nonexistent in streams that drain the Ogallala outcrop, and runoff occurs only after heavy rains.

Table 1.—Reservoirs in the Red River Basin in Texas Having Capacities of 5,000 Acre-Feet or More. 1

(The purpose for which the impounded waters are used is indicated by the following symbols: M, municipal; I, industrial; Ir, irrigation; P, hydroelectric power; F, flood control; R, recreation.)

	DATE	070544	*CAPACITY				
RESERVOIR	COMPLETED	STREAM	(AC-FT)	OWNER	COUNTY	USE	
Buffalo Lake	1938	Tierra Blanca Creek	18,150	Fish and Wildlife Service, U.S. Department of Interior	Randall	R	
Bivins Lake	1927	Palo Duro Creek	5,120	City of Amarillo	Randall	M	
Baylor Creek	1950	Baylor Creek	9,220	City of Childress	Childress	M	
Greenbelt	1966	Salt Fork Red River	59,800	Greenbelt Municipal and Industrial Water Authority	Donley	М, І	
Lake Kemp	1923	Wichita River	461,800	City of Wichita Falls and Wichita County Water Improvement District No. 2	Baylor	I, Ir	
Diversion Lake	1924	do	40,000	do	Baylor, Archer	I, Ir	
Santa Rose Lake	1929	Beaver Creek	11,570	W. T. Waggoner Estate	Wilbarger	i, ir	
North Fork Buffalo Creek	1964	North Fork Buffalo Creek	15,400	Wichita County Water Control and Improvement District No. 3	Wichita	м	
Lake Wichita	1901	Holliday Creek	14,000	City of Wichita Falls	Wichita, Archer	M	
Lake Kickapoo	1945	North Fork Little Wichita River	106,000	do	Archer	М	
Lake Arrowhead	1966	Little Wichita River	228,000	do	Archer, Clay	M, I	
Farmers Creek	1960	Farmers Creek	25,400	North Montague County Water Supply District	Montague	М, І	
Hubert H. Moss Lake	1966	Fish Creek	23,200	City of Gainesville	Cooke	М, І	
Lake Texoma	1943	Red River	5,393,000	U.S. Army Corps of Engineers	Cooke, Grayson	P, F	
Lake Randall	1909	Shawnee Creek	5,400	City of Denison	Grayson	M	
Brushy Creek	1961	Brushy Creek	16,800	Texas Power and Light Co,	Fannin, Grayson	1	
Coffee Mill Creek Lake	1938	Coffee Mill Creek	8,000	U.S. Forest Service	Fannin	R	
Pat Mayse	1967	Sanders Creek	124,500	U.S. Army Corps of Engineers	Lamar	M, I, F	
Lake Crook	1923	Pine Creek	9,960	City of Paris	Lamar	М	

^{1/} Existing or under construction as of December 31, 1967.

* Total capacity is that capacity below the lowest uncontrolled outlet or spillway and is based on the most recent reservoir survey. available.

Downstream from the Ogallala outcrop, the drainage area of the Prairie Dog Town Fork Red River is underlain by rocks of Triassic and Permian age. The Dockum Group of Triassic age consists of shale and sandy shale, crossbedded sandstone, and conglomerate. The chemical quality of the water from the Dockum Group varies with local conditions, but it is generally unsuitable for irrigation or public supply.

Rocks of Permian age crop out over much of the basin east of the High Plains Escarpment and west of the eastern boundary of Montague County. The Permian rocks consist predominantly of shale, anhydrite, gypsum, limestone, dolomite, and sandstone. The chemical composition of water contributed to streams by these rocks varies. During periods of sustained low flow, water of the Prairie Dog Town Fork Red River, and the North and Salt Forks Red River is of a highly mineralized calcium sulfate type. Water of the Pease River and North and South Forks Wichita River is of a highly mineralized sodium chloride type. Significant natural brine emission areas as identified by the U.S. Public Health Service (1964) are shown on Figure 4. Numerous small alluvial deposits of Quaternary age are present in this area, and ground-water flow from them probably causes some of the variations in chemical composition and dissolved-solids concentration of surface waters.

Downstream from Lake Kemp, the Wichita River drains rocks of the Wichita Group of Permian age. The Wichita Group consists of shale, sandstone, and limestone. Ground-water effluent in this reach is generally of a mixed chemical type having sodium, sulfate, and chloride as the predominant ions. Dissolved-solids concentrations are much less than those of waters above Lake Kemp.

The drainage area of the Little Wichita River is underlain by rocks of the Cisco and Wichita Groups. The Cisco Group of Pennsylvanian age is composed of shale, limestone, sandstone, and conglomerate. Ground-water effluent in Little Wichita River watershed is generally of a mixed chemical type, having sodium, bicarbonate, and chloride as the predominant ions; the water is relatively low in dissolved solids. However, oil-field brine pollution in some reaches in the Wichita and Little Wichita Rivers has in the past altered the composition of water to a sodium chloride type.

Downstream from the Montague-Cooke County line, the streams drain rocks of Cretaceous age. The principal outcrops are the Eagle Ford Shale consisting of shale, limestone, and sand; rocks of Austin age consisting of chalk, marl, and sand; rocks of Taylor age consisting of marl, chalk, and sandy marl; and the Washita and Fredericksburg Groups undifferentiated consisting of limestone, marl, and clay. Waters draining these rocks, although varying slightly in composition from one formation to another, are low in dissolved-solids content

and usually contain calcium and bicarbonate as the predominant ions. Quaternary alluvium is exposed from Lake Texoma eastward to the northeast corner of Texas. However, these alluvial deposits are present along the river in the form of terraces which hold water in bank storage. The alluvium is recharged during high flow, but it releases the water to the river when the high flow subsides.

Streamflow

For many streams not regulated by upstream reservoirs, the concentrations of dissolved minerals vary inversely with the water discharge. The minimum concentrations usually occur during periods of high flow because most of the water is surface runoff that has been in contact with rocks and soils for a relatively short time. The maximum concentrations usually occur during periods of low flow when the water is predominantly ground water that has been in contact with the rocks and soils for a sufficient time to dissolve part of their soluble minerals.

In the western part of the Red River basin, dissolved-solids content and water discharge are not related in a predictable manner. Many of the streams are dry or almost dry much of the time, and salt deposits accumulate on the beds and banks of the streams. Subsequent runoff dissolves these deposits causing erratic variation in the salt content of the runoff.

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In the eastern part of the basin, the dissolvedsolids content of ground-water effluent is generally low, and therefore the dissolved-solids content of streams varies only slightly with discharge. Consequently, the dissolved solids-water discharge relationship is poorly defined for streams in the Red River basin in Texas.

Activities of Man

The activities of man often degrade the chemical quality of surface water. Depletion of flow by diversion and by consumptive use, increased evaporation from impoundment, and return flow from irrigation increase the dissolved-solids concentration of water in streams. Also, the discharge of municipal and industrial wastes into a stream degrades the chemical quality of water.

Eighteen reservoirs presently impound water of Texas tributaries to the Red River (Table 1). All of these except Lake Kemp, Diversion Lake, and Lake Wichita contain waters of good to excellent quality. Because these waters are stored and prevented from reaching the river, the quality of water of the main stem has been degraded to some extent. Much of the lake water eventually returns to the river system, but its quality has been affected by evaporation and other hydrologic changes due to impoundment. Most often the water

diverted from the lakes returns to the river system as waste water from municipal, industrial, and agricultural uses. As the needs for more water and number of reservoirs increase, the quality-of-water problems of the main stem may increase.

At present, degradation of water quality by return flow from irrigation in the Red River basin, Texas is considered minor and localized. Although 1.4 million acres in the basin was irrigated with over 2 million acre-feet of water in 1964, more than 95 percent of both the land irrigated and the water used was in the Texas Panhandle (Gillett and Janca, 1965). Nearly all of the Panhandle supply is from ground water; the Ogallala Formation supplied about 1.5 million acre-feet of water to irrigate approximately a million acres in the High Plains. Irrigation return flow contributes very little to streamflow in the Panhandle, and any flow that does reach a stream probably will enhance rather than degrade the quality of the natural saline waters in most areas. The use of surface water for irrigation is limited primarily to the Wichita Falls area and the area along the Red River north of Texarkana. Minor, localized degradation of small tributaries to the Red River may be occuring in these areas.

Oil is produced in many areas in the Red River basin (Figure 4). Brine is produced in nearly all oil fields and, if improperly handled, eventually reaches surface streams. According to an inventory by the Texas Railroad Commission in 1961, more than 95 percent of the salt water produced in oil fields of the Red River basin, Texas, was injected underground (Texas Water Commission and Texas Water Pollution Control Board, 1963). The remainder of the salt water was disposed of in open surface pits, most of which were unlined. From these surface pits, much of the brine has seeped into the ground and eventually reaches the streams, or it is washed by surface runoff directly into the streams. Also, brine from abandoned wells and unplugged or improperly plugged test holes may reach streams.

The composition of oil-field brine varies, but the principal chemical constituents, in order of magnitude of their concentrations, are chloride, sodium, calcium, and sulfate. Generally, an erratic variation of the sodium chloride content of water in streams draining areas where oil fields are located is evidence that oil-field brine pollution is occurring. Because of widespread contamination of streams in the Red River basin by naturally occurring sodium chloride brines, distinction between natural contamination and manmade pollution is difficult. However, the saline waters of several streams in the central part of the basin have contained salts from both natural sources and oil fields. Chemical-quality records indicate that several streams, not affected by naturally occurring brines, have periodically shown effects of oil-field drainage. Generally, these streams are in the Beaver Creek, Buffalo Creek, and Little Wichita River sub-basins.

Relation of Quality of Water to Use

Quality-of-water studies usually are concerned with determining the suitability of water—judged by the chemical, physical, and biological characteristics—for a proposed use. In the Red River basin, surface water is used for municipal and industrial supplies and for irrigation. This report considers only the chemical character of the water and its relation to these principal uses.

Most of the mineral matter dissolved in water is dissociated into charged particles, or ions. Principal cations (positive charged) in natural water are calcium (Ca), magnesium (Mg), sodium (Na), potassium (K), and iron (Fe). The principal anions (negative charged) are carbonate (CO₃), bicarbonate (HCO₃), sulfate (SO₄), chloride (Cl), fluoride (F), and nitrate (NO₃). Other constituents and properties are often determined to help define the chemical and physical quality of water. Table 2 lists the constituents and properties commonly determined by the U.S. Geological Survey, and includes a resume of their source and significance.

Domestic Use

Because of differences in individuals, varying amounts of water consumed, and other factors, it is difficult to define the safe limits for the mineral constituents usually found in water. The limits usually accepted in the United States for drinking water are the drinking-water standards established by the United States Public Health Service. Originally established in 1914 to control the quality of water used on interstate carriers for drinking and culinary purposes, these standards have been revised several times. The latest revision was in 1962 (U.S. Public Health Service, 1962). These standards have been accepted by the American Water Works Association and by many state departments of public health as minimum standards for all public water supplies.

The maximum concentrations permitted by the standards are given for selected constituents in the following table:

CONSTITUENTS	MAXIMUM CONCEN- TRATION (MILLI- GRAMS PER LITER)
Sulfate	250
Chloride	250
Nitrate	45
Fluoride	a/0.9
Dissolved solids	500

a/Recommended limits based on the average of maximum daily air temperatures. Concentration cited is the optimum based on temperature records for lowa Park.

Table 2.—Source and Significance of Dissolved-Mineral Constituents and Properties of Water

CONSTITUENT OR PROPERTY	SOURCE OR CAUSE	SIGNIFICANCE
Silica (SiO ₂)	Dissolved from practically all rocks and soils, commonly less than 30 mg/l. High concentrations, as much as 100 mg/l, generally occur in highly alkaline waters.	Forms hard scale in pipes and boilers. Carried over in steam of high pressure boilers to form deposits on blades of turbines. Inhibits deterioration of zeolite-type water softeners.
Iron (Fe)	Dissolved from practically all rocks and soils. May also be derived from iron pipes, pumps, and other equipment. More than 1 or 2 mg/l of iron in surface waters generally indicates acid wastes from mine drainage or other sources.	On exposure to air, iron in ground water oxidizes to reddish- brown precipitate. More than about 0,3 mg/lstains laundry and utensils reddish-brown. Objectionable for food processing, tex- tile processing, beverages, ice manufacture, brewing, and other processes. U.S. Public Health Service (1962) drinking-water standards state that iron should not exceed 0,3 mg/l. Larger quantities cause unpleasant taste and favor growth of iron bacteria.
Calcium (Ca) and magnesium (Mg)	Dissolved from practically all soils and rocks, but especially from limestone, dolomite, and gypsum. Calcium and magnesium are found in large quantities in some brines. Magnesium is present in large quantities in sea water.	Cause most of the hardness and scale-forming properties of water; soap consuming (see hardness). Waters low in calcium and magnesium desired in electroplating, tanning, dyeing, and in textile manufacturing.
Sodium (Na) and potassium (K)	Dissolved from practically all rocks and soils. Found also in ancient brines, sea water, industrial brines, and sewage.	Large amounts, in combination with chloride, give a salty taste. Moderate quantities have little effect on the usefulness of water for most purposes. Sodium salts may cause foaming in steam boilers and a high sodium content may limit the use of water for irrigation.
Bicarbonate (HCO ₃) and carbonate (CO ₃)	Action of carbon dioxide in water on carbonate rocks such as lime- stone and dolomite.	Bicarbonate and carbonate produce alkalinity. Bicarbonates of calcium and magnesium decompose in steam boilers and hot water facilities to form scale and release corrosive carbon dioxide gas. In combination with calcium and magnesium, cause carbon- ate hardness.
Sulfate (SO ₄)	Dissolved from rocks and soils containing gypsum, iron sulfides, and other sulfur compounds. Commonly present in mine waters and in some industrial wastes.	Sulfate in water containing calcium forms hard scale in steam boilers. In large amounts, sulfate in combination with other ions gives bitter taste to water. Some calcium sulfate is considered beneficial in the brewing process. U.S. Public Health Service (1962) drinking-water standards recommend that the sulfate content should not exceed 250 mg/l.
Chloride (CI)	Dissolved from rocks and soils. Present in sewage and found in large amounts in ancient brines, sea water, and industrial brines.	In large amounts in combination with sodium, gives salty taste to drinking water. In large quantities, increases the corrosiveness of water. U.S. Public Health Service (1962) drinking-water standards recommend that the chloride content should not exceed 250 mg/l.
Fluoride (F)	Dissolved in small to minute quantities from most rocks and soils. Added to many waters by fluoridation of municipal supplies.	Fluoride in drinking water reduces the incidence of tooth decay when the water is consumed during the period of enamel calcification. However, it may cause mottling of the teeth, depending on the concentration of fluoride, the age of the child, amount of drinking water consumed, and susceptibility of the individual. (Maier, 1950)
Nitrate (NO ₃)	Decaying organic matter, sewage, fertilizers, and nitrates in soil.	Concentration much greater than the local average may suggest pollution. U.S. Public Health Service (1962) drinking-water standards suggest a limit of 45 mg/l. Waters of high nitrate content have been reported to be the cause of methemoglo-binemia (an often fatal disease in infants) and therefore should not be used in infant feeding. Nitrate has been shown to be helpful in reducing inter-crystalline cracking of boiler steel. It encourages growth of algae and other organisms which produce undesirable tastes and odors.
Dissolved solids	Chiefly mineral constituents dis- solved from rocks and soils. Includes some water of crystalli- zation.	U.S. Public Health Service (1962) drinking-water standards recommend that waters containing more than 500 mg/l dissolved solids not be used if other less mineralized supplies are available. Waters containing more than 1000 mg/l dissolved solids are unsuitable for many purposes.
Hardness as CaCO ₃	In most waters nearly all the hardness is due to calcium and magnesium. All the metallic cations other than the alkali metals also cause hardness.	Consumes soap before a lather will form. Deposits soap curd on bathtubs. Hard water forms scale in boilers, water heaters, and pipes. Hardness equivalent to the bicarbonate and carbonate is called carbonate hardness. Any hardness in excess of this is called non-carbonate hardness. Waters of hardness as much as 60 ppm are considered soft; 61 to 120 mg/l, moderately hard; 121 to 180 mg/l, hard; more than 180 mg/l, very hard.
Specific conductance (micromhos at 25°C)	Mineral content of the water.	Indicates degree of mineralization. Specific conductance is a measure of the capacity of the water to conduct an electric current. Varies with concentration and degree of ionization of the constituents.
Hydrogen ion concentration (pH)	Acids, acid-generating salts, and free carbon dioxide lower the pH. Carbonates, bicarbonates, hydroxides, and phosphates, silicates, and borates raise the pH.	A pH of 7.0 indicates neutrality of a solution. Values higher than 7.0 denote increasing alkalinity; values lower than 7.0 indicate increasing acidity. pH is a measure of the activity of the hydrogen ions. Corrosiveness of water generally increases with decreasing pH. However, excessively alkaline waters may also attack metals.

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Industrial Use

The quality requirements vary greatly for almost every industrial application, as is indicated by the water-quality tolerances given in Table 3. One requirement of most industries is that concentrations of the various constituents of the water remain relatively constant. When concentrations of undesirable substances in water vary, constant monitoring is required and operating expenses are increased.

Hardness is one of the more important properties of water that affects its utility for industrial purposes. Excessive hardness is objectionable because it contributes to the formation of scale in steam boilers, pipes, water-heaters, radiators, and various other equipment where water is heated, evaporated, or treated with alkaline materials. The accumulation of scale increases costs for fuel, labor, repairs and replacement, and lowers the quality of many wet-processed products. However, some calcium hardness may be desirable because calcium carbonate sometimes forms protective coatings on pipes and other equipment and reduces corrosion.

The corrosive property of water receives considerable attention in industrial water supplies. A high concentration of dissolved solids in a water may be closely associated with the corrosiveness, particularly if chloride is present in appreciable quantities. Water that contains a large concentration of magnesium chloride may be highly corrosive because the hydrolysis of this salt yields hydrochloric acid.

Irrigation

The chemical composition of a water is an important factor in determining its usefulness for irrigation, because the quality of the water should not adversely affect the productivity of the land irrigated. The extent to which chemical quality limits the suitability of a water for irrigation depends on many factors, such as the nature, composition, and drainage of the soil and subsoil; the amounts of water used and the methods of applying it; the kind of crops grown; and the climate of the region. Because these factors are highly variable, every method of classifying waters for irrigation is somewhat arbitrary.

The most important characteristics in determining the quality of irrigation water, according to the U.S. Salinity Laboratory Staff (1954, p. 69) are: (1) total concentration of soluble salts, (2) relative proportion of sodium to other cations, (3) concentration of boron or other elements that may be toxic, and (4) the excess of equivalents of bicarbonate over equivalents of calcium plus magnesium.

High concentrations of dissolved salts in irrigation water may cause a buildup of salts in the soil solution,

and may make the soil saline. The increased salinity of the soil may reduce crop yields by decreasing the ability of the plants to take up water and essential plant nutrients from the soil solution. The tendency of irrigation water to cause a buildup of salts in the soil is called the salinity hazard of the water. The specific conductance of the water is used as an index of the salinity hazard.

High concentrations of sodium (Na) relative to the concentrations of calcium (Ca) and magnesium (Mg) in irrigation water can adversely affect soil structure. Cations in the soil solution become fixed on the surface of the soil particles; calcium and magnesium tend to flocculate the particles, whereas sodium tends to deflocculate them. This adverse effect on soil structure caused by high sodium concentrations in an irrigation water is called the sodium hazard of the water. An index used for predicting the sodium hazard is the sodium-adsorption ratio (SAR), which is defined by the equation:

SAR=
$$\frac{Na^{+}}{\sqrt{\frac{Ca^{++} + Mg^{++}}{2}}}$$

where the concentrations of the ions are expressed in milliequivalents per liter.

The U.S. Salinity Laboratory Staff has prepared a classification for irrigation waters in terms of salinity and sodium hazard. Empirical equations were used in developing a diagram reproduced in modified form as Figure 5, which uses SAR and specific conductance in classifying irrigation waters. This classification, although embodying both research and field observations, should be used only as a general guide because many additional factors also affect the suitability of water for irrigation. With respect to salinity and sodium hazards, waters are divided into four classes; low, medium, high, and very high. The classification range encompasses waters that can be used for irrigation of most crops on most soils as well as waters that are usually unsuitable for irrigation. The salinity and sodium hazards of water at selected sites in the Red River basin are given on Figure 5.

Geographic Variations in Water Quality

Variations in dissolved solids, hardness, and chloride in the Red River basin are shown in Figures 12, 13, and 14. These values are based on the discharge-weighted average concentrations, as calculated from chemical-quality data. The discharge-weighted average represents the chemical character of the water if all the water passing a point in the stream during a period were impounded in a reservoir and mixed, with no adjustments for rainfall, evaporation, or chemical changes that might occur during storage. For many of the streams

Table 3.-Water-Quality Tolerances for Industrial Applications \underline{J}

	GEN- ERAL	A, B	;	1	1	0,0	OO	0 %, o	0	< m	111	111 1
	Na ₂ S04 TO Na ₂ S03 RATIO	::	1 to 1	2 to 1	3 50 1	::	: :	::::	:: :	111 1	:::	111 1
	CaSO4	::	;	:	;	100-200	::	1111	::::	111 1	:::	111-1
	НО	; ;	20	07	30	1.1	: :	1111	11 :	::::	:::	:::::
	HCO3	1.1	20	30	8	::	::	::::	:: :	:::::	111	111 1
	C03	1:	200	100	05	::	: :	1111	11 1	111 1	111	111 1
	Day"	1.1	;	1	3		!	7 ; ; ;	11 1	::::	:::	::: :
	n Cr	: :	;	1	1	::	1.1	::::	:: :	111 1	5::	:::::
	5102	::	05	20	5	::	::	::::	10	111 1	8::	111 1
	A1203 S102	: :	S	3.	.05	::	::	::::	11 1	111 1	6:11	111 1
[pa	Fe+ Mn	0.5	:	:			2. 2.	4444	.2 .0.	1.0	2005	 1.0
Indicat	Ë	0.5	1	:	:	77	2.2	4444	.2		.00	1.0
xcept as	e A	0.5	:	ŧ	:	44	2, 2,	4444	.2	1.0	.005	.25 .25 1.0
[Allowable Limits in Milligrams Per Liter Except as Indicated]	Ca	::	:	:	:	100-200	::	1111	:: :	1111	111	111 1
igrams Pe	TOTAL	::	3,000-	2,500-	1,500-	1,000	::	100	300	300 200	100	111 1
in Mill	Вф	::	8.0±	8.5+	4.6	7.00	::	1811	11 1	111 1	7.8-8.3	::: :
e Limits	ALKA- LINITY (AS CaCO ₃)	::	;	ŧ	:	150	::	9:11	30-50	:::::	50	111 1
[Allowabl	HARD -	: 3	75	05	80	::	25-75	250	1 05 1	180 100 100 50	8 55 50-135	20 50 50 50
	CODOR	; ;	:	:	;	Low	Low	Low Low	11 1	::: :	:::	::: ₹
	DIS- SOLVED GXYGEN (m1/1)	::	2	.2	0	::	::	1111	:: :	111 1	:::	111 1
	COLOR +02 CON- SUMED	::	100	20	10	::	;;	0 ! ! !	11 1	3 1 1 3	111	111 1
	COLOR	10	80	0.5	5	11	::	8:::	2 ; 2	20 115 10	10-100	5-20 70 5
	TUR- BID- IIY	10	20	10	S	10	10	2 50 10	1-5	50 25 15	3 20	ent e
	INDUSTRY	Air Conditioning ³ / Baking	Boiler feed: 0-150 psi	150-250 ps1	250 pst and up	Brewing: 5/ Light Dark	Canning: Legumes General	Carbonated beverages Confectionary Cooling 9	Laundering Plastics, clear, undercolored	Croundwood Kraft pulp Soda and aulfite Light paper, HL-Grade	Rayon (viscose) pulp: Production Manufacture Tanning 11	Textiles: General Dyeing 12 Wool scouring 13 Cotton band- age 13

Mercican Water Works Association, 1950.

2 A-Ne corrosiveness; B-No slime formation; C-Conformance to Pederal drinking water standards necessary; D-NaCl, 275 mg/l.

3 Waters with algae and hydrogen sulfide odors are most unsuitable for all conditioning.

3 Waters with algae and hydrogen sulfide odors are most unsuitable for all conformations are sirable.

5 Water for distilling must meet the same general requirements as for brewing (gin and spirits mashing water of light-beer quality; whiskey mashing water of dark-beer quality).

5 Water for distilling must meet the same general requirements as for brewing (gin and spirits product.)

7 Water can yet for syrup and carbonization. Water consistent in the recessary as a lower favors inversion of sucrose, causing sticky product.

7 Water can yet requires plu of 7.0 or greater, as low value favors inversion of sucrose, causing sticky product.

8 Control of corrosiveness is necessary as la also control of organisms, such as sulfur and iron bacteria, which tend to form slimes.

9 Ca (HCO₃)? particularly troublesome. Mg (HCO₃)2 tends to greenish color. Co₂ assists to prevent cracking. Sulfates and chlorides of Ca, Mg, Na should each be less than 300 mg/l.

(white buits).

19 Uniformity of composition and temperature desirable. Iron objectionable as cellulose adsorbs iron from dilute solutions. Manganese very objectionable, clogs pipelines and is oxidized to permanganate by chlorine, causing reddish color.

12 Excessive iron, manganese, or turbidity creates spots and discoloration in tanning of hides and leather goods.

12 Constant composition; residual alumina 0.5 mg/l.

13 Calcium, magnesium, iron, manganese, suspended matter, and soluable organic matter may be objectionable.

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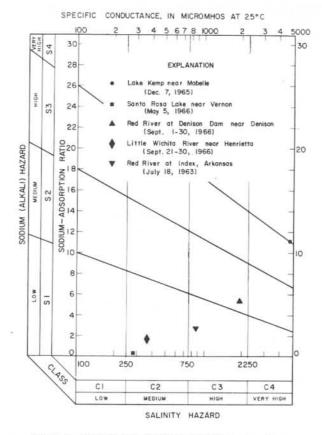


Figure 5.-Diagram for Classification of Irrigation Waters

chemical-quality data are limited, especially data on the chemical quality of flood flows; therefore, the sub-divisions shown on the maps should be considered as generalized. All the streams will at times have concentrations exceeding those shown, but the averages are indicative of the quality of water that would be stored in a hypothetical reservoir.

Dissolved Solids

The concentrations of dissolved solids in streams in the Red River basin range from several thousand to less than 250 mg/l (milligrams per liter) (Figure 3). Water from the outcrop areas of Tertiary age in the extreme western part of the basin usually have dissolved-solids concentrations less than 250 mg/l. Downstream from the Tertiary outcrop, rocks of Triassic and Permian age contribute water containing very high concentrations of dissolved solids; more than 10,000 mg/l is common in some areas. The highly concentrated water from the Prairie Dog Town, Salt, and North Forks Red River, Pease River, and North and South Wichita Rivers cause the mainstem Red River to contain more than 1,000 mg/l of dissolved solids throughout most of its reach in Texas. About midway between Lake Texoma and Index, Arkansas, good quality inflow from the tributaries is of sufficient quantity to cause the Red River to contain less than 1,000 mg/l of dissolved solids.

The discharge-weighted average concentrations of dissolved solids of the Red River near Gainesville for the periods 1944-46, 1953-63, and 1966-67 has ranged from a minimum of 891 mg/l in 1945 to a maximum of 1,950 mg/l in 1958. The discharge-weighted average concentration of dissolved solids of the Red River at Denison Dam for the period 1944-1967 has ranged from 486 mg/l in 1946 to 1,230 mg/l in 1961. The discharge-weighted average concentrations of dissolved solids at Index, Arkansas for 1961, 1962, and 1963 were 728 mg/l, 609 mg/l, and 538 mg/l, respectively. The analyses showing annual maximum and minimum dissolved-solids concentrations and the weighted averages for the stations are given in Table 5. Annual dissolved-solids averages for Red River at Denison Dam are shown on Figure 7.

Time-weighted averages represented by duration curves are usually higher than discharge-weighted averages. The duration curve for dissolved-solids concentrations for the Red River near Gainesville during 1953-62 is shown in Figure 6. Dissolved solids equaled or exceeded 3,560 mg/l 10 percent of the time, 3,040 mg/l 30 percent of the time, 2,620 mg/l 50 percent of the time, and 1,100 mg/l 90 percent of the time.

Downstream from the Wichita River, Texas tributaries drain Cretaceous rocks and contribute water containing less than 250 mg/l of dissolved solids. The discharge-weighted average concentration of dissolved solids of the Little Wichita River near Henrietta for the periods 1953-55 and 1959-66 ranged from 124 to 286 mg/l and averaged 211 mg/l; and the Little Wichita River near Ringgold for the 1959-62 period ranged from 151 to 187 mg/l and averaged 171 mg/l.

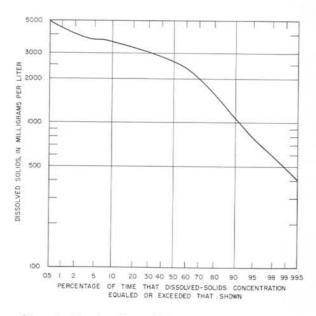


Figure 6.—Duration Curve of Dissolved Solids for Red River Near Gainesville, Texas, 1953-63

Chloride

The concentrations of chloride in surface water of the Red River basin vary from several thousand to less than 100 mg/l. Concentrations are generally less than 250 mg/l in all streams not affected by natural or oil-field brines. Brines in the drainage areas of the Prairie Dog Town, Salt, and North Forks Red River, Pease River, and North and South Wichita Rivers degrade the quality of the Red River throughout its reach in Texas. The annual weighted-average chloride concentration of the Red River near Gainesville (1944-46, 1953-63, 1966-67) has ranged from 283 to 717 mg/l, and at Denison Dam (1944-67), it has ranged from 139 to 431 mg/l. Chloride concentration of the main stem is generally more than 500 mg/l almost as far downstream as Lake Texoma, but it is less than 250 mg/l through the last 100-150 miles of its reach in Texas. Tributaries downstream from the Wichita River generally have chloride concentrations less than 50 mg/l.

Hardness

Surface water in the Red River basin generally ranges from moderately hard (61-120 mg/l) to very hard (more than 180 mg/l). Waters of streams in the western and central parts of the basin that contain high concentrations of dissolved solids are very hard, often having more than 500 mg/l hardness. The streams draining Cretaceous rocks in the eastern part of the basin generally contain waters that are moderately hard, even though they usually have a low dissolved-solids content.

Other Constituents

Other constituents of importance in the evaluation of the chemical quality of a water include silica, sodium, bicarbonate, sulfate, fluoride, and nitrate.

Silica concentrations in the Red River basin range from less than 10 to nearly 70 mg/l. In the western part of the basin, water from streams draining the Tertiary, Triassic, and Permian rocks usually contains more than 20 mg/l of silica. In the central part of the basin, streams draining rocks of Pennsylvanian age usually contain 10 to 15 mg/l of silica; and in the eastern part of the basin, water from rocks of Cretaceous age usually contains less than 10 mg/l of silica. The annual weighted-average silica concentration of the Red River at Denison Dam has usually been about 10 mg/l.

Sodium concentrations range from several thousand to less than 100 mg/l. Concentrations are generally less than 100 mg/l in streams unaffected by natural or oil-field brines. The sodium concentration of the Red River is usually more than 250 mg/l upstream from Lake Texoma and 100 to 250 mg/l from Lake Texoma to Index, Arkansas.

Bicarbonate concentrations are usually less than 250 mg/l in surface waters in the basin; bicarbonate is the principal anion in most waters unaffected by brines. The annual weighted-average bicarbonate concentrations of the Red River near Gainesville and at Denison Dam have usually been less than 150 mg/l.

Sulfate concentrations vary widely in the Red River basin. Streams draining Permian rocks north of the Priarie Dog Town Fork Red River have a sulfate content of several hundred mg/l. Sulfate is the principal anion in these waters. In Prairie Dog Town Fork Red River and in the streams draining Permian rocks south of the Red River, sulfate occurs in high concentrations; but chloride is the principal anion. Downstream from the Permian rocks, the tributaries contain less than 50 mg/l of sulfate. The annual weighted-average sulfate concentration of the Red River near Gainesville (1944-46, 1953-63, 1966-67) has ranged from 169 to 450 mg/l, but it has usually been more than 250 mg/l. The annual weighted-average of sulfate at Denison Dam (1944-67) has ranged from 100 to 297 mg/l, but it usually has been less than 250 mg/l.

Fluoride concentrations are generally less than 1.0 mg/l except in some of the streams that drain the Ogallala Formation. Tule Creek near Silverton at times contains more than 5.0 mg/l of fluoride.

Nitrate concentrations are usually less than 5.0 mg/l, except in some of the heavily irrigated areas of the High Plains where concentrations sometimes exceed 10 mg/l.

Water Quality in Reservoirs

Chemical analyses for most of the principal reservoirs in the Texas part of the Red River basin are given in Table 6. Most of the reservoirs are on tributaries where quality-of-water problems are less severe than on the main stem.

Buffalo Lake

When sampled in 1951, water in Buffalo Lake contained 472 mg/l dissolved solids, 27 mg/l of chloride and was very hard. Principal chemical constituents were sodium and bicarbonate.

Bivins Lake

Chemical analyses are not available for Bivins Lake, but analyses for downstream sites indicate that the stored water contains less than 500 mg/l of dissolved solids, is hard, and has calcium, sodium, bicarbonate, and sulfate as the principal chemical constituents.

Baylor Creek Reservoir

Very limited data indicate that Baylor Creek Reservoir impounds water containing between 500 and 1,000 mg/l of dissolved solids. The water is very hard and of a calcium sulfate type.

Greenbelt Reservoir

Greenbelt Reservoir was not impounding water during this study, but the chemical quality of its water can be inferred from analyses of Salt Fork Red River near Clarendon. Analyses of samples collected during low flow indicate that the dissolved-solids content seldom exceeds 500 mg/l. No data are available on the quality of water at high flow, but it is likely that the dissolved-solids content would be less than at low flow. Therefore, water impounded in the reservoir probably will contain less than 500 mg/l of dissolved solids, be very hard, and of a mixed chemical type.

Lake Kemp and Diversion Lake

Lake Kemp and Diversion Lake were constructed in 1923 and 1924, respectively, and are two of the oldest major reservoirs in the Red River basin. Water is released from Lake Kemp into Diversion Lake and then released or withdrawn for industrial use and irrigation. The reservoirs are downstream from natural salt-contributing areas. Sources of natural pollution of the Wichita River above Lake Kemp were investigated by Joerns (1961) and by the U.S. Public Health Service (1964). Chemical analyses indicate that since construction the impounded water has usually contained 2,000 to 3,000 mg/l dissolved solids. Calcium, sodium, sulfate, and chloride are the principal dissolved constituents. The water is suitable for irrigation for only highly salt-tolerant crops.

Santa Rosa Lake

Water stored in Santa Rosa Lake is low in dissolved solids, moderately hard, and of a calcium bicarbonate type.

North Fork Buffalo Creek Reservoir

Chemical analyses are not available for North Fork Buffalo Creek Reservoir, but analyses for North Fork Buffalo Creek near Iowa Park indicate that at high flow the water is of good quality; dissolved-solids content is less than 200 mg/l. Analyses of water at low flow, however, indicate oil-field brine pollution. The quality of the stored water will depend upon the extent to which brine reaches North Fork Buffalo Creek upstream from the reservoir.

Lake Wichita

Natural runoff into Lake Wichita is probably of good quality. However, the lake receives return flow from areas irrigated with water from Lake Kemp, and also is degraded with water from oil fields. The dissolved-solids content usually exceeds 1,000 mg/l. An analysis in June 1965 showed that the water in Lake Wichita contained 1,450 mg/l of dissolved solids; calcium, sodium, sulfate, and chloride are the principal chemical constituents.

Lake Kickapoo

Water stored in Lake Kickapoo usually contains less than 250 mg/l of dissolved solids and is moderately hard. Principal dissolved constituents are calcium, sodium, and bicarbonate.

Lake Arrowhead

Impoundment of water in Lake Arrowhead began in 1966, and no analyses of the stored water were available during the study period. The quality of the stored water can, however, be inferred from records for the daily sampling station, Little Wichita River near Henrietta. During the period of daily record (1952-55, 1959-66), the annual weighted-average dissolved-solids concentration has ranged from 124 to 286 mg/l, and averaged 218 mg/l. The water was of a sodium chloride type—probably because of oil-field brine reaching the stream.

Farmers Creek Lake

When sampled in 1967, water in Farmers Creek Lake contained 294 mg/l of dissolved solids and was hard. Principal chemical constituents were calcium, sodium, bicarbonate, and chloride.

Hubert H. Moss Lake

Chemical-quality data are not available for Hubert H. Moss Lake, but records from adjoining watersheds in the Trinity River basin indicate that the reservoir, when filled, will contain moderately hard water having a low dissolved-solids content.

Lake Texoma

Denison Dam which forms Lake Texoma was built in 1942 by the U.S. Army Corps of Engineers for flood control and hydroelectric power. Increasing needs for water have caused Lake Texoma to be considered as a source of water for public supply even though it has generally been too highly mineralized for this use. Water from Lake Texoma is pumped to Lake Randall to augment the municipal supply for the city of Denison. The city of Sherman has studied the practicability of damming off the Big Mineral Arm of the lake to obtain a municipal supply (Mendieta and Skinner, 1966).

Since 1965, the dissolved-solids content of water in Lake Texoma has ranged from 969 to 1,230 mg/l. Chloride has ranged from 325 to 442 mg/l, and sulfate from 228 to 296 mg/l. Although the dissolved-solids content of Lake Texoma water varies from year to year, 23 years of records collected since impoundment began in 1944 show a definite trend of increasing mineralization. Annual weighted-average concentrations of dissolved solids for the outflow station at Denison Dam are shown on Figure 7. The net quantity of water available annually at Denison Dam (expressed as annual

outflow plus change in storage, in thousands of acrefeet) is also shown on Figure 7. The net annual water available also represents the inflow to Lake Texoma minus losses due to evaporation, infiltration, and diversion. Along with the trend of increasing mineralization in Lake Texoma, Figure 7 shows a general trend toward less available water.

The mineralization of water of Lake Texoma is increasing because of (1) a continuous salt load reaching the reservoir from upstream natural and man-made brine sources, (2) decreasing inflows to the reservoir because of upstream impoundments, (3) increasing dissolved-solids concentrations of inflows because of upstream impoundments of good-quality water, and (4) increasing dissolved-solids concentrations because of evaporation from Lake Texoma and from impoundments upstream.

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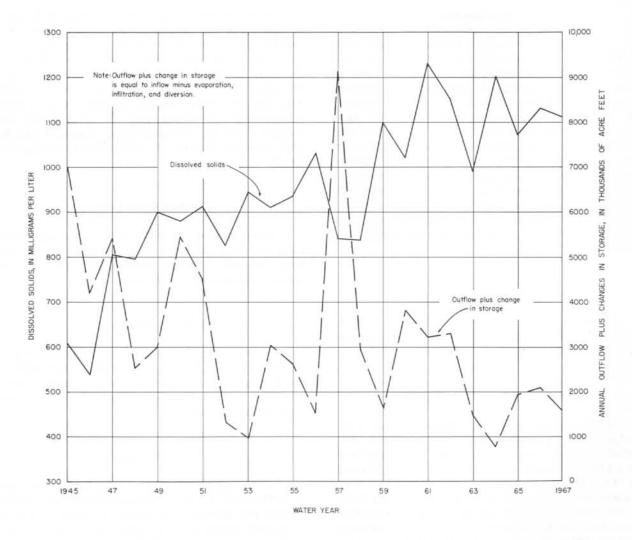


Figure 7.-Graph Showing Dissolved-Solids Content and Quantity of Water in Lake Texoma, 1945-67

Kane (1967, p. 17) shows average annual net lake-surface evaporation for Lake Texoma during the 1940-65 period to be 30 inches per year. At elevation 617 feet (top of power pool), Lake Texoma covers 89,000 acres. Therefore, evaporation losses from Lake Texoma may be more than 200,000 acre-feet per year. West of Lake Texoma average net annual evaporation losses increase rapidly to more than 50 inches in the High Plains. Waters that are released from impoundments above Lake Texoma after having been degraded by evaporation, further degrade the quality of inflows to the reservoir.

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To aid in the evaluation of water quality in Lake Texoma, the Geological Survey made two reconnaissance-type surveys of the reservoir. The surveys were made in March and July 1967 to obtain data for different seasons of the year. Measurements of specific conductance, temperature, and dissolved oxygen were made at sites throughout the reservoir and at various depths at each site. Water samples were collected at selected sites for laboratory analysis. Figure 8 is a map of the reservoir showing the observation sites.

The first survey was made March 21-23. During this period, surface elevation remained almost constant at approximately 603 feet above mean sea level (contents, 1,710,000 acre-feet), and water was being released through the powerhouse. The reservoir was well mixed during this survey. Dissolved-solids content, estimated from specific conductance values and verified by laboratory analyses (dissolved solids equals approximately 0.58 specific conductance), increased only slightly with depth at each site. In the Red River arm, dissolved-solids content was nearly uniform at about 1,200 mg/l from site 1C to site 24C, but increased upstream to about 1,700 mg/l at site 38C. In the Washita River arm, concentrations decreased in an upstream direction to about 1,000 mg/l at site 12C. Temperatures generally were about 1°C lower at the bottom of the lake than at the surface. Dissolved oxygen was nearly uniform throughout the vertical profile at each site-near saturation at the surface and only about 1 mg/l less near the bottom. Vertical profiles for sites 1C and 3C are shown on Figure 9.

During the second survey, made July 25-27, surface elevation was about 614 feet (contents, 2,480,000 acre-feet, an increase of 770,000 acre-feet since the March survey). The dissolved-solids content varied only slightly with depth at each site except at site 38C where the concentration was 1,090 mg/l at the surface and 2,290 mg/l at the bottom. At all the other sites on the Red River arm, dissolved-solids content was near 1,000 mg/l, usually slightly less at the surface and slightly more at the bottom. In the Washita River arm, the dissolved-solids content varied from 935 mg/l at site 7C to about 700 mg/l at site 12C. A thin layer of water about 40 feet below the surface was less concentrated than the water above and below. Temperature and oxygen stratification was evident in all areas of the

reservoir. At site 1C, temperature decreased from 25.8°C at the surface to 21.4°C at the bottom, and dissolved oxygen decreased from 7.2 mg/l at the surface to 0.0 mg/l at the bottom. At most sites, temperature and oxygen decreased slightly with depth through the top 50 feet, then decreased sharply through the next 10 feet, and was nearly uniform through the remaining depth. Vertical profiles for sites 1C and 3C are shown on Figure 9, and longitudinal profiles for the Red River and Washita River arms during the July 25-27 period are shown on Figure 10.

Lake Texoma, like many reservoirs in the southwest, undergoes thermal stratification in the summer and becomes almost completely mixed during the winter. During the summer, the more concentrated inflow from the Red River tends to flow along the bottom of the reservoir and the less concentrated inflow from the Washita River tends to seek an intermediate depth. Dissolved-solids content of the reservoir increases upstream in the Red River arm and decreases upstream in the Washita River arm. During the winter, oxygen is available at all depths throughout the reservoir, but during the summer is generally deficient at all depths greater than 50 feet.

Lake Randall

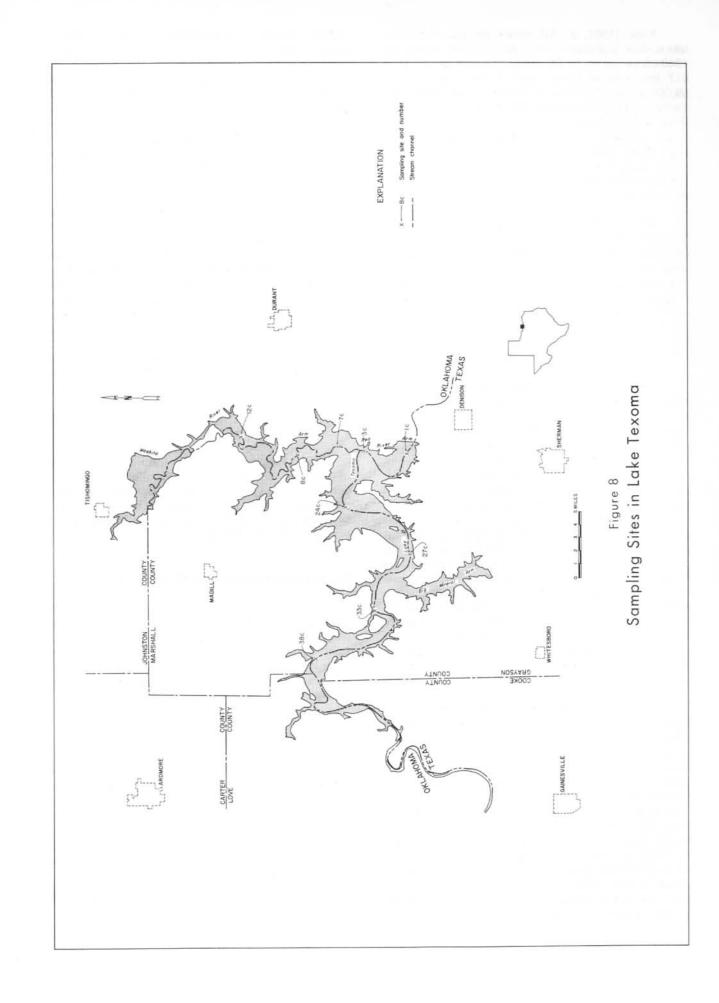
Lake Randall, a small reservoir owned by the city of Denison, is used as a municipal supply. Water is pumped from Lake Texoma to augment the normal yield, and quality of the water in Lake Randall is therefore determined by the proportion of water that is pumped from Lake Texoma.

Brushy Creek Reservoir and Coffee Mill Creek Lake

Although no chemical analyses are available for either of these reservoirs, the water quality can be inferred from records for nearby Bois d'Arc Creek and from records for watersheds in the adjacent Trinity River basin. Water in these areas is usually hard and of a calcium bicarbonate type. The dissolved-solids content averages less than 250 mg/l.

Pat Mayse Reservoir

Pat Mayse Reservoir was not impounding water during this study, but its quality can be inferred from analyses of Sanders Creek near Chicota (site 104). High flows in Sanders Creek contained less than 100 mg/l of dissolved solids, and the reservoir should store water containing less than 200 mg/l of dissolved solids. The water will be moderately hard and of a calcium bicarbonate type.



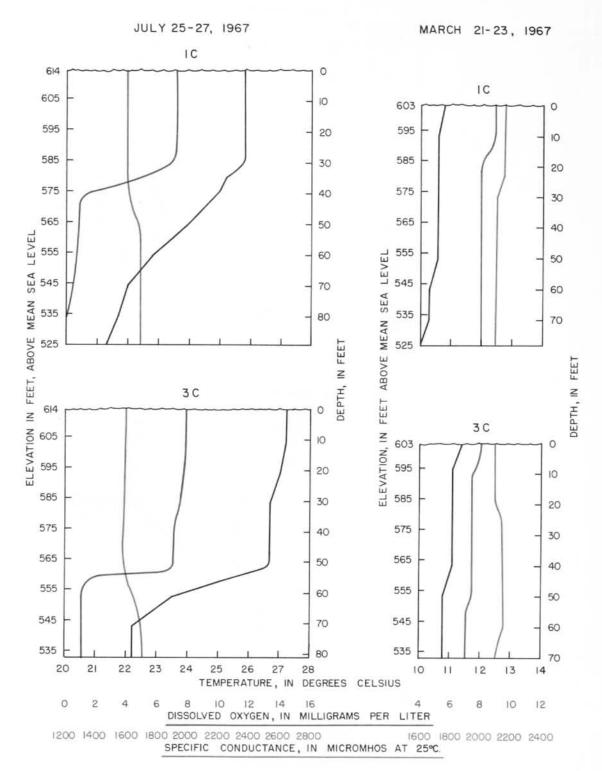
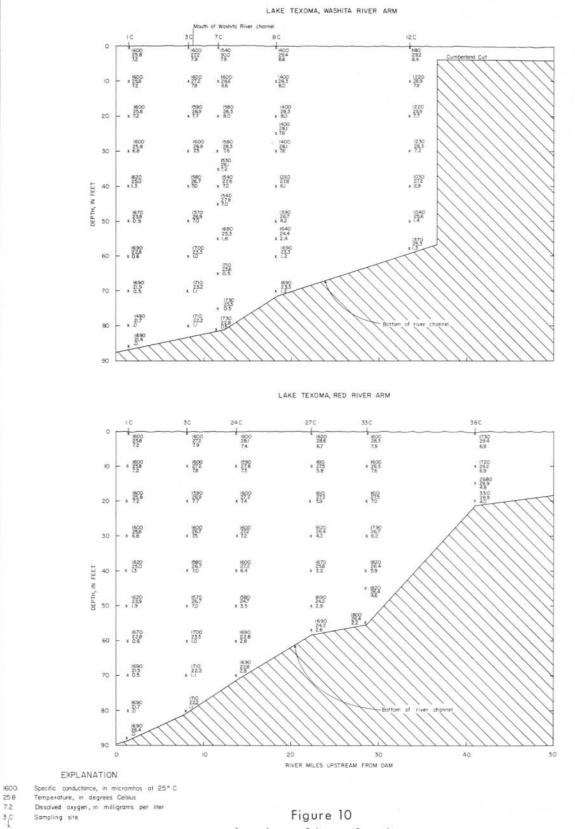


Figure 9

Vertical Profiles of Specific Conductance, Dissolved Oxygen, and Temperature of Lake Texoma



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Longitudinal Profiles of Lake Texoma Showing Water Quality, July 25-27, 1967

Lake Crook

Lake Crook is owned and operated by the city of Paris for municipal water supply. When sampled in 1960, the reservoir water contained 70 mg/l of dissolved solids, was soft, and of a calcium bicarbonate type.

Water Quality at Potential Reservoir Sites

One of the principal objectives of this study was to appraise the quality of water available for storage at potential reservoir sites in the Red River basin. Several potential sites suggested by various agencies are shown on Figure 11. In the following discussion, evaluations of water quality are based on 1967 conditions and the names of potential reservoir sites are those in use as of December 31, 1967.

Mackenzie

A reservoir on Tule Creek at the Mackenzie site would impound water of good quality; the dissolved-solids content would be less than 250 mg/l. The water would be hard and have calcium and bicarbonate as its principal ions.

Buck Creek

Information is not available on the chemical quality of flood flow in Buck Creek. Low-flow samples show high concentrations of calcium and sulfate. A reservoir on Buck Creek would probably impound water containing more than 1,000 mg/l of dissolved solids.

Lelia Lake Creek

Although water samples have been collected periodically for several years, data on the chemical quality of flood flows in Lelia Lake Creek is lacking. Available data indicate that a reservoir on Lelia Lake Creek would impound water of mixed chemical composition containing about 500 mg/l dissolved solids. The water would undoubtedly contain more than 500 mg/l of dissolved solids at times.

Dozier Creek

A reservoir on the Salt Fork Red River at the Dozier site would probably store water containing more than 1,000 mg/l of dissolved solids. A 2-year (1953-54) daily record is available for a station near Wellington (site 47). The weighted-average dissolved-solids concentrations were 1,300 mg/l in 1953 and 1,100 mg/l in 1954.

Lower McClellan Creek

Limited chemical-quality data indicate that a reservoir on McClellan Creek would store water containing about 500 mg/l dissolved solids. The water would be of a mixed chemical type and would be very hard.

Sweetwater Creek

A reservoir on Sweetwater Creek would store water of acceptable quality for most uses. The water would be of a calcium sodium bicarbonate type and contain less than 500 mg/l dissolved solids.

Ringgold

Daily chemical-quality records for the Little Wichita River near Henrietta and Ringgold indicate that a reservoir near the Ringgold site would impound water of good quality. At the Henrietta station the annual weighted-average concentration of dissolved solids was less than 300 mg/l each year during the period of record (1953-66). Oil-field brine has reached the streams in the watershed, and has caused some deterioration of the otherwise excellent-quality water.

Timber Creek and Bois d'Arc Creek

The water available for storage at these sites is of a calcium and sodium bicarbonate type and is moderately hard. The dissolved-solids content should be less than 250 mg/l.

Big Pine

A reservoir on Big Pine Creek would impound water containing less than 150 mg/l dissolved solids. The water would be low in all dissolved constituents and would be soft.

Pecan Bayou

The water in Pecan Bayou is always low in dissolved constituents, therefore, water impounded at the Pecan Bayou site would contain less than 100 mg/l dissolved solids and would be soft.

Barkman Creek

A reservoir on Barkman Creek would impound water containing less than 100 mg/l dissolved solids. The water would be soft and low in all dissolved constituents.

Present and Future Water-Quality Problems

Natural and in the past, oil-field brines are the principal degrading influences on water-quality of the Red River. The highly mineralized waters from salt sources in the western part of the basin cause the water of the Red River to be undesirable for public supply throughout most of its reach in Texas. The salinity problems in the Red River basin have been intensively studied by various Federal agencies. The U.S. Public Health Service (1964) reported that there are 10 primary natural brine emission areas in the Red River basin. Figure 4 shows the locations of the primary sources and several secondary sources, and includes the average daily salt load that the Public Health Service calculated to be contributed by each primary source. A detailed description of each source is given in the report by the Public Health Service (1964).

The U.S. Army Corps of Engineer District at Tulsa, Oklahoma, has prepared a report (1966) on the feasibility of plans to control the major salt sources in the Arkansas and Red River basins and has constructed an experimental control project at Estelline Springs on Prairie Dog Town Fork Red River. Congress has authorized construction of additional salt-control projects on three tributaries of the Wichita River above Lake Kemp, and the Corps of Engineers has proposed five additional projects in the Red River basin, four in Texas and one in Oklahoma.

The plan for control of salt in the Wichita River consists of three low-water dams, pumping facilities, and pipelines for collecting highly mineralized water and moving it to storage basins for evaporation. The Corps estimates that the chloride load reaching Lake Kemp would be reduced by about 80 percent and the sulfate load by about 30 percent. Chloride concentrations would rarely exceed 200 mg/l and sulfate would usually be less than 500 mg/l.

Maximum control of both man-made and natural brine pollution throughout the upper Red River basin, as

proposed by the Corps of Engineers, would greatly improve the quality of the water impounded in Lake Texoma. The Corps estimates that chloride concentrations would be less than 110 mg/l 50 percent of the time, and would seldom exceed 150 mg/l. Dissolved solids would be reduced to an average of about 820 mg/l and sulfate to about 220 mg/l. With its quality thus improved, water of the lower Red River will be suitable for a wider variety of beneficial uses.

Impoundments on tributaries in Texas and Oklahoma and in Lake Texoma are also causing degradation of water quality in the main stem. Of the 18 existing reservoirs on tributary streams in Texas, 16 are impounding water of better quality than is carried by the Red River. Of the 11 potential reservoir sites discussed, 9 would impound water having better quality than water of the main stem.

The city of Sherman is considering damming off Big Mineral Arm of Lake Texoma which is less mineralized than the main body of the reservoir. Most existing and any future reservoirs on Oklahoma tributaries will impound waters of better quality than is carried by the mainstem Red River. Removal of the water of tributary streams leaves water of poorer quality in the Red River. Surface impoundments in the Red River basin will further degrade water of the basin because of evaporation losses. Figure 7 shows the trend of increasing mineralization of Lake Texoma waters, which must result in part from evaporation losses. According to Kane (1967, p. 17) average annual net lake surface evaporation rates vary from 10 inches near the Texas-Arkansas line to more than 50 inches in the High Plains.

The population of the Red River basin in Texas has doubled in the past 25 years. The population growth is expected to be greater in the next 25 years. Along with increasing demands for municipal supplies, more water will be needed for industry and irrigation. If a significant part of these supplies is to come from surface waters of the Red River basin, a maximum effort to improve water quality will be required.

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WATER YEAR	U.S.G.S. WATER-SUPPLY PAPER NO.	T.W.D.B. REPORT NO.	OKLAHOMA WATER RESOURCES BOARD REPORT
1940-45	-	*1938-45	-
1946	1050	*1946	+1946-49
1947	1102	*1947	+1946 49
1948	1133	*1948	+1946-49
1949	1163	*1949	+1946-49
1950	1188	*1950	+1950
1951	1199	*1951	+1951
1952	1252	*1952	+1952
1953	1292	*1953	+1953
1954	1352	*1954	+1954
1955	1402	*1955	+1955
1956	1452	Bull. 5905	+1956
1957	1522	Bull. 5915	+1957
1958	1573	Bull. 6104	+1958
1959	1644	Bull. 6205	+1959
1960	1744	Bull. 6215	+1960
1961	1884	Bull. 6304	+1961
1962	1944	Bull. 6501	+1962
1963	1950	Rept. 7	+1963
1964	2	#	#
1965	=	#	#
1966	-	#	#

^{* &}quot;Chemical Composition of Texas Surface Waters" was designated only by water year from 1938 through 1955.

Published as U.S. Geological Survey open-file report,

^{* &}quot;Chemical Character of Surface Water of Oklahoma" was designated only by water year from 1946 through 1963.

Table 4. -- Index of Surface-Water Records for the Red River Basin, Texas

		Drainning +						
ence no.	Stream and location	area (8q. miles)	area Daily (sq. miles) chemical quality	Discharge	Periodic chemical quality	Periodic discharge measurements	Reservoir	Water
	Therra Bianca Creek above Buffalo Luke near Umbarger	2075		1938-54,				1949-54, 1966
e4	Buffalo Lake near Umbarger	2075			1921		1938-54,	
n	Therra Blunca Greek below Buffalo Lake near Umbarger	ľ		1966				1966
-	Palo Duro Creek near Canyon	982		1942-54				1949-54
673	Prairie Dog Town Fork Red River near Canyon	3369		1937-49				
9	Prairie Bog Town Fork Red River above Stockton Dam near Canyon	1		1965		1961-65		1961-65
1-	Lake Stockton near Canyon	1			1965-66			
x	Prairie Bug Town Fork Red River below Stockton Dam near Canyon	1				1961-65		1961-65
6	Prairie Dog Town Fork Red River above Palo Duro Park near Canyon	1			1961	1961-65		1961-65
1.0	Prairie Dog Town Fork Red River below Palo Duro Park near Canyon	į			1950, 1961,	1961-65		1961-65
11	North Tule Draw at Reservoir near Tulia	189		1938-66			1938-66	1949-66
12	Tule Creek near Silverton	1150		1964-66	1964-66			1964-66
52	Prairie Dog Town Fork Red River near Brice	5972	1950-51	1938-44, 1949-51, 1959-62				1949-51, 1959-62
14	Mulberry Creek near Brice	534	1950-51	1949~51				1949-51
1.5	Prairie bog Town Fork Red River near #478 Lakeview	6792		1963-66				1963-66
16	Little Red River at State Highway 70 near Turkey	1			1959			
1.7	Prairie Bog Town Fork Red River near Estelline	7293		1937-47	1949-50			
18	Estelline Spring near Estelline	ì			1959, 1962	1959, 1962		
119	Baylor Creek Reservoir near Childress	1			1949-50			
20	Baylor Creek near Childress	1			1948			
2.1	Salt Creek 12 miles northwest of Childress	1			1959			
22	Prairie Dog Town Fork Red River near Childress	7725		1964-66	1948-49, 1963			1964-66
55	Buck Creek near Wellington	210			1945, 1947-48 1951-53, 1955-56 1959, 1962	1950-64		1950-64
24	Red River near Quanah *750	8321		1959-66	1959			1959-66
25	North Groesbeck Creek near North Groesbeck	150			1951-53	1951-64		1951-64
26	South Groesbeck Creek near Goodlett	;			1962	1962-64		1962-64
27	South Groesbeck Creek near Acne	146			1951-53 1957-58, 1961	1951-64		1951-64
28	Groesbeck Creek at State Highway 283 near Quanah	303		1962-66	1950-53, 1957-58 1960, 1965-66			1962-66

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Table 4, -- Index of Surface-Water Records for the Red River Basin, Texas--Continued

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Refer-		Drainage			Type and period of record	od of record		
100.	Stream and location	(sq. miles)	Daily chemical quality	Discharge	Periodic chemical quality	Periodic discharge measurements	Reserveir	Water
29	Wanderers Creek at Odell	199			1957-58, 1960	1949-66		1949-66
30	Carroll Creek near Clarendon	ŧ			1951-53	1948-60		1949-60
31	Kelly Creek near Clarendon	1				1961-65		1961-65
32	Greenbelt Reservoir near Clarendon						1966	
23	Salt Fork Red River near Clarendon	457			1956, 1960, 1962	1950-60		1950-64
34	Sait Fork Red Miver above Saddlers Creek north of Lelia Lake	1			1951			
100	Barton Creek northeast of Clarendon	1			1959			
36	Saddlers Creek 8 miles northeast of Clarendon	1			1921			
37	Salt Fork Red River north of Lelia Lake	1			1951, 1959			
38	Lelia Lake Crcek below Bell Creek near Hedley	7.4				1964-66		1964-66
38	Lelia Lake Creek near Hedley	4.6			1957-58, 1964	1951-66		1951-66
40	Salt Fork Red River near Hedley	744	1957-61			1951, 1956-62		1951, 1956-62
4.1	Whitefish Creek near Alanreed	I			1962			
42	Whitefish Creek south of McLean	1			1951, 1962			
43	Whitefish Creek northeast of Hedley	t			1951, 1962			
44	Gyp Creek north of McKnight	1			1921			
4.5	Salt Fork Red River north of Quall	İ			1959, 1963			
46	Dozier Creek near Wellington	1			1950-51, 1953	1950-60		1950-60
47	Salt Fork Red River near Wellington *703	1222	1952-54	1962-66				1962-66
48	North Fork Red River west of Kellerville	1			1959			
40	McCiellan Creek at State Highway 70 near Boydston	1			1950			
20	Lake McClellan near Jericho	1			1951			
51	McClellan Greek at State Highway 273 near.	1			1965	1965		
52	North Fork Red River near Shamrock	1082		1964-66	1958-59, 1964-66	1951-63		1951-66
53	Sweetwater Creek at State Highway 152 west of Moheetie	1			1951			
24	Sweetwater Creek at State Highway 152 southeast of Mobertle	;			1921			
100	Sweetwater Greek near Wheeler	164			1951-53	1951-64		1951-64
26	Sweetwater Creek near Kelion	287		1961-66	1962-66			1961-66
23	Elm Creek near Shamrock	ļ			1946-47, 1950-53 1955, 1958-59 1962	1947-65		1947-65
58	Elm Creek above Wolf Creek near Lutle	1			1962			

Table 4 .-- Index of Surface-Water Records for the Red River Basin, Texas--Continued

	c	(sq. miles)	area Daily (sq. miles) chemical quality	Discharge	Periodic	Periodic discharge	Reservoir	Water
		1 1			_	Bynama tampa		temperature
		1			1962			
		1			1959, 1962			
		293		1945-59	1945-46, 1950-51	1960-66		1960-66
		I			1937, 1952-56 1958-60, 1962	1937, 1943-66		1949-66
		1			1950, 1959			
	tt Lles south of lles southeast east of Paducah	- 1			1959			
	tt Lles south of lles southeast east of Paducah	2747		1959-62	1959			1959-62
	uiles south of niles southeast	3037	1942-43	1924-47				
	D 0	3488		1959-66	1942, 1951			1959-66
		20570		1959-66	1959	1924-25		1959-66
		;			1951-54			
	- 10				1951-54			
		1			1951-52, 1958			
	Sait Creek at mouth 8 miles southeast of Paducah	1			1939, 1951-54, 1956, 1958-59			
	North Wichita River below Salt Creek 12 miles southeast of Paducah	Į.			1952, 1958			
	North Wichita River near Paducah *771	540		1961-66	1958-59, 1965-66	1951-54		1961-66
	North Wichita River near Truscott *757	937		1959-66	1954, 1956, 1959, 1965-66	1952-57		1959-66
	South Wichita River at Guthrie	1			1950, 1958-59 1963			
	South Wichita River tributary 6 miles east of Guthrie	1			1958			
	South Wichita River 6.5 miles east of Guthrie	1			1953-54, 1956 1958-59			
	South Wichita River near Benjamin *756	584		1959-66	1949,1953-54,1956 1959, 1965-66	1952-57		1959-66
80	Wichita River near Seymour	1874		1959-66	1953-54, 1958, 1965-66			1959-66
81 L	Lake Kemp near Mabelle	2086			1939,1942,1946,1952 1954-55, 1964-65	C)	1922-66	
82 W	Wichita River near Mabelle *752	2086		1959-66	1965-66	1952-58		1959-66
83 8	Santa Rosa Lake near Vernon	1			1966			
84 B	Beaver Creek near Electra +751	652		1960-66	1966			1960-66
85	North Buffalo Creek near lows Park	1			1961-63	1961-63		1961-63
86 E	Buffalo Creek near lown Park	1			1964-65	1963-65		1963-65
87 W	Wichita River at Wichita Falls *503	3140		1900-1902, 1910-11, 1938-66	1951			1949-66

See footnote at end of table.

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Table 4. -- Index of Surface-Water Recards for the Red River Basin, Texas -- Continued

Refer-			Destenden			Type and period of record	d of record		
no.	Stream and location	- 5	area	area area Chemical quality	Discharge	Periodic 1	Periodic discharge measurements	Reservoir	Water
88	Lake Wichita at Michita Palls		1			1944, 1946, 1952 1954, 1959, 1965			
68	Withits River at Farm Road 171 near Byers		1			1949, 1951, 1958			
90	Lake Kickapoo near Archer City		275			1946, 1952, 1954		1946-66	
16	Little Wichita River near Archer City		481	1953-55	1932-56				1949-56
92	Lake Creek near Henrietta		1			1959			
63	Little Wichita River near Henrietta	*704	1037	1953-55, 1959-66	1953-66				1953-66
94	Dry Fork Little Wichits River near Henrietta		*			1959			
9.8	Enst Fork Little Wichits River near Henrietta		178		1953-66	1959, 1964-66			1963-66
96	Little Wichits River near Ringgold		1350	1959-62	1959-65				1959-65
2.6	Red River near Terral, Oklahoma	*507	28723		1938-66				1949-66
88	Farmers Creek Reservoir near Nocona					1961			
66	Red River near Gainesville	* 508	30782	1944-46, 1952-63 1966-67	1936-67				1944-46
100	Washita River at Farm Road 2564 near Allison		1			1965	1965		
101	Lake Texoma near Dentson		39719					1942-66	
102	Red River at Denison Dam near Denison	* 522	39720	1944-1967	1923-67				
103	Bois d'Arc Creek near Randolph		7.5		1962-66	1966			1962-66
104	Sanders Creek near Chicota		1		1961-62	1961-62, 1965-66			
105	Red River at Arthur City	*527	44531		1905-11,	1961-63			
106	Lake Crook near Paris		1		1960				
107	Big Pine Creek near Manchester		1		1961-62	1961-62			
108	Pecan Bayou near Clarksville		100		1962-66				1962-66
109	Red River near New Boston		47555		1960-63				
110	Barkman Creek near Leary		*		1961-62	1961-62			
111	Red River at Index, Ark.	+529	48030	1960-63	1936-66				

[.] U. S. Public Health Service has collected chemical quality records at this site. Number in Public Health Service mite number.

Table 5.--Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas

(Analyses listed as maximum and minimum were classified on the basis of the values for dissolved solids only; values of other constituents may not be extremes. Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

							Po-	Bi-	Car-						solved s		Hard as Ca		50-	Specific con-	
Date of collection	Menn discharge (cfs)	Silica (SiQ _g)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	tas- sium (K)	car- bon- ate (HCO ₃)	bon- ate	Sulfate (SO ₄)	Chloride (C1)		Ni- trate (NO ₃)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	
						13.	PRAIR	IE DOG	TOWN	FORK RE	D RIVER N	EAR B	RICE					/			
Maximum, May 16-18,21,1950. Minimum, Oct. 10, 1949 Weighted average.		26		650 397 327	134 42 56	296 20 77	5	114 82 110		1750 1100 930	4740 280 1190		2.5 2.9	10300 2090 3360	14.0 2.84 4.57	284 1600 629	2170 1160 1050			15200 2680	
Maximum, Jan. 15-16, 1951 Minimum, May 17-20 Weighted average	6032	25 20 23		813 202 229	217 36 41	441 30 45	8	153 110 129		2270 583 669	7110 440 663		3.8	14900 1650 2140	20.3 2.24 2.92	406 26900 940	2920 652 752	2800 562 647		21400 2540 3370	8.
							1	4. MU	LBERRY	CREEK	NEAR BRIC	E									
Water year 1950 Maximum, June 24, 1950 Minimum, July 24 Weighted average	204	18		472 128 244	144 17 43	21 4 8	3	79 80 102		1730 334 697	270 49 111		7.5 1.8 1.7	2920 693 1260	3.97 .94 1.73	2.0 382 131	1770 391 786	1700 326 702		3480 918 1650	8.0
Water year 1951 Maximum Mar. 2, 1951 Minimum, June 1-3 Weighted average	597	30 28 31		500 101 200	113 19 40	22 3 8	6	124 113 112		1680 235 566	255 49 115		.0 2.5 2.6	2870 526 1120	3.90 .75 1.52	3.1 893 63.0	1710 330 664	1610 238 572		3390	7.
							40.	SALT	FORK I	RED RIVE	R NEAR HE	DLEY									
Water year 1957 Maximum, Jan. 18, 1957 Minimum, Aug. 29		34 15		371 38	125	26 5 3		276 111		1260 57	328 25	1.0		2520 231	3.43		1440 126	1210 35	3.0	3260 382	7.8
W.ter year 1958 Maximum, Jan. 1, 3, 1958 Minimum, Oct. 15, 17-18, 1957.		28 17		408 64	108	18 10		217 136		1280 166	250 125	. 6	.5	2370 575	3.22		1460 238	1280	2.1	2869 925	
Maximum, Mar. 11-14, 16,20, 22, 25, 23, 1959		24 17		215 72	74 24	27 8		184 100		800 218	325 101	. 9	1.8	1810 563	2.46		841 278	590		2570 917	7.6
Maximum, Nov. 3-9, 17-21, 23-30, 1959		45 13		130 45	49 8 - 1	14 5 2		113 131		516 59	148 27		1.8	1090 270	1.48		526 147		2.8	1510 413	7.9
Water year 1961 Maximum, June 26-29, July 6, 1961 Minimum, Oct. 10-16, 1960		46 19		92 50	39 12	14 4	3 9	78 135		380 90	169 52	. 8	. 5	968 347	1.32 .47		390 174		3.1 3.8	1370 565	

Table 5 .-- Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas -- Continued (Analyses listed as maximum and minimum were classified on the basis of the values for dissolved solids only; values of other constituents may not be extremes. Results in miligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

	Н		7.8		7.7	7.7					8.0	8.0	7.8		1.7	1 7.4
Specific	duct- ance (micro- mhos at 25°C)		3240	3370	1080	5470 960 1550		16000	20400 2420 5540		2360	3730	3550 156 337		3290	
8	ad- ad- sorp- tion ratio		10.0		1.0	1.1					4.0		11.3		21 - 22	
ico,	Non- car- bon-		1700	1800	352	1780 296 518		1	2740 886 1230		184	455			296 1	
Hardness as CaCO,	Cal- cfum, Mag- ne- shum		1810	1940	455	1870 404 634		3350	2860 949 1320		282 79	590	475 40 80		362	22.4 32.4
solids ated)	Tons per day		24.2	116	382	67.6 757 357		1	1840 4580 1620		19.3	570	8.42 998 34.2		474	654
Dissolved solids (calculated)	Tons per acre- foot		3.78	3.89	. 99	5.41 .92 1.50		1	18.9 2.18 5.09		1.66	119	2.57		2.31	1.78
Dis	Milli- grams per liter		2780	2860	730	3980 677 1100		11100	14200 1600 3740		1220 230 328	2340 137 168	1890 95 197		1700	1310
	Bo- ron (B)															
	N1- trate (NO ₃)		3.0		1.5	3.0		5 -			8 42 62	1 8 7.	0.44		3.55 2.55 2.55	3.0
	Fluo- ride t (F) (NGTON	8.0	œ	10.1	8 4 7		1	0.6	R CITY	8.5	100	4 0 0	LETTA	111	0.8
	Chloride (C1)	RIVER NEAR WELLINGTON	185	255	134	1030 71 141	CROWELL	4260	6480 208 1250	NEAR ARCHER	675 65 128	1130 25 53	11110 20 51	NEAR HENRIETTA	988 27 116	728
	Sulfate (SO ₄)	RIVER NE	1690	1650	359	1590 295 518	PEASE RIVER NEAR			RIVER NE	14 7.1 7.1	1 22 4	13 7.8 5.6	RIVER NI		
200		C RED					E RIVI							WICHITA		
	car- bon- ate (HCO ₃)	SALT FORK RED	135	174	125	114	PEAS	99	145 76 112	LITTLE WICHITA	121	165 59 73	86 94	LITTLE WI	81 59	93
Po-			151	31	51	613 53 102	99	0.00	3 4 0	1000	01024	340	2 8 9		400	10 -1 1
	Sodium (Na)	47	22	7	* 3	63		2660	4050 134 793	91.	362 53 84	1 64 65	542 18 36	93	514	415
Mag-	ne- sium (Mg)		101	108	25	141 21 40		199	170 33 64		25 5.9 8.0	1	37 2.4 5.5		31 4.3 7.5	
ē	cdum (Ca)		558	600	141	518 127 188		1010	864 326 424		22 22 27	14	129 12 23		94	8.2
	(Fe)															
	Silica (SiO ₂)		28 40	20	26	29 25		11	122		118	10 12	8.6 6.4 9.7		14 9.4 12	16.
Monn	discharge Sillea (cfs)		3.22	15.0	194	6.29 414 120		1.1	49.0 1060 161		5.87 21.1 6.1	.05 1542 60.0	1.65 3890 64.3		103 137 12.6	185 3326 204
Date	uo		June to September 1952 Maximum, Sept. 21-30,1952 Minimum, June 22-24	Maximum, Dec. 18-30, 1952	1953. Weighted average	ater year 1954 Maximum, Aug. 12-20, 1954 Minimum, Oct. 21-24, 1953 Weighted average		July to September 1942 Maximum, July 16, 1942 Minimum, June 10.	ater year 1943 Maximum, Dec. 24, 1942 Mathmum, Apr. 17, 1943		January to September 1953 Maximum, Aug. 15-17, 1953. Minimum, Aug. 20-28.	19, 1954 22-27, 1953	Maxhaum, Nov. 17-18, 1954 Minimum, Sept. 25-26, 1955. Weighted average		Dec. 1952 - Sept. 1953 Maximum, Mar. 15-16 Minimum, Mar. 14, 17-18	ater year 1954 Maximum, Oct. 6, 1953 Maximum, Oct. 22-24, 26-29 Weighted avorage
			June to S Maximum, Minimum,	Water year 1953 Maximum, Dec.	1953. Weighted	Water year 1954 Maximum, Aug. Minimum, Oct. Weighted averag		July to Se Maximum, Minimum,	Water year 1943 Maximum, Dec. Minimum, Apr. Weighted avera		January t Maximum, Minimum, Weighted	Water year 1954 Maximum, Sept. Minimum, Oct. Weighted avera	Water year 1955 Maximum, Nov. Minimum, Sept. Weighted avera		Maximum, Minimum, Weighted	Maximum, Oct. Maximum, Oct. Weighted avera

Table 5.--Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas--Continued

(Analyses listed as maximum and minimum were classified on the basis of the values for dissolved solids only; values of other constituents may not be extremes. Results in milligrams per liter except as indicated.

- 63	н		7.4	7.8	6.3	6.6	7.2	7.6	6.8	7.8	6.4		6.9	8.9
Specific con-	日 日 4		3250 97 306	2300 116 404	7520 204 498	4590 104 458	3170 129 436	4560 44 290	1990 182 455	1600 134 439	2490 162 231		5200 60 279	7860
8	ad- sorp- tion ratio		1.9	80 (5) 80 (6)	1.0	1. 2	2.2	9.4	9.0	2.2	7.5		12	16
CO,	Non- car- bon-		358	184	1010 7 33	589 1 22	327 0 21	697	169 0 17	144 0 19	280 2 6		648 1 10	1100
as CaCO,	Cal- clum, Mag- ne- slum		418 29 64	289 31 78	1060 56 91	666 29 80	392 34 87	836 8 63	220 45 88	236 40 89	388 44 54		770 19 55	1150
ated)	Tons per day		415 220 51.7	1510 388 46.7	1280 72.8 45.7	3.29 126 33.7	679 188 66.3	.67	23.7 33.8 73.0	191 86.5 48.0	261 920 172		244 39,3 53,0	1590
(calculated)	Tons per acre- foot		2.27	1.62	5.60	3.32	2.33	3.36	1.39	1.16	1.75		3.82	6.04
	Milli- grams per liter		1670 57 166	1190 63 218	4120 110 270	2440 59 243	1710 76 234	2470 30 158	1020 106 242	855 77 232	1290 90 124		2810 38 151	4440
	Bo- ron (B)													
	N1- trate (NO ₂)	inued	999	2.2	1.9 8.1	1 00 65	3.2	1.2	2.2 1.8 2.7	3.2	5.7		2.0	1
	Fluo- ride (F)	-Cont	0. 4.0.	8 54 65	155	1 44 44	411	111	14:1	1 2 1	ಗು ಬ ಬ	COLD	0.3	KT.
	Chloride (C1)	HENRIETTAContinued	980 10 56	642 10 85	2500 26 114	1450 9.2 100	1000 10 89	1440 8.0 53	570 14 89	450 11 87	710 20 37	NEAR RINGGOLD	1640 4.0 52	2680
	Sulfate (SO.)	NEAR	88 4.3	25 4.4 6.6	6.8	6.6	16 4.2 6.1	7.8	27 8.0 7.5	17 6.0 8.9	3.2	A RIVER	34 2.6 5.1	6.9
	bon- ate (CO ₃)	RIVE										WICHITA		
Bi-	car- bon- ate (HCO ₃)	WICHITA RIVER	74 34 69	128 39 69	58 59 70	94 34 71	79 51 79	170 8 63	62 72 92	113 50 85	132 50 58	LITTLE W	150 22 55	90
ė	Fo- tas- sium (K)	LITTLE W	481 8.4 36	345 10 50	18	18.8	505	628	306 20 56	235 11 52	343 16 25	96. LI	783 4.3	1260
	Sodium (Na)	93. LI	÷	6			iñ	9	6	61	es	6	7	1.9
	Mag- ne- slum (Mg)		23.3	23 3.4 6.1	3.8	6.29	29	76	3.1	18 3.1 6.7	30.4.0		63 1.5 4.3	96
10	Cal- cium (Ca)		113 7.8 19	78 6.8 21	16	6.8	109	210	74 13 25	65 11 25	106 12 15		205 5.2 15	302
	(Fe)													
	Silica (SiO ₄)		11 7.2 9.7	0.08 0.40	18.6	9.3	8 1 51	8, 1.3	7.2	11 8.4 9.5	6.6 8.0		7.8	1.0
	Mean drscharge (cfs)		92.0 1430 114	171 2280 79 1	245 62.7	789 51.4	147 918 105	5.9	8.6 118 56.3	82.7 416 46.2	75.0 3788 203		32.1 383 130	133
	Date of collection		Mater year 1955 Maximum, Sept. 24, 1955. Minimum, May 19. Weighted average.	March to September 1959 Maximum, May 12. Minimum, June 23.	Matter year 1960 Maximum, June 2, 1960 Minimum, Mar. 26. Wellpited average.	Maximum, June 1-8, 1961 Minimum, Oct. 15-16, 1960., Weijhted average	ater year 1962 Maximum, Nov. 4, 1961 Minimum, June 30, 1962.	Ater year 1963 Maximum, Feb. 1-4, 8-11, 16-19, 1963. Maximum, Nov. 24-25, 1962. Weighted average.	nater year 1964 Maximum, Dec. 15-17, 1963. Minimum, Sept. 16-17, 1964. Weighted average	Maximum, Apr. 166.13, 1964 Minimum, Aug. 16-17 Weighted average.	aler year 1966 Maximum, July 12, 1966 Minimum, Apr. 26-30 Weighted average.		March to September 1959 Maximum, Mar. 16-18, 1959. Mainimum, Sept. 4 Weighted average	Naxter year 1960
			Mater year 1955 Maximum, Sept. 24 Minimum, May 19 Weighted average.	March to September Maximum, May 12 Minimum, June 23. Weighted average	Water year Maximum, Minimum, Weighted	Water year 1961 Maximum, June Minimum, Oct. Wei, hied avera	Water year 1962 Maximum, Nov. Minimum, June Weighted avera	Water year 1963 Maximum, Feb. 16-19, 1963. Minimum, Nov. Weighted avera	Water year 1964 Maximum, Dec. Minimum, Sept. Weighted avera	Water year 1965 Maximum, Apr. Minimum, Aug. Weighted avera	Water year 1966 Maximum, July Minimum, Apr. 2 Weighted averag		March to Maximum, Minimum, Weighted	Water year 1960

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Table 5.--Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas -- Continued (Analyses listed as maximum and minimum were classified on the basis of the values for dissolved solids only; values of other constituents may not be extremes. Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

So- Specific	dium duct- ance pH Borp-(micro- tion mhos at ratio 25°C)		9.7 2570 6.7			1320	6550	5510 403	2.5 584 8.0	0 97	7 00 00	8670 8.		2
Hardness as CaCO,	Non- car- bon-		247	34) 1190 5 191 5 525			1380	1400 1	1090 72 377	1370	1140 1	002
Han	Cal- clum, Mag- ne- stum		316			1310 285 628	1170	928 120	1510	1500	1190 170 484	1480	1240 158	202
solids ed)	Tons per day		155	42.9 231 74.4		3570 1960 7410	5500	00101	47940 2850	11310 46670 9510	13470 15830 9730	7860	5120 44850	Decer
Dissolved solids (calculated)	Tons per acre- foot		1.82	2.54		6.51 1.03 2.56	5.49	*	8.81	56	5.21	7.47	. 79	
ā~	Milli- grams per liter		1340	1870 45 168		4790 757 1880	4040	3290	6480 342	5210 412 1140	3830 400 1370	5490 446	4260 335 917	
	Ni- Bo- trate ron (NO ₂) (B)	ed	2,614			3.2.2	3.5		0.4	1.1	1.7	18:1	1.4.1	
	Fluo- ride (F)	-Continu	4.0.		E .									
	Chloride (Cl)	NEAR RINGGOLDContinued	770		GAINESVILLE	2070 252 705	1740 62 335	1440	2750 101 698	2000 114 394	1400 118 462	2250 155 533	1700 67 283	
	Sulfate (SO.)		18 4.2		NEAR	940 181 450	751 31 169	521	1190 42 412	1190 70 246	900 70 326	1120 57 341	917 58 209	
	car- Car- bon- ate ate (HCO ₃)	TA RIVER	325	64 30 66	RED RIVER	4:0.0	447	F-# 1	W 10 to	2 2 2	900	# 0 10	m C 15	
BI-	tas- sium ate (K) (HCO ₃	LE WICHITA	4	9	99. R	1154	214	!	105	123 102 123	126 120 130	134 90 136		
	Sodium (Na) s	6. LITTLE	39.5	564 34		1240 163 433	1030 36	866	1730 66 436	1140 74 242	867 283	1370 89 323	1020 49 169	
N.	nag- ne- stum (Mg))6	23.5	0.0110		86 19 43	97 7.3	75	94 8.4 38	86 6.2 23	73 12 32	103 12 37	69 7.5 23	
	Cal- cfum (Ca)		89 6.5 18	112 5.0 17		384 83 181	310 36 97	36	450 39 169	460 52 115	356 48 141	424 45 146	384 51 107	0.000
	(Fe)													
	(SiO ₂)		6.4 7.4 8.8	12										
	Mean discharge (cfs)		26.0 1046 80.6	34.1 1900 164		276 957 1459	504 42300 4193	1206	2740 3090 651	804 41950 3090	1303 14660 2630	530 1220 2177	445 49580 7484	200
	Date of collection		Maximum, Oct. 8-15, 1960 Minimum, Oct. 16-17 Weighted average	2.7.5		Max to September 1944 Maximum, Sept. 21-30. Minimum, June 13-16. Weighted average.	Water year 1945 Maxhaum Jan 11-20, 1945 Weighted average.	October 1945 to April 1946 Maxhmum, Jan. 23-31, 1946. Minimum, Oct. 1-3, 1945 Weighted average	Maximum, Apr. 1-11, 1953 Minimum, July 22-23 Weighted average	Water year 1954 Maximum, Aug. 30-31. Minimum, May 12-13.	Water year 1955 Maximum, Aug. 1-6, 1955 Minimum, Sept. 26-30 Weighted average	Water year 1956 Maximum, June 21-31, 1956 Minimum, July 11 Weighted average	Water year 1957 Maximum, Sept. 11-12, 1957. Minimum, Apr. 26-304	Water year 1958

Table 5,--Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas--Continued (Analyses listed as maximum and minimum were classified on the basis of the values for dissolved solids only; values of other constituents may not be extremes. Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

	ш		7.7	7.9	0.88	7.9	8 8 2 8 2 5 2 5 2	7.6	1	7.8	0.01	111	111	111	111
Ific	tt- pH ro- C)													500 230 310	
02	duct- ance (micro- mhos at 25°C)		7150 0 802 9 2560	7080 3 362 0 2590	8050 2 785 3 2830	6180 8 294 0 2120	8800 7 460 0 2660	5350 8 787 0 2400		2430 1610 2040	. 2280 852 1070	1050 762 874	2290 1070 1340		1710 1340 1520
os -	ad- sorp- tion ratio		15 3.	7 - 7	15	6. 12	7.	11.2.9		111	111	111	111	111	111
CO3	Non- car- bon- ate		1060 100 409	1130 28 397	1340 72 450	969 16 314	1140 52 372	807 77 358		389 222 314	316 9 166	3 114 120	378 136 202	228 179 196	242 203 222
Hardness as CaCO,	Cal- ctum, Mag- ne- stum		1160 185 512	1240 116 515	1450 168 571	1120 130 429	1260 180 498	997 180 473		522 352 446	457 248 281	272 224 245	197 269 323	340 289 310	352 310 334
olids ed)	Tons per day		8260 2500 6790	10270 16350 13070	32580 6770 14960	3260 498 9370	11310 3330 5540	1600 5970 5080		2880 497 946	523 34200 11900	3680 47000 9000	8620 4400 17200	5190 4840 7590	6480 4530 9440
Dissolved solids (calculated)	Tons per acre- foot		6.38	6.47	7.66	5.28 .30 1.82	7.59	4.50 .59		1.94	1.77 .67 .83	.65	1.85 .88 1.09	1.23	1.41
Dise	Milli- grams per liter		4690 472 1640	4760 217 1660	5630 463 1820	3880 221 1340	5580 292 1590	3310 433 1430		1430 902 1180	1300 489 607	592 410 486	1360 644 805	905 762 797	1040 774 901
	Bo- (B)		111	111	111	0.57	.43	111							
	Ni- trate r (NO ₃) (p	18:1	141	3.6	111	111	2.2	N	6.4.4 6.8.5	2.0	1.0	1.5	.8	2.8 1.9
	Fluo- N ride tr (F) (N	ontinue						0.2	DENISON						
	Chloride (C1)	NEAR GAINESVILLE Continued	1950 60 566	1900 45 590	2120 146 644	1580 21 456	2350 19 611	1400 146 545	DENISON DAM NEAR	520 318 424	465 146 195	195 115 139	490 185 250	288 215 239	332 246 290
	Sulfate (SO ₄)	R GAINES	815 73 375	990 32 342	1210 66 390	718 31 283	1150 72 324	610 60 302	DENISON	323 183 255	290 98 129	106 86 100	321 120 164	184 176 175	217 175 193
	bon- ate (CO ₃)								AT						
Bi-	car- bon- ate (HCO ₃)	RED RIVER	116 104 125	138 108 144	132 118 148	184 140 140	146 148 150	232 126 141	RED RIVER	162 158 161	172 149 140	165 134 152	145 163 148	137 134 140	135 131 137
,	Fo- tas- slum (K)	99. RI	1170 93 359	1190 32 364	95 399	923 21 287	1550 23 381	8.4.6 1.4.8	102. RI	315 194 255	296 84 114	115 63 84	298 116 149	170 141 150	209 150 178
	Sodium (Na)							812 86 329							
	Mag- ne- sium (Mg)		100 16	90 6.3	100 12 43	120 3.6 29	69 8.6 36	86 10 30		37 32 32	34 19 21	23 17 19	34 23 24	27 23 24	25 24 26
	Cal- clum (Ca)		300 48 154	348 36 147	416 48 158	252 46 124	392 58 141	258 56 140		148 98 126	127 68 78	7.1 62 67	143 70 90	92 78 85	100 85 91
	Iron (Fe)														
	Silica (SiO ₂)		111	111	111	111	111	4.6 7.9 7.9		111	111	111	1 1 1	01	12 7.8 8.6
	Mean discharge (cfs)		652 1960 1534	799 27900 2916	2143 5412 3044	311 835 2591	751 4220 1289	179 5110 1316		746 204 297	149 25880 7261	2043 36430 6199	2347 2528 7923	2124 2351 3528	2307 2166 3880
	Date of collection		ater year 1959 Maximum, April 20, 1959 Mannaum, Sept. 5	Maximum, July 1-8, 1960 Minimum, Oct. 4, 1959	Maximum, July 16-18, 1961. Minimum, Sept. 16-20.	Maximum, April 18-22, 1962. Minimum, Sept. 2.	arter year 1963 Maximum, June 26-July 1,1963 Minimum, Nov. 26, 1962	Maximum, Mar. 1-31, 1967 Minimum, July 7-8 Weighted average		ay to September 1944 Maximum, Aug. 11-20, 1944. Minimum, June 11-20.	ater year 1945 Maximum, Jan. 1-10, 1945 Minimum, July 1-10	ater year 1946 Maximum, Aug. 11-20, 1946 Minimum, Oct. 11-20, 1945 Weighted average	ater year 1947 Maximum, Oct. 20-31, 1946 Minimum, Oct. 1-10	Maximum, Sept. 1-30, 1948 Minimum, Dec. 1-31, 1947 Weighted average	ater year 1949 Maximum, July 1-31, 1949 Minimum, Nov. 1-30, 1948 Weighted average
		I	Water year 1959 Maximum, April Minimum, Sept. Weighter avera	Water year 1960 Maximum, July Minimum, Oct.	Water year 1961 Maximum, July Minimum, Sept. Weighted avera	Water year 1962 Maximum, April Minimum, Sept. Weighted avera	Water year 1963 Maximum, June 2 Minimum, Nov. Weighted avera	Water year 1967 Maximum, Mar. Minimum, July Weighted avera		May to September 1944 Maximum, Aug. 11-20, Minimum, June 11-20. Weighted average	Water year 1945 Maximum, Jan. Minimum, July Weighted avera	Water year 1946 Maximum, Aug. 11-2 Minimum, Oct. 11-2 Weighted average	Water year 1947 Maximum, Oct. Minimum, Oct. Weighter avera	Water year Maximum, S Minimum, D Weighted a	Water year Maximum, J Minimum, N

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Table 5.--Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas--Continued

(Analyses listed as maximum and minimum were classified on the basis of the values for dissolved solids only; values of other constituents may not be extremes. Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

	Date	Mean			Cert	Mag-		Po-	Bi-	Car-							ssolved a		Hard as C		So-	Specific	-
	of collection	discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	ne- sium (Mg)	Sodium (Na)	tas- sium	car- bon- ate (HCO ₃)	bon- ate	Sulfate (SO ₄)	Chloride (C1)	Fluo- ride (F)	Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	pH
							LO2. REI	RIVE	ER AT I	ENISO	N DAM NE	AR DENIS	NCo	ontinu	ied								
Water year																					-		
	Mar. 1-31, 1950	1772	11		96	27	18		129		215	300	-	3.5		977	1.33	4670	350	245		1600	
	Sept. 1-30 average		12 13		78 88	21	15		126		165	245		2.0	-	790	1.07	22000	281	178		1320	
		7043	1.0		00	24	17	4	130		191	276		3.1		882	1.20	16800	318	212		1460	
Water year Maximum.	June 1-30, 1951	25960	11		99	29	19	10	1.10		0.07			102.0									
	Sept. 1-30	2563	14		77	21	14		149 132		207 157	325 225	77	. 8		1010	1.37	70800	366	244		1670	
	average	6992	11		91	25	17		141		187	290		1.0		725 913	1.24	5020 17200	278	170		1220	
Water year	1952											200		1.2		513	1.24	17200	330	214	77	1500	-
	Aug. 1-31, 1952	3140	8.2		89	27	18	13	145		200	285		2.0		894	1.22	7500	200		9.2		
Minimum,	Oct. 1-31, 1951	1841	11		68	23	14	4	135		160	212		2.0		722	.99	7580 3590	333 264	214 154	3.9	1530 1180	
	average	2301	9.5		83	26	16	1	142		185	250		1.9		827	1.12	5140	314	198	3.9	1380	
Water year		120000	2:25																			1.575.31	
	Aug. 1-31, 1953 Oct. 1-31, 1952	2932 2394	11		92	28	19		140		205	315	***	1.0	44.40	995	1.35	7880	344	230	4.6	1620	8.
	average	1853	9.5		88 92	28 29	18 19		140		203	295		1.2		912	1.24	5890	334	220	4.5	1520	
Water year		1000	5.0		52	20	1.9		142		207	305	77.77	1.9		944	1.28	4720	348	232	4.4	1570	-
	Nov. 1-30, 1953	1186	8.8		100	27	23	9	123		000	222	9.5	7.1.									
Minimum,	July 1-31, 1954	4608	15		84	21	16		128		239 178	370 275	0.5			1040	1.41	3330	360	260	5.3	1750	
Weighted	average	3950	12		89	24	18		128		200	299			. 20	908	1.13 1.23	10330 9680	296 320	191	$\frac{4.2}{4.5}$	1390	
Water year													2005	10000	3 40	200	4.1.60	5080	320	210	4.5	1530	-
	Sept. 1-30, 1955	2688	11		106	21	21	6	122		240	342	. 4	1.2	.14	1000	1.36	7260	351	251	4.0	1700	-
Woightod	Oct. 1-31, 1954	1109	12		86	22	17		122		190	278	. 4	1.5	. 08	880	1.20	2630	305	205	4.9	1720 1480	7 - 1
	average	2762	9.9		96	22	19	3	126		209	306	. 3	1.5	. 14	937	1.27	6990	330		4.5	1570	12
Water year	1956 Sept. 1-30, 1956	1400	1.0																				
Minimum.	Jan. 1-31	1423 3627	12		128 102	32 21	28		126		315	448	- 5	- 5	.20	1280	1.74	4920	450	346	5.7	2190	7.8
Weighted	average	3550	11		106	23	19		121		228	305	. 3	. 9	. 17	954	1.30	9340	341	242	4.6	1600	
Water year							20.20		122		248	346	. 4	1.1	. 17	1030	1.40	9870	359	259	5.0	1720	
	Dec. 1-31, 1956	677	12		134	32	31	1	123		342	485		~		1000	120200	8425					
Minimum,	June 1-30, 1957	66910	11		78	15	13		107		165	202		1.8		1380 696	1.88	2520 125700	465		6.3		7.8
Weighted a	average	10890	11		89	18	16	7	112		195	258		2.2		840	1.14	24700	256 296	168 204	3.6	1130	
Water year																		21100	230	204	4.2	1370	-
	Sept. 1-30, 1958	1614	9.4		100	24	22	0	148		209	345		- 5		981	1.33	4280	348	226	5.1	1700	0
Weighted	Oct. 1-31, 1957	5720 4320	15		85	20	14		132		173	225		1.2		733	1.00	11320	294	186	3.6		8.0
		1320	11		91	20	17	1	136		185	268	-	1.0		837	1.14	9760	309		4.2	1400	0.4
Water year	Aug. 1-31, 1959	4623	10		110	0.7	1000				0.000												
Minimum, C	Oct. 1-31, 1958	1823	8.8		112 99	27	26 23		$\frac{131}{138}$		259	408		- 4		1140	1.55	14230	390		5.7	1900	7.4
Weighted a	average	2298	9.4		104	28	25		138		218 246	365 390		. 5		1020	1.39	5020	346		5.5	1770	8.2
Water year								251	-		210	330		. 0		1100	1.50	6830	374	264	5.7	1880	
Maximum, S	Sept. 1-30, 1960	1930	12		113	28	26	4	153		266	400	C	9		1100					140000		
Minimum, !	May 1-31	2703	8.6		99	25	18		155		223	280		2.5		900	1.58	6040 6570	397		5.8	1960	
weighted :	iverage	5203	9.5		101	26	22	n	129		238	343		1.7		1020	1.39	0370	350	223	4.3	1530	7.5

Table 5.--Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas--Continued

(Analyses listed as maximum and minimum were classified on the basis of the values for dissolved solids only; values of other constituents may not be extremes. Results in milligrams per liter except as indicated.

					:			Bi-						Dissolved solids (calculated)	d solids	Har as	Hardness as CaCO,	& :	Specific con-	- 0
Date of collection	Mean discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	clum (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Fo- tas- sium (K)		Coar- bon- ate (SO ₄)	Chloride (C1)	Fluo- ride (F)	N1- trate (NO ₂)	Bo- ron Milli- (B) grams per liter	i- Tons	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	ad- gorp- tion ratio	E 8	Hg
						102. R	RED RIVER	AT	DENISON DAM	NEAR	LSON	DENISONContinued	р							
Maximum, Sept. 1-30, 1961 Maximum, Oct. 1-31, 1960 Weighted average	3593 12040 4299	11 9.4 9.9		120 110 117	37 33	300 263 278	830	138 128 134	312 286 297	470 410 431	0.3	2.8	1320 1170 1230	20 1.80 70 1.59 30 1.67	0 12810 9 38030 7 14280	452 410 428	338 305 318	5.2	2210 2010 2100	7.7
Maximum, Oct. 1-31, 1961 Minimum, Sept. 1-30, 1962 Weighted average	4814 3772 4527	9.9 11 8.9	£ \$	121 100 111	37 30 34	297 238 253	138	130 139 136	316 256 277	470 360 403	***	1.2	132	320 1.80 060 1.44 150 1.56	0 17160 4 10800 6 14100	454 373 420	348 259 308	5.7	2220 1800 1980	7.3
Maximum, Sept. 1-30, 1963 Minimum, Apr. 1-30, 1963 Weighted average	1501 2862 3029	0.00	1971411	105 98 99	31 27 29	22	225 199 211	150 140 133	249 236 244	350 302 326	404	1.8	96	050 1.4 941 1.2 989 1.3	43 4260 28 7270 35 8090	390 356	266 241 256	5.0 4.6 4.8	1820 1570 1670	7.7
Maximum, Sept. 1-30, 1964 Maximum, Nov. 1-30, 1963 Weighted average	1177 803 1510	8.8	<u> </u>	===	3 2 2 2	298 253 267	2338	160 130 135	300 285 290	440 402 420	444	1.22	1270 1160 1200	70 1.73 80 1.58 00 1.63	3 4036 8 2515 3 4900	418 421 422	314	5.43	2060 2000 2040	7.3
Maximum, Nov. 1-30, 1964 Minimum, May 1-31, 1965 Weighted average	815 1939 1943	6.2 4.5		108 96 101	30030	283 209 238	83	120 135 135	296 236 251	440 325 373	646	1.8 2.2 1.1	10.	230 1.67 969 1.32 070 1.46	7 2710 2 5070 6 5610	414 363 376	316 252 266	6.1	2060 1720 1850	7.4
Maximum, Mar. 1-31, 1966 Minimum, Oct. 1-31, 1965 Weighted average	1310 2056 2813	3.2		108 101 110	30 30	2 2 2	261 247 255	127 126 138	272 246 264	442 400 403	444	1.8	110	1180 1.60 1090 1.48 1130 1.54	0 4170 8 6050 4 8610	393	289 282 284	5.7	2040 1940 1980	7.0
Mater year 1967 Maximum, Feb. 1-28, 1967 Minimum, Aug. 1-31, 1967 Weighted average	1931 2840 2339	2.3	10 m 10	114 99 106	28 28 28	273 220 248	000 004	123 122 123	292 228 253	438 355 404	1.4.4	.8 1.0	12	1220 1.6 996 1.3 1110 1.5	66 6360 35 7640 50 6990	404 350	303 250 279	5.3	2070 1720 1920	7.1
								111. RE	RED RIVER AT	INDEX,	ARK.									
Water year 1961 Maximum, Sept. 6-10, 1961 Whinhaum, May 9-14, 1961 Weighted average	3112 25000 10190			118 29 75	37 4.3 20	87.7	255 30 139	172 76 116	275 38 161	405 38 219		1.0	12	260 1.7 185 .2	71 10590 25 12490 99 26000	445 90 269	28	3.5	2020 306 1200	7.80
Maxkmum, Oct. 21-31, 1961 Maximum, June 6-9, 1962 Weighted average	6577 14600 10930			111	111	5,2	245 21 111	136 92 116	268 24 122	405 24 177		111	11 19	180 1.6 157 609	60 20950 21 6190 83 18000	430 88 0 230	318	1.0	1940 271 972	7.7
Maximum, Sept. 12-14, 1963. Minimum, May 1-6 Weighted average	2300 25883 6970			111	111	N	217 40 96	148 104 121	245 47 110	348 62 149		111	10	090 1.7 292	.48 6770 .40 24600 .73 10120	396 138 218	275	1.5	1740 476 887	8.3

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas

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	Н		8.2		800			7.5	7.1	7.5		100	1		1 - 0			7.0	40		8.2		6.5	9.6	6.8	9.9	00.0	9	7.4	60	1	1
Specific	duct- ance (micro- mhos at 25°C)		804		727					466		055			1			2920		- 1	414				380 6			950 7				
	dium ad- Borp- tion ratio				4.6			1.4	0.0	1.2		c.				*	011	1.9					9.	ıç ı	2 12 1		~ 67		86			1.00000
	Non- car- bon-		0		0.0		- 1			00		479 9			65	350 2	7	530	-		0		0 0		00				34 1.	10		9
Hardness as CaCO,	Cal- clum, Mag- ne- stum		238		30			88	42	157		634				_			-	3	66		7	510	169							0000
	0.0% 8		2				1					9			8	14	13	1620	14	,	1		Ξ:	22	16	7 6	17	24	258	163	ŀ	0000
olids ed)	Tons per day																															
Ussolved solids (calculated)	Tons per acre- foot		0.64		0.62			0.40				1.65			2.22	3.50	3.24	3.64	01.0	00			0.22	. 23	.30	12.	25	į.	1.1	11		
on o	Milli- grams per liter		472	CANYON	455		200	157	233	265	CANYON	1210	The state of	CANYON	1720	2570	2380	2680	0000	934	100		164	169	221	304	381	580	481	235 166		
	Bo- ron (B)		0.11	NEAR CA							NEAR CA		WEAD OF							0.05	20.										EY	
	Ni- trate (NO ₃)		2.0	DAM NE	8.5		0 0		1.0	2.0	PARK N	0.0	DADY V		0.5	0.	0.0	90	THETA				0.5	1.2	ci a		. 2	25	1.0	2.2	R TURKEY	
	Fluo- ride (F)	ER	2.0	STOCKTON	5.0	z		2 00	1.1	1.5	DURO 1	3.1	n varia		!	5.4	2.7	2.6						. 6	010	- 10	ın.	6.6	63	1.3	70 NEAR	
	Chloride (Cl)	R UMBARG	27	ABOVE STOC	32	NEAR CANYON	99	6.8	17	25	PALO	42	DATA	LUTY	36	80 0	55	52	I B		T WED	STEVERTON	4.4	2.0	3.6	8. 2	24	40 KG	46	3,6	HIGHWAY 7	115000
	Sulfate (SO ₄)	BUFFALO LAKE NEAR UMBARGER	7.3	RIVER ABO	96	STOCKTON NEA		16	8 6 6	49	RIVER ABOVE	899	RIVER RELOW	- 11	1040	1660	0691	1770	AT		WEAD	NEAR	138	11	13	86	125	190	112	38 9.6	STATE H	5980 17
Car-	-	FALO 1		RED							RED RI		RED RT						E DRAW		Adday.										ER AT	100
	bon- ate (HCO ₂)		372	N FORK	288	. LAKE	237	152	180	85	FORK	197	FORK		90	7.5	44	02	TH TULE	233	THE	101	47	6.5	142	32	82	376	7.4	187	RED RIVER	119
Po-		23	14	G TOWN		7			0.6	0	TOWN		TOWN						NORTH .	8.8			m	-	7 -	63		15 3	63	7.4 1	LITTLE R	-
	Sodtum (Na)		81	PRAIRIE DOG	122		43	15	34	41	AIRIE DOG	133	AIRIE DOG	- 1	68	110	162	137	11	17		1	93	13	13	58	69	115		5.7	16. LIT	73200
Mag-			3.1	6. PR	12		20	8.5	1.4	1.7	9. PRAI	42	10. PRAI		60	100	82	70		12		0	49	8.0	6.1	21		49 1		6.5		1190
Cal.	(Ca)		44		32		42	34	35	29	91	185	10		314	468	510	462		48		2.0	23	37	34	50	40	51	44	41		2000 1
	(Fe)		0.09																	0.17										Ì		2(
	Silica (SiQ ₂)		4.6		13		0.3	7.0	. 1.	6.		38		0	428	9	7	7		4.8		0	100	10.0	12	2	40	19				
	Discharge (cfs)				b0.08					-		0.57 3			0.27	.15.2	.40	b.28 2				13 3 1	20		186	.70 2	. 72	.2		185 11		b0.02
Date	uo		1951		24		4, 1965	1966	20, 1967			1961		1950	1961	1964	1965	24.2		1951		1964				y 14		1966				1959
	ğ		0,		Apr. 26 Aug. 24		Feb. 4,	01 .0	Jan. 20	Aug. 25.		c. 1,			c. 1,	c. 2,	b. 4,	8 24		5.5		Sept. 24	1. 7,	11.	ie 13.	ly 14.	. 16.	. 10,	10	25.		. 24,
	1		May	-	Ap		Fe	Au	Ja	Au	1	Dec.		100	D o	De	Feb.	Aug.		May		Ser	Jan.	June	June	July	Nov.	Mar	Aug	Aug.		Mar

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

Mag-	Mag-	Mag-	Mag-	Mag-
Sodium (Na) s	Silica Iron ctum sium (Na) sium (SiO _a) (Fe) (Ca) (Mg) (Mg) (K)	Silica Iron ctum sium (Na) sium (SiO _a) (Fe) (Ca) (Mg) (Mg) (K)	Silica Iron ctum sium (Na) sium (SiO _a) (Fe) (Ca) (Mg) (Mg) (K)	Silica Iron ctum sium (Na) sium (SiO _a) (Fe) (Ca) (Mg) (Mg) (K)
17. PRAIRIE DOG				
1000	9	1000.5	1000.5	9
18.	.81	18.	.81	18.
4.1 1510 283 17700 144 1500 270 17400 139 1460 302 16800 230 5.05 14 1470 275 16200 91	4.1 1510 283 17700 1500 270 17400 1460 302 16800 5.05 14 1470 275 16200	4.1 1510 283 17700 1500 270 17400 1460 302 16800 5.05 14 1470 275 16200	4.1 1510 283 17700 1500 270 17400 1460 302 16800 5.05 14 1470 275 16200	1510 283 17700 1500 270 17400 1460 302 16800 14 1470 275 16200
19. BAYLOR	1	1	1	1
18 179 28 11 83 216 20 16 98	18 179 28 11 216 20 16	18 179 28 11 216 20 16	18 179 28 11 216 20 16	179 28 11 216 20 16
20.	20.	20.	20.	20.
11 96 12 6.7 61 5.0 0.1 100 9.7 4.6 46	0 0.1 100 9.7 4.6	5.0 0.1 100 9.7 4.6	5.0 0.1 100 9.7 4.6	0 0.1 100 9.7 4.6
21. SALT CREEK	1. SALT	1. SALT	1. SALT	1. SALT
1030 213 5990 110	213 5990	1030 213 5990	1030 213 5990	213 5990
22. PRAIRIE DOG	. PRAIRIE	. PRAIRIE	. PRAIRIE	. PRAIRIE
15 1750 459 21400 106 1860 407 23500 86 9.8 1740 393 20600 96	15 1750 459 21400 1 1860 407 23500 9.8 1740 393 20600	15 1750 459 21400 1 1860 407 23500 9.8 1740 393 20600	15 1750 459 21400 1 1860 407 23500 9.8 1740 393 20600	- 1750 459 21400 1 1860 407 23500 8 1740 393 20600
23.	23.	23.	23.	23.
147	574 147 206	574 147 206	574 147 206	574 147 206
606 134 258 1 408 116 96	119 606 134 258 1 408 116 96 1	11.5 19 408 116 96	11.5 19 408 116 96	11.5 19 408 116 96
35 622 138 167	35 622 138 167	1.3 35 622 138 167	35 622 138 167	1.3 35 622 138 167
22 608 141 177 15 568 130 149	22 608 141 177 15 568 130 149	1.24 22 608 141 177 2 06 15 568 130 149	1.24 22 608 141 177 3 06 15 568 130 149	1.24 22 608 141 177 2 06 15 568 130 149
2.68 29 568 125 137 77 3.57 19 574 117 196 157 b.08 390 187 125 247	2.68 29 568 125 137 3.57 19 574 117 196 b.08 390 187 125	2.68 29 568 125 137 3.57 19 574 117 196 b.08 390 187 125	2.68.29 568 125 137 3.57 19 574 117 196 b. 08 390 187 125	2.68.29 568 125 137 3.57 19 574 117 196 b. 08 390 187 125
16 184 44 46	16 184 44 46	16 184 44 46	16 184 44 46	16 184 44 46
24				
721 197 2670				

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

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-,.	Нď			7.8					10			7.9	1	7.9	7.3			7.8	7.8	7.9	7.6	7.7	7.6	7.9	7.9	6.7	7.9	10		7.7		7.6		7.7	1	6.2
Specific	duct- ance (micro- mhos at 25°C)			4760	4550	5220	5360	0010	930			3550	3570	3300	3340			3590	3810	3950	4030	4010	3980	1830	4050	4170	4290	3570	4020	3030	4010	4210				329
- So-			1	ľ		4.9	4.9		0	- 10		1.8	į	2.0	1			1	1	: :		2.8	9.5	1.7	3.0	7.7						3.5		1	15	2.7
	Non- car- bon-			1940	1 1	1980	2020	2	18			1780	ŀ	1730	1730			1570	1830	1810	1910			760		2000				1350		2070		377		337 2
Hardness as CaCO,	Cal- ctum, Mag- ne- stum			2080	1 1	2100	2180	200	86			1950	1	1840				1640	1950	1920	1970	1970	1980	848		0902	1900	1700				2150 2		531		652
lds	Tons per day																																			
Dissolved solids (calculated)	Tons per acre- foot			5.03	1	5.54			0.22			4.15	1	3.92	ŀ			4,53	4.42	4.39	4.73		4.43	1.90	4.00	4.03	4.64	3.88	4.57	3.35		1		1.56	100	1.60
Dist	Milli- grams per liter			3700	1	4070	4200		163		2000	2990	1 0	2880	1			3330	3530	3230	3480	3300	3190	1400	0000	07+0	3410	2850	3360	2460	3710	3700		1140	100	1180
	Bo- ron (B)															AH	l																			
	Ni- trate (NO ₂)	GROESBECK		0.0	5.9	0.9	0.0	TT	4.1		ti ci	1.8	1.1	9.00	1	QUANAH		1.2		2 8	3.5	3.0	. t	3.5		4.4	3.8	5.2	2.2	00 0				8.7	a .	0.0
	ride t							GOODLETT	0.2	ACME						3 NEAR		!	1	1	į	1	! !	1								4	7		-	10
	Chloride (C1)	NEAR NORTH	0.00	530	570	805	780	NEAR	4.0	CREEK NEAR	909	290	270	235	240	HIGHWAY 283		420	420	398	428	418	400	190	47.9	415	535	375	450	298	538	260	AT ODELL	150	170	148
	Sulfate (SO ₄)	CREEK	4000	1830	1810	1890	1960	BECK CREEK	32	GROESBECK CR	1750	1690	1700	1730	1750	STATE HI	-	1780	1770	1780	1940	1260	1710	714	1870	1740	1740	1540	1830	1330	1980	1950	WANDERERS CREEK	415	524	446
Car-	ate (CO ₃)	ESBECK						GROESBECK								AT																	ANDER			
	car- bon- ate (HCO ₃)	NORTH GROESBECK	041	0 !	i	151	188	SOUTH	86	SOUTH	157	203	137	93	172	K CREEK	000	150	102	125	0	141	190	106	72	1	132	204	7.5	135	151	96	29. W	881	901	384
-0-		NORT				0.0		26.	2	27.	1				~	GROESBECK											0	0			5.3	9.7				
	Sodium (Na)	25.	200	1	i	520	522		2		197	179	20	151	198	28. GRC	1	27	292	277	000	282	246	307	284	268	359	247	320	209	328	17		126	122	158
Mag-			197	; ;	1	125	1		4.4		66	102	84	26	ī	2	40	122	117	108	, , , ,	118	115	52	131	F	102		109	83				99	1 1	74
ć			400	1	I	636	1		32				009	595	;					590				254				632	585					104	!!	139
	(Fe)																													i er						
	Silica 1 (SiO ₂) (1.1	0.	9	16			26		23	17	00	0	;		90	T.	5	18		מו מ	3	14	61	10.	14 28	1 0	26	6.0	3.5	5.0		8.4	0 ~	30 +
	Discharge ((cfs)		5 93 2	4.14	2.84	2.33.2	1.90 -		6.25 2			4.00 1			4.36 -		6			14.3 1	0 0	9.92	9.66 2	6.50 1	.34 1	11.0 1	.56	1 1	4.72 2	4.0	2.3	1.6		4.54	77.8 2	2.04 18
Date	uo		1951	1952	1953	20, 1958	3, 1961		1962		1951	22, 1952 8 1953	1957	1958	3, 1961		1950	, 1951	*********				952	1953	**********		21.	1958	1060	1965	May 31, 1966	71.07.67.44.0		1950	1952	1953
q	Col		v 16	. 22	80	. 20	65		7.		y 16,	222	. 19	. 20.			100	1.5	15	y 16.	-	18	22,	13,	11	Dec. 8	21.	10,	00	8.	31,			20,		1
			Lul	Jan.	Dec.	Feb.	Oct.		June		July	Jan.	Feb.	Aug.	Oct.		Apr	Jan.	Mar	July Aug.	1	Nov.	Jan.	Jan.	May	Dec	Oct.	Jan.	Aug. 2	Dec.	May	July		Feb.	July	Jan. 13 May 14.

See footnotes at end of table,

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

	Cal- Mag- Po-
sium (Na) sium (Kg) (K)	(Ca) (Mg) stum (Mg) (K)
29. WANDERERS	
1	1
67 155	67 155
186 64 131 5.6 263 103 248	64 131
38	388
19	19
31 20 23	20
1	30
5.6 90 da	90 43
24	2
18	18
	222
67 22 3- 61 24 36	22
21	21
21	21
40 17 4	171
48 24 81 42 21 48	24
21	21
13	13
61 17 44 39 6.6 18	17.
18	18
11	1 1
52 18 74	18
34. SALT FORK	SALT
74 28 118	28
35	35
112 47 276	4.3

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

(

	Н		7.7		8.0			1 1	7.6	2.0	1	8.5	8.0	8.7			7.5		8.4		8.2		8.1	1	1		7.7
Specific	duct- ance (micro- mhos at 25°C)		1300		1490			890	921	933 858	942	864	891	525	961	806	767		373		2430		2820		16500		3050
	dum ad- gorp- tion ratto				3.4			1 1	!	1.4	1	1.9	* 65	0.1	6.	. 5	0.0		0.3		4				F		2
- 7	Non- car- bon-		156		168			158	167	202	ļ		194	280	157	174	156		2		300 1		1190		3140		1150 3.
Hardness as CaCO,	Cal- cium, Mag- ne- stum		352		334		07.0	382	372	371	1	273	280	181	320	268	298		168		1010		1320 1		3310 3		1290 11
ids d)	Tons per day																										
Dissolved solids (calculated)	Tons per acre- foot		1.17		1.18		10 0	77.	.95	.76	;	.81	.74	10.	88	.70	80.1				2.39		3.06		16.3		
Dis	Milli- grams per liter		858 995		871		200	565	869	561	1	599	542	9	649	513	549		223		1760		2250 3030		12100		
	ron (B)																				0.19		0.37		-		
	Ni- trate (NO ₂)	NDON	1.0	A LAKE	0.5			10	12	6.3	6.9	0.9	9.5	2 0	8.9	6.3	5.5	0	4.5	IN	9.0	LEY	0.0		2.3	III	
	Fluo- ride (F)	CLARENDON		LELIA		EDLEY			1	11	Į	1 1	1 1	-	1	1 0	7.	ALANREED	0.4	MCLEAN		OF HEDLEY		MCKNIGHT		OF QUAIL	
	Chloride (C1)	EAST OF	174 246	NORTH OF	240	K NEAR HEDLEY	54	52	52	48	54	48	51	23	57	56	21	NEAR AL	4.4	SOUTH OF	245	NORTHEAST	260	OF	4950		315
	Sulfate (SO ₄)	MILES NORTHEAST OF	216	D RIVER	221 150	LELIA LAKE CREEK	221	181	217	223	205	237	217	225	231	199	202	SH CREEK	11	CREEK	332	CREEK NO	1210 1360	CREEK NORTH	66	RED RIVER NORTH	1240
Car	ate (CO ₂)	8 MIL		FORK RED		SLIA L												WHITEFISH		WHITEFISH				GYP CRI		FORK	
Bi-	bon- ate (HCO ₃)	CREEK	240	SALT F	202	39.	198	273	250	191	100	143	105	1	199	142	188		202		182	WHITEFISH	196 124	44. G	200	SALT	169
Po-		SADDLERS	9.2	37. 8	4.0	es.	1	01.5	2 6	m	1		0) 0)			6.3	3.9	41	9.0	42		43.				45,	
	Sodium (Na)		145		184		ia	52	0.4	12	1	ů.	322	59	62 29	57	61				178 66		307		3120		268
Mag-		36	36		32		27	26	29	28	108	28	28		25						83		114		304		98
	(Ca)		93		81		95	110	101	78	19	73	51	1	87	70	80				269		354		824		355
	(Fe)																				.,						.,.
	Silca (SiQ _g)		28		24		!	15	26	41	31	44	50	8	32	4	6				26		15				20
	Discharge (cfs)		6.0		b2.0		3.0	10					8.22 5	6.82 3	6.31 3	6.80	.06				S.I		0.03 1				b0.25 b1.5 2
Date	of		. 1951		1951 1959		1950	9, 1951		, 1951	22, 1952		1903		1957	1964	1967		1962		1951 1962		1951		1951		1959
Q	coll		21		22.		13,	9, 1	21.	4, 1	222	8	2 2		13	10	21,		12,		21,		12,		1, 19		222
			Feb.		Mar.		June	Jan.	Mar.	Oct.	Jan.	Oct.	Aug.	Nov.	Feb.	Apr.	June		June		Mar. June		Feb.		Jan.		Mar.

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

2	bH tr			2.7	7.8	- 1	7.9		7.9		7.6		8.7 7.4 7.4 7.6			2.5 7.8 7.6		6.8 8.1 8.1 8.3
Specific	duct- ance (micro- mhos at 25°C)		2820	2820	2820	1160	3560		179		148		886 929 986 961 1140	1070		1750 2130 1310 2530 1970	2320 2050 2559 2740 2010	259n 889 1170 2260 1510
-os	ad- sorp- tion ratio		11	7.0	. 7	1	4.7						2 4 - 5 5	33.2		11151	1 0 4 4	23.7 23.5 4.0 4.0 4.0 4.0 4.0 4.0 4.0 4.0 4.0 4.0
CO3	Non- car- bon-		1740	1700	1690	090	858				-		68 52 63 67 98	73 90 80		350 540 617 1120	474 624 660 11110	1200 365 226 794 272
Hardness as CaCO ₃	Cal- clum, Mag- ne- slum		1860	1790	1740	800	980		74		69		216 210 254 244 248	277 228 254		514 678 710 1210	620 696 790 1190	1300 450 111 766 387
ilds	Tons per day																	
Dissolved solids (calculated)	Tons per acre- foot		3.59	3.48	3.43	1							0.70	111		1.44 1.89 1.39 2.58	2.12	2.84 .87 1.01 2.07
Dis	Milli- grams per liter		2640	2560	2520	:					96		886 549 590 559 624	647 631 662		1060 1390 1020 1900	1290	2090 642 744 1520 860
	Bo- ron (B)							STON			0.27	LEAN						
	N1- trate (NO ₂)		6.8	9.0	8.0	1 111		BOYDSTON	5.0		2.2	273 NEAR MCLEAN	0		CK	0.5.5.	1.85	0.12
	Fluo- ride (F)	VGTON				ariinaariaa		NEAR (CHO	0.1	73 NE	0 8 8 8 1 8	x c. x	SHAMROCK	11111	11116	က်ယ်က်ကော်ပ
	Chloride (C1)	AR WELLINGTON	75	78	7.5	. 100	010	HIGHWAY 70	2.9	NEAR JERICHO	1.7	STATE HIGHWAY 2	121 117 125 120 175	141 171 169	NEAR	315 340 41 210 308	388 360 438 528 84	245 22 130 402 292
	Sulfate (SO ₄)	CREEK NEAR	1720	1690	1660	BIVER	156	STATE	8.0	MCCLELLAN	2.7		123 122 125 118 123	133 136 137	RED RIVER	268 489 593 1040 428	445 390 530 565 1080	1130 357 238 584 186
300	ate (CO ₃)	DOZIER				K BED		EK AT		LAKE Me		EEK AT			FORK			
Bi-	car- bon- ate (HCO ₃)	46. DO	144	99	69	да ковк		AN CREEK	98	SO LA	83	LAN CR	160 193 230 216 183	249 168 213	NORTH	200 168 113 107	178 89 159 103	128 104 226 76 141
Do		+	74	65	7.1	YORTH	335	MCCLELLAN		15	1.2	MCCLELLAN CREEK	99 112 112 105 4.6	3.0	52.	207 28 28 117	202 252 310 63	152 20 95 218 155
	Sodium (Na)				1 110	48	1 1 1	49. M			1.2	51.	121	124 132 134		- 2	61 61 67	- 8-
Mag	nag- ne- sium (Mg)		8.4	25 25 24 44	84	;	112				2.8		22 17 18 19 21	20 19 20		35 22 57	36 45 49	55 10 21 54 20
	Cal- ctum (Ca)		585	584	560		208				23		50 56 72 66 65	78 69		148 194 248 392	205 235 105	130 164 130 218 122
	Iron (Fe)										0.23							
	Slitca (SiO ₂)			34							10		22 24 24 24 24	3 2 2		22 32 28 26 21	26 20 42 	111 12 18 7.5
	Discharge (cfs)			223			99						11.4 12.1 11.2 10.9	- 010		162 12.0 3 2.19 22 .73 28 18.5 2	13.0 15.2 5.89 b6.0 12.9	81.9 2080 18.3 18.3
	Date of collection		1950	Mar. 5, 1953.	12, 1955	1300.	1959		, 1950,		1951		6, 1965 22 20 22 8, 1966			26, 1951 9, 25 18, 1952	Dec. 8, 1953 Jan. 9, 1958 Feb. 13. Mar. 22, 1959 May 27, 1964	28. 29. 14. 6. 1965.
•	Col		le 14.	. 5.)t. 18	1. 12.		. 22		ot. 12		ie 28.		22. 22. 22. 8.	1, 17. 1y 25		25. 25. 1y 25.	22. 27.	May 28 June 14 Jan. 5. 19 Oct. 18
			Jur	Sep	Jan.	100	Mar.		Sept		June		Jan. June Oct. Nov. June	Jan. July Sept.		Feb. July Nov. Jan.	Jan Jan Man May	May Jur Jar Oct

Table 6 .- - Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

	Н		7.1 7.1 6.9 7.4	7.0		7.7		8.2		8.1	8.2	7.8	8.1	8.0	8.2	8 : 2	2	1	8.2	0 1		6.6	7.6
Specific	duct- ance (micro- mhos at 25°C)		2470 1480 2070 2960 2710	1170 2200 1750		499		593		718	619	717	448	560	426	435	506	206	385	526		860 488 709	716
So-	ad- sorp- tion ratio		3.3	3.5						1	1 1	1 1	1	1.7	1.6	1.5	N 1	ia	1.2	1.1		4.5	1.1
ess CO ₃	Non- car- bon- ate		736 685 514 872 608	390		00		00		54	17	00	0	0	00	0	1	10	0	0		127 22 46	30
Hardness as CaCO ₃	Cal- clum, Mag- ne- stum		855 822 635 1020 756	507 708 494		238		227		310	275	278	155	162	132	137	161	308	138	214		346 220 281	294
lids σ)	Tons per day																						
Dissolved solids (calculated)	Tons per acre- foot		11111	111		0.40		0.50		99.0	. 55	.62	.40	.43	.49	.37	6 1	1 85	.35	. 1 4		.39	.61
Dis	Willi- grams per liter		1600 1150 1270 2000 1680	811 1420 1040		294	LIE	368		488	458	457	295	318	362	272	1007	423	258	326		598 287 443	450
	Bo- ron (B)	p			EETIE		MOBEETIE																
	Ni- trate (NO ₃)	Continued	-000 000000	2.2	OF MOBEETIE	2.0	OF	2.0		2.5	3.8	2.0	3.0	0.8	1.5	3.5	1.0	1.0	5	1.2		8.1.8	1.8
	Fluo- ride (F)	Con	5.0	6.0	WEST O		SOUTHEAST		WHEELER												ron	0.7	7.
	Chloride (Cl)	SHAMROCK	375 40 378 528 580	109 425 360	HIGHWAY 152 W	10	152	24	NEAR WHE	26	2 5 4	238	18	30	21	26	19	16	13	243	NEAR KELTON	32 1 32	24
	Sulfate (SO4)	RIVER NEAR	618 661 366 712 416	364 432 175	STATE HIGHW	14	HIGHWAY	26	ER CREEK	104	79	75	25	38	252	22	20	18	25	2 2 2	ER CREEK	202 44 100	197
Car	ate (CO ₃)	RED RI			AT STA		STATE		SWEETWATER												SWEETWATER		
Bi-	car- bon- ate (HCO ₃)	FORK R	146 168 148 176 180	142 132 237	CREEK	298	EK AT	308		312	336	340	222	230	326	201	0 1	391	200	285	SWE	267 241 287	322
-0d		NORTH	202 8.3 6.0 4.9 7.1	5.2		17	TER CREEK	41 20	55.	13	50	55 39	37	Z !	41	42	66	33	33	2.9	56.	61 18 46	15
	Sodlum (Na)	52.	24 188 272 293	60 211 180	SWEETWATER		SWEETWATER			77.9			*2	***						42			
Мак-	slum (Mg)		47 21 30 59 46	19 40 23	53.	11	54.	14		22	22	21	14	12	14	15		13	11	15		21 8.6 18	18
ī	ctum (Ca)		265 295 205 310 227	172 218 160		74		68		88	81	77	39	1 40	30	30	1	104	37	60		104 74 83	102
	(Fe)																						
	Silica (SiO ₂)		19 13 13 23 18	111 17 18		17		18		18	27	31	20	35	34	50 50	36	32	40	42		28 12 25	27
	Discharge (cfs)		10.8 35.6 34.8 4.79 28.4	92.0 77.0 137		2.73		8.84		11.5	1 8	39.2	8,48	10.7	16.9	3.83		13.5		6.14		11.9 224 10.8	7.17
Data	oo		. 22, 1965. y 24, 1966. it. 19.	28. 5.		13, 1951		. 13, 1951		12, 1951	27	27	y 25.	21, 1952	171953	May 11.		24, 1957	- 0	13		. 8, 1962. . 13, 1963. . 19.	e 11.
	-		Nov. 2 July 2 Sept. Jan. 1 Apr. 1	June Aug.		Feb.		Feb.		Feb	Feb	Feb.	July	Jan.	Apr.	May	Dec	Apr.	Oct	Feb		Aug. Aug. Nov.	Jan

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

as CaCO ₃ So-con-	Non- ad- car- dorp- bon- tion m ate		455	741	357	38	10	0 0	- 01	-																	-		-	-	-		0	C	-	00		0
aco, So-	Non- ad- car- gorp- bon- ratio					7	795	144	91	1250	715	720	1150		1410	1240	1620	1050	1320	1320	1240	1220	1240	1220	1250	1270	1270	1230	1120	1100	1170	150	1130	1300	1280	1400	1380	128
dness aco,	Non- car- bon-			9 6	6.	1.3		1.4	. 1	1	1.2	2.5	1.4		1	1	1	1	į	i	1	1 1	1	1	! !	;	1		1	Ĭ	;	1.5	1	1	1.6	1.6	1 3	1.8
Hardn ab Ca(0 5	465	c	46	72	291	206	367	44	4.5	326		1	i	1	1 1	1	1	1	1 1	1	1	1 1	1	1 6	334	334	366	396	342	1	1	380	391	376	342
	cdum, Mag- ne- stum		170	313	132	306	318	480	378	526	291	236	505		681	541	707	474	1	572	504	517	492	579	524	558	424	498	421	445	487	468	360	1	442	594	160	525
(P	Tons per day																																					
Dissolved solids (calculated)	Tons per acre- foot		.36	1.41	Co.	1	;	!	1 1	ł	1	1	1		t t	1	1	1 1	;	1	1	1 1	ļ	1	: :	1	1	1.29	1.11	1.09	1.07	1.13	0 1	1	1.17	1.34	1.62	1.26
Dis	Milli- grams per liter		266	1040	211	502	199	778	1000	i	454	397	837		1010	923	1 0	743	1	866	864	891	868	853	043	946	833	948	815	801	787	833	000	- 1	863	982	0611	930
	ron (B)																																					
;	Ni- trate (NO ₅)	panu	0.0	. c	3.0	1.0	1.5	2.0	0.1	1	1.0		2.5		2.2	5.0	9.0	o. 11	8 6	14	4.0	3.8	1.2	7.4	1 10	0.00	5,3	6.5	0.4	4.2	6.1	20.0	0 -	3.0	2.5	1.5		5.5
	Fluo- Ni- ride trate (F) (NO ₃)	Conti	0.4	9,0	c e	00	80	iù i	C.	;	. 7	6,0		×	1	ì	ì	1 1	1	1	ŀ	1 1	1	;	1	1	1	1	1 1	1	1	1	!	1	1	1	1	0.0
	Chloride (Cl)	SWEETWATER CREEK NEAR KELTONContinued	13	78	12	35	31	42	22	32	28	26	4.5	SHAMROCK	92	86	82	90	80	94	90	92	06	83	95	91	92	68	8 G	92	16	94	161	98	98	100	112	103
	Sulfate (SO.)	EK NEAR	27	528	112	121	127	355	224	433	86	69	400	CREEK NEAR	525	103	305	655	471	352	359	343	351	310	136	346	324	328	355	354	337	360	373	377	363	425	607	370
Car	ate (CO,)	ER CRE												ELM CE																								
	car- bon- ate (HCO ₃)	ETWAT	234	196	314	318	300	230	180	194	302	284	224	57.	162	241	277	306	104	260	196	242	204	264	100	270	166	269	166	96	111	153	102	5 1	76	248	161	274
Do-			23	96	00 1	2.9	2.3	2.6	3.0	1:1	2.3	6.0	2.8		0	001	1	57		70	3	8.1		38	1:	72	**	89	98	35	41	78	7.33	: :	2.5	89	04	73
	Sodium (Na)	96	m	6	40	52	51	69	66	11	46	47	72		9	10	1	10		12	7	113 04			1.0		-	ж.	~ .								1	
Mag	mag- ne- sium (Mg)		7.4	46	22	19.4	19	34	47	122	18	12	33		37	29	1	36	4	31	28	35	200	31	1.0	29	23	53	28	200	34	30	32	1 1		255	33	28
	Cal- cium (Ca)		56	175	68	91	96	136	166	117	87	7.0	148		212	169	1	260	000	178	156	173	151	181	1	176	132	152	142	124	139	138	129		1.01	197	225	194
	(Fe)																																					
			1.5	30	22	25	22	24	30	1 1	24	23	21 26		1	1	1	1		1 1	1	;		: :	1	11			31	38	40	53	34	96	2 6		25	11
	Discharge (SiO ₂)		346	.17	-	15.2				1.0		32.4	1.80		1	2.54	1.01	.34	10.	2.51	2.32	2.42		3.3	2.71	1.47	2.58	3.0	1.97	2.39	2.17	1.71	1.67	1.35		1.41	b2.0	b4.0
	Date of collection		June 14, 1964.	Oct. 9	Jan. 6, 1965	Oct. 18	May 8 1966	June 8		Aug. 2	Jept. 12	Apr. 14	May 16.		1046	26	26	Oct. 26		Oct. 26	16	Feb. 13, 1947		May 6. June 12.	y 7	Aug. 12.	Dec 1	e 10, 1950	Sept. 1	Sept. 28	May 15 1951	June 28	May 11, 1953		6	Jan. 12, 1955	22	21, 1962.

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

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	Hď		7.9		8.0		7.8		1	!	! !	1 1	8 . 4	x x x x x x	80	3.0	11	1	1	7.7	6.7	8 8 9		1	8.1		20	15	7.7		1	17
Specific	duct- ance (micro- mhos at 25°C)		2390		2770		2670		650	672	717	724	719	778	681	890	791	753	508	245	389	398	ŀ	1		948			926		707	
8 :	ad- gorp- tion ratio		0.9		9.0		1.0																	1	2.8	1	1.8	1	9.1	1 8	1	10
co,	Non- car- bon-		1320		1750		1570		1	10	000	c rs	00	000	00	0	00	0	0 0	00	0	00#		1	28	1	99	10	26	65		1 1
Hardness as CaCO,	Cal- cium, Mag- ne- sium		1450		1880		1670		202	212	218	236	248	224	197	237	210	217	158	256	135	244 244		305	320	1	314	100	316	320	ı	
olids ed)	Tons per day																															
Dissolved solids (calculated)	Tons per acre- foot						3.35		1	0.61	557	. 28	. 61	. 63.	. 57	99.	. 59	.60	. 61	.61	. 32	.36		1 0	.78		62.	1	11	.78	1	7.0
Dis	Milli- grams per liter		2250		2780		2460		1	449	420	429	447	466	416	485	436	442	445	452	238	264 420		459	570	11	580	1	1	577	1	199
	Bo- ron (B)																															
	N1- trate (NO ₃)	LUTIE	3.0		5.2	LUTIE	3.9		0.0	N 0	20.00	0.	0 10	999	0.5	0.	.8	100 to	0 00	2.0	2.0	.00		s:	2 00	9.2	25	00	1	29	1	35
	Fluo- ride (F)	NEAR LU		LUTIE	0.6		9.0	QUE					3.6										SPRINGS		1 1			1 1	1			1 2
	Chloride (C1)	CREEK NE	82	NEAR	09	CREEK NEAR	70	R QUITAQUE	46	51	50	25	51	50	52	63	60	99	909	10	23		ROARING SE	69	95	95	92	73	. 20 X	83	87	88
	Sulfate (SO.)	ABOVE WOLF C	1320	AT MOUTH	1710	BELOW WOLF C	1610	CREEK NEAR	17	43	24 22 22	42	42	57	2 2	45	40	40	40	9.0	18	18	NEAR RO	51	77	77	92	7.4	75	7.7	11	7.5
	bon-Si ate (CO ₂)			CREEK																			SPRINGS									
	car- bon- ate (HCO ₂)	CREEK	156	WOLF (64	ELM CREEK	115	QUITAQUE	04	202	304	84	30	294	287	37	14	15	12	335	77	185		325	294	11	303	180	8 !	315	-	317
-	stum (K)	8. ELM		.69	-			61.	1				2.0								M.		ROARING					0.0	6			
	Sodium (Na)	58	78		57	09	94			99	67		70	79	70	8	9	8 1	67	1.5	28	35	62.	59	73	1 1	75	6.7	99 62	7.4	1	75
Mag	ne- stum (Mg)		99		96		85		1	39	36	36	38	37	37	40	35	37	35	39	14	38		28	31	31	29	82	27	30	!	28
	Cal- cfum (Ca)		472		596		528		1	32	28	35	12	27	18	29	40	26	37			321		76	27	11	7.8	: !	82	42	1	81
	(Fe)											0	60.																			
	Suica (SiO ₂)						13		13	1	1.1	13		1.1	1:1	1 1	1.1:	1 1	1	21	61 4	32		1 00	44	no	38	e 1	3.3	4	t	1 77
	Discharge S (cfs)		b10		b5.0		b4.0 1 b16		1	4.6	4.72 -	4.87	6.6	5.3	9.6	20.00	200	4.3	1	11	11	18.1 3		1.80 6	1.47 4	1.34 43	1.54 3	1.57	1.26	34	1.16 -	1.28 34
	Date of collection		1962		1962		1959		1945	30	1-10	61	1946	Feb. 15	Apr. 15					June 29, 1950		17, 1951		Dec. 1937	1953	1954	1956	110000	Mar. 20, 1959	May 20, 1960		962
	coll		. 21,		21,		22		13,	14-2	1-10	11-1	16.	. 15.	. 15	15	15.	12.	15.	29,	23	v 28.		1937	10,	18,	19,	16.	20,	20, 1	22	9, 1
			Feb.		Feb.		Mar. Feb.		Nov	Nov	Nov.	Dec	Jan	Feb	Apr	Jun	Aug	Sep.	Oct.	June	July	July Mar.		Dec.	June	Jan.	Jan.	Apr.	Mar.	May	Sont	Aug.

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas .- Continued

ific	te pH		60 7.1		00 7.4			90 7.3		70			20 7.2		7	7	9.7 00		1-1	- 1-	00 7.9	1	-	1	7		1-1	00 7.4
Specific	£ 8.		1460		52000		32600	28200		75			6524		205	221	24000		- 114	12800	9 6100		7660		18700		48000	45700
·S	ad- sorp- tion ratio									1		77	13.4							200		1		2.5	1			
CO3	Non- car- bon-		463		4480 3560		3880	3980 3240 2160		1	1680	1540	1120		2530	2720	2700		2000	2100	1760	4140	1180	2960	2070		4360	4160
Hardness as CaCO,	Cal- ctum, Mag- ne- stum		552		4570		1000	3340 2250		1920	1820	1740	1180		2650	2850	0//2		2090	2240	1830	4230	1280	3060	2180		4460	4260
solids ted)	Tons per day																											
Dissolved solid (calculated)	Tons per acre- foot		1.36	CAH							8.72				18.6	20.1	21.2		10.2	11.5	5.92	47.7	1	1 1	1			
Dis	Milli- grams per liter		666	OF PADUCAH			1	18500		5270	6410	5710	4260		13700	14800	24000		7470	8450	4350	35900	i	394	12300		37000	36000
	Bo- ron (B)			NORTHEAST										AH				UCAH										
	N1- trate (NO ₂)		3.0	1000						6.7	'n	10	2.0	РАБИСАН				OF PADUCAH	1	2 1	16	;	1	4.2	; ;	рарисан		
	Fluo- ride (F)	DUCAH		MILES		SS								TH OF										9.0	0.0	OF PAD		
	Chloride (C1)	RIVER NEAR PADUCAH	152 17300	RIVER 14	21700	CHILDRESS	12100	12700 9300 6600	AR VERNON	1610	2540	2050	1640	MILES SOUTH	6410	0669	7660	RIVER 10 MILES SOUTHEAST	2920	3460	1250	18500	1980	9050	5800	SOUTHEAST O	19100	18400
	Sulfate (SO.)		458 3730	PEASE	4100	RIVER NEAR	3270	2330 2190 1730	RIVER NEAR	1820	1540	1560	1070	RIVER 11 M	2230	2300	2460	R 10 MIL	1860	1890	1600	3870	1	2740	1880	MILES SO	3850	3890
	bon- ate (CO ₃)	MIDDLE PEASE		MIDDLE		PEASE RI			PEASE R					1				1000								4		
BI-		MIDDI	109	TO	111		151	124		196	165	235	77	WICHITA	156	150	128	WICHITA	112	183	90	117	128	188	126	SALT CREEK	126	125
	Po- tas- sium (K)	63.	90	IBUTAR	0.0	65		8 8 17	-				9.0	NORTH	0	0	0 0	NORTH W	0	0 0	100	0	ı	22	1	100	0.0	00
	Sodium (Na)		106	SPRINGS TRIBUTARY	13900		7620	8040 5700 4110		108	1610	1360	1050	.69	406	438	4820	70. NC	187	219	3080	11900		5720	3690	71.	1230	12100
	Mag- ne- sium (Mg)		27	SALT SP	243			229 180		15	139	132	41		180	194	191		145	145	133	265	1	185	129		304	285
	Cal- ctum (Ca)		1320	64. 8	1430		1200	1260 1040 705		580	500	478	113		767	821	167		599	661	514	1260	1	920	099		1290	1240
	Iron (Fe)																											
	Silica (SiO ₂)		22				1	110		1			9.6		18	10	77		23	16	18	27	;	9 2	4.8		18	212
	Discharge (cfs)		b200 b1.0		b0.02		1	1.83		1	1	0.35	30.6		13.9	11.3	10.2		5.4	4.17	3.04	.60	1	6.40	4.35		0	0.0
	uo		11, 1950		11, 1959		1959	26, 1967 24.		1942	1951	1967			1951	1952	1953		1951	1952	1953	1956	1958				1951	1958
	Date of collecti				21.		21.	24.		16	10.	27,	24		28.	12	10.		25,	12	13		30.	22.	24		28.	30.
			Sept.		Feb.		Mar.	Jan.	1	Tulto	Apr.	Jan.	Apr.		Nov.	Mar.	Feb.		July	Mar.	Jan.	Nov.	July	June.	Aug.		Nov.	July

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Dally Stations in the Red River Basin, Texas -- Continued

· 63	н			8					7.6		7.7		7 6	7.8	7.1	6.6		7.9				C 10 +		7.8	7.5	7.5			4	8.03
Specific	E 8			36700	45000	47000	48500	44000	2190		25900		19900	23800	29700	29200		20800	21500	19700	15400	955 18400	0000	3140	9530	9880		38500	36800	23000
-08	ad- gorp- tion ratio			1	1 22	0	1	;	8.5		42		26	45	1 1	1		1	34	1.1				1:		13				
CaCO,	Non- car- bon-			3600	4210	4310	1	1680	216		2920		1410	2490	3170	3220		2140	3080	2960	2500	2730		748	2270	2620		4180	4000	2710
Hardness as CaCO,	Cal- clum, Mag- ne- sium		0 20 0	3640	4290	4390	1	1730	291		3030		1490	2620	3300	3290		1		3330		2860	- 1			2750		4280		2810
solids ted)	Tons per day																													
Dissolved solic (calculated)	Tons per acre- foot			34.3	47 8	46.9	1	46.9	1.65	-	23.7		10.6					91.9						2.77	80.6	10.1		37.1	37.3	21.6
DIS (C:	Milli- grams per liter		2400	25600	34000	35300		35300	1210	PADUCAH	17400		7810	1	19800	20300		15600	1	12700	10600	12700		2040	6680	7430		27800	27900	15900
	Bo- ron (B)	TAH								AST OF																	IE			
	N1- trate (NO ₃)	PADUCAH		: 1	1 1	:	1	1 1	1.0	SOUTHEAST		E					TT					· ·		1.0			GUTHRIE			
	Fluo- ride (F)	ST OF								MILES S		РАБИСАН					TRUSCOTT						GUTHRIE				EAST OF			
	Chloride (C1)	SOUTHEAST		13000	17600	18200	18800	7240	19200	12	8440	NEAR		8020	10700	9800	NEAR	6360	6930	5630	4480	5500	AT	605	2350	2400	MILES EA	14300	13300	7730
	Sulfate (SO.	8 MILES	-	3030	3730	3800	3890	2000	196	SALT CREEK	2490	IITA RIVER	1310	2610	3100	2970	ITA RIVER	2590	2890	2480	2190	2580 2750	WICHITA RIVER	989	2020	2470	RIVER 6.5			2220 3290
Car	bom- ate (CO ₃)	MOUTH								BELOW		H WICHITA					H WICHITA										10017			
Bi-	car- bon- ate (HCO ₃)	EK AT	186	51	87	9.1	91	57	92	RIVER	138	NORTH	106	154	95	7.2	NORTH	128	114	147	142	160	SOUTH	103	103	166	WICHITA	117	113	121
-0d		LT CREEK		00	00	0	0		00	WICHITA	0	74.	0	0.0	15	23	75.	0.0	0	43	20	18 19	76.		0.0		SOUTH		1 1	
	Sodfum (Na)	72. SALT		8230	1180	1170	11700		332	NORTH WIC	5320		231	5300	6720	6300		3860	446	4060	2860	3470		377	1450	1540	78.	8810	8870	4830 9530
Mag-	ne- sium (Mg)		124	271	319		979		270	73. 3	209		94	121	231			216	218	231		192		64	122	092				178
2	Cal- clum (Ca)			1010			1200		1300		871				1040					961		830		228						1240 2
	Iron (Fe)																													
	Silica (SiO ₂)		1	9.1	7.5	14	1 4	1	13		11		10	1 9	1.2	œ		8.4	1		0.0	2.3		23	8.8	3.4		91	22	1.5
	Discharge (cfs)		1	2.7	2.22	1.96	1.04		11		5.89		1	1 3	0.00	2.2		13.7	1	4.1	1.9	14.5				b.25		4.23	0 1	1 1
Date	of		1939	1951	1952	1953	1954	1958	1959		1952		1958	1959	May 31, 1966	444444444		10, 1954	1959	966		1967		Sept. 11, 1950	1959	1963		1953	1956	1958
Č	colle		16.	25,	12.	13,	10,	30,	18,		12,		.30,	18,	31, 1	0		21,	19	1, 1	16	3.7		. 11,	19	20,		13,	21.	30,
			Nov.	July	Mar.	Jan.	Feb.	July	July Mar.		Mar. July		July	Mar.	May	July		Feb.	Mar.	June June	July 6	Dec.		Sept	Mar.	Apr.		Jan.	Nov.	July Mar.

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

	Hď		1.8	8.1	7.2	6.8	7.5		7.8	7.8	7.2		5.0		9.7		1	7.5	7.4	7.6	6.7	6.5	7.3		6.4	7.9	6.9	7.0	7.3
Specific con-	duct- ance (micro- mhos at 25°C)		39900	32200 28900 36300	24300	2070	5010		18400	19100	17000	10500	1430	460	10500		9020	5230	4650			4650			3920	4550	3250	3690	3540
	ad- sorp- tion ratio		1 1	41 50	1 1	3.0			1	1 0	1	18	2.5				1		6.5	6.0	11	1112	8.5		9.2	10	8.0	9	8.8
	Non- car- bon- ate		5060		3610	3180	1380 3860		2760	1550	2740	1760	388	348	1820		1	1	400	397	688	704	613		662	724	545	620	595
Hardness as CaCO,	Cal- ctum, Mag- ne- stum		5160			3290 698			2830	1610			921				710	972	834	468	784	776	969				838		684
olids ted)	Tons per day																												
Dissolved solids (calculated)	Tons per acre- foot		38.6	29.1	1 1	1.1			17.1	10	601	ŧ	1	! !	1		1	1 1	3.85	1.88	3.84	3.81	1						
Diss (C:	Milli- grams per liter		28900	21700	16100	11300	3440		12600	1000	11000	6980	921	936	6950		2020	3210	2830	1380	2820	2800	2210		2410	2740	2850	0000	2190
	Bo- ron (B)																1	1	0.31	.39	1 1	1 1	1						
	Ni- trate (NO ₃)		1 1	111	1.1	0.2	121		1	1	1 1	ŀ	0.5	1 10	1		0.0	0 8	c io	φ,	0 10	1.5	1.2		0.0	10	1.5	9.7	1.8
	Fluo- ride t (F) (NJAMIN				0.5		æ	:	ŀ	1 1	E)	0.3	1 10	1		0.3	9.	7.7	÷		1 1	1	Tel.					
	Chloride (CI)	SOUTH WICHITA RIVER NEAR BENJAMIN	6080	11300 10700 13600	7500	4820	1060 7140	RIVER NEAR SEYMOUR	5710	5860	4950	2920	173	200	2900	MABELLE	685	1250	1100	470	1110	1110	820	AR MABELLE	898	1080	1110	2000	810 760
	Sulfate (SO4)	TA RIVER	3410	3040 3000 3750	2690	2310	2940 1170 3070	IVER NE/	2320	2240	2030	1470	370	362	1520	KEMP NEAR	594	774	373	387	716	652	260	RIVER NEAR	616	640	672	210	556 534
	Car- bon- ate (CO ₃)	WICHI						WICHITA B								LAKE KI								WICHITA					
Bt-		SOUTH	110	106 118 136	182	133	88 154	WIC	85	102	167	140	129	110	127	81. L	86	104	106	86	116	888	102	WIC	95	109	109	00.	103
	Po- tas- slum (K)	79.		1	a	5.1	10	80.		I.		13	4.8	18		3.0	10 1	22	5.6		4 4	60	6.2	82	en =	6.9	7.0	0.0	0.0
	Sodium t (Na) s		8890	6330 8330	4490				3550	3490	2920	830	112	140	760		435	762	324 694		794	709	518		573	929	694	100	526 512 477
	Mag- ne- sium (Mg)		394			193 2			207		88	27					44	65	57	25	61	52	42		46	707	52	33	358
	Cal- cium (Ca) 8	1	1420 3			999 1 250 1			792 2		500		17.0				212	198	240	146	248	223	210				250		213 208 204
	Iron cd (Fe) (d	1			-	+											1	100	.02	90.	1	1 1	1						
	Silica In (SiO ₂) ()	-	2.4	8.7	2.0	8.3	8.8 6.6			1	14	9.3	10	m or	0.0		7.2	T 3		10	6.7	100	5.7		6.1	7.3	9.0	6.3	8.08
	Discharge (Si (Cfs)		10	1.65		.4 1			8.0 18	17.6		00		906	63											0	461		150 1340 399
	Date of collection		29, 1949	13, 1953 10, 1954	7, 1965	16	25, 1967 13		20	10, 1954	30,	7. 1966	. 1		23		10, 1939	15, 1942	June 16, 1952	1955.	23, 1964	31	23, 1967		12, 1965	1, 1966	7	11	20
			July	Yeb.	Dec.	July (Jan. June Aug.		To m.	Feb.	July	July	Sept	Dec.	Aug. 23.		Oct.	July	June	Nov.	Oct.	Aug.	Jan.		Dec.	June	July	Oct.	July Aug.

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Dally Stations in the Red River Basin, Texas -- Continued

	нф		7.4				7.7		6.3	6.0	0.0	5.9		7.1	6.8	6.7		6.6		7.4		7.5	7.4	7.6	+ 1		7.5
Specific con-	duct- ance (micro- mhos at 25°C)		281		1020	4760	8120				309			1340	2270	866	118.00	331		4320		3240	2070		2570		4350 4720 2750
S a			0.5		4.2	1	13				1.5					191		5,6		6.8		1 1	18.9	6.0	7.6		9.4
	Non- car- bon-		3 1		99	626	1350				21			122	254	588		35		732		11	362	254	325		696 780 450
as CaCO,	Cal- cium, Mag- ne- sium		114		174	1980	1510		604	75	77	1660		194	368	156	040	293		825 651		548	305	350	435		816 964 545
solids ted)	Tons per day																										
Dissolved solic (calculated)	Tons per acre- foot								3,39	23	.22	8.85		0.94	1.62	.61	07.9	1.19		3.54		11	1.78	1.46	1.97		3.45
Dis.	Milli- grams per liter		159		522	2720	4560		2490	166	163	6510		688	1190	448	0007	177		2600		1680	1310	1070	1450		2850
	Bo- ron (B)																					1 1	0.35	1	1		1.15
	Ni- trate (NO ₂)		0.8		2.2			1	0.0	1.5	ic a	1		6.2	-	100		3.0		1.5		1.5	8.5	4.	1.0	BYERS	3.5
	Fluo- ride t (F) (NO	0.2	A	0.3			PARK	0.3	. 2	03.00	1	RK	0.3	ei e	10,1		6, 6,	FALLS		LS.	6.0	ώ ci π	. 10	7.	NEAR B	
	Chloride (C1)	NEAR VERNON	5.3	AR ELECTRA	243	1500	2750	NEAR IOWA PARK	1460	64	31	3900	R IOWA PARK	355	632	199	0.5	430	WICHITA FA	1000	WICHITA FALLS	775	545	450	290	ROAD 171 N	1080
	Sulfate (SO.4)	ROSA LAKE	22	CREEK NEAR	11	1 2	38.33	BUFFALO CREEK		8.4	9.8	137	CREEK NEAR	23	40	122	0.11	30	RIVER AT W.	629	AT	264	101 239	175	270	AT FARM R	486
ě	ate (CO ₃)	SANTA RC		BEAVER				UFFAL											100		WICHITA					RIVER A	
	car- bon- ate (HCO ₃)	SAN	138		132	906	200	NORTH B	67	54	91	13	BUFFALO	88	139	120	263	164	WICHITA	113	LAKE	74	104	118	134		146 226
	tas- slum (K)	83	8.8	84	2.9		27.7	85. NO	22	32	30	6,6	86.	88	513	111	4.0	4.8	87.	614	88.	407	21.8		63	WICHITA	615
	(Na)		112		28	25.0	1170		7			870		1	m			35		480		44	304		672	89	9 9
Neg	mag- ne- sium (Mg)		7.2		12		168 11		20	0.9	6.0	-		18	33	133	2	30		61		41	365		33		84
_	Cal- clum (Ca)		324		50	200	330		160	20	20	425		48	93	24.0	000	18		230		152	120	101	120		248
	(Fe)																					0.10	.03	1	;		
	Silica I		6.0		7.8	10	1.00		6.7	7.3	9.6	5.3		6.4	4.0	7.50		5.8		12 6.5			0.8.6		8.2		10
	Discharge ((cfs)					2	2.20		0.15	309	2, 13, 1	. 95				39.1		1000		354 1							-
7	Date of collection		21, 1966,		1966		7, 1967		1961	1962		3, 1963		1964		Sept. 23	1000	7961		12, 1951 14, 1966		1944	e 6, 1946 24, 1952	1959	- I		1949
	col		27.	-	. 26.	NO	100		3	e 19,	e 19.	6, 1		12,		t. 23		2 2				. 19,	e 6,	. 11			y 21.
			May Nov.		Apr.	Jun	Jan. Feb.		Nov.	June	June	May 6		Feb	May	Sep	OWE	June Apr.		Oct.		Oct.	June Mar.	Mar.	June		July Dec.

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

California Process Station Part California Part Part Part Part Part Part California Part									Bit							-	(calculated)	ted)	as CaCO,	as CaCO,	8	Specific	
State	Date of collection	Discharge (cfs)		Tem-en-	Cal- clum (Ca)	Mag- ne- stum (Mg)	Sodfum (Na)		car- bon- ate (HCO ₃)	113		Chloride (C1)		N1- trate (NO ₂)		rams per iter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	3.500	1		Hd
1.5 1.5								90.	LAKE														
1.5 1.5	2 1040			000	1.0	*	1.1		00			:		1		116			0.4				1
1.0 1.0	0, 1952			.00	33	- 11	25	. 8	176			18			0.07	197			128	00	15.	335	
1.5 1.5	5. 1954			. 0.4	24	8.4			139			1.5	. 2		00.	153	.21		94	0	1.0	274	
1.5 1.5	1957			20.	32	8.0	63.6	nu	188			19	4.4	ci c	i	209	-27		120	0 0		389	
1.0	1304.			1	0 %	0.0	7	0	1.33		1	10	*	N	ĺ	200	7.		108		1.0	3/1	
1.6 1.6			0 0	į	26	00 0	62.3	6	151		13	33	7.	64 6	!	200	.27		101	0	1.7	379	
1.5 2.0			0.6	1 1	30	10.0	. 4	4 45	176		14	000	4.4	N C	1 1	234	. 32		116	00		432	
12 14 77 20 363 88 640 677 0.7 0.2 1200 1.63 250 77 184 13 2920 7. 12 14 77 20 363 18 46 670 0.7 0.8 1500 2.16 274 184 13 2920 7. 13 14 17 20 363 110 46 870 8. 8 9 1500 2.16 274 184 13 2920 7. 14 15 15 1.8 78 78 78 78 78 78 78	7, 1967		5.0	1 1	31	8.0		44	153		10	32	0.0	Si ru	1 1	194	1 1		105	00	1.4	356	
Photo Phot										AKE CRE	EK NEAR	HENRIE	TTA										
18	1959				69	19	36	23	88		34	099		0.2		1200	1.63		250	178	10	2220	1
12 9.1 1.8 7.8 44 0.8 6.5 0.1 0.8 5.5 0.1 0.8 5.5 0.1 0.8 5.5 0.1 0.8 5.5 0.1 0.8 5.5 0.1 0.8 5.5 0.1 0.8 5.5 0.1 0.8 5.5 0.1 0.8 5.5 0.1 0.8 5.5 0.1 0.8 5.5 0.1 0.8 5.5 0.5		.			:	202	Ď.	2	011		94	078	φ.	0.		0601	2.10		214	184	13	2920	×
12 1.8 1.8 1.4 1.8 1.4 1.8 1.4 1.8 1.4 1.1 1.0 1.1 1.1 1.0 1.1							94.					1000	AR HEN	RIETTA									
	1, 1959		12		9.1				44		0.14		0.1			61	0.08		30	0.0	0.6	182	7.6
							95.				ICHITA	100		NRIETT	V.								
	1959	;	17			-			34			10	0.0	0.8		57	0.08		24	c	0.8	00	1 4
10.25 12 14 3.9 14 14 15 10 14		1 6	16		14			00 0	99			22	23.0	10.0		109	.15		50	00		176	
1.88 12	0, 1904	22.01	200		14			5.4	60			20		8.6		00	07.		5.0	0 6	1.1	181	
State	0	1.88	12		16	4.9		9	72			17		1.5		112	.15		60		6.	206	
Secondary Color		302	7.7		15	3.5		97	40		4.8	49	0.	2.8		129	.18		52	19		257	
	, 1965	641	6.8		10	2.9			43		6.2	19		io i		84	111		37	20		146	
273010	N.G	.32	9.6		12	3.9		12	64		5.8		N ES	o.		84	. 11		46	0 0		141	4 4
2730 — 10	4, 1966	606	5.1		5.0	643			22		9.9		.2	ů.		42	;		22	ΰ		99	
2730 6.8 6.5 2.2 4.8 3.9 28 1.8 6.9 .2 .2 147 25 2.2 17 .2 170 85 2.2 2.2 1.4 9.1 6. 5 9.5 5 14 6.5 85 3.2 1.4 9.1 6. 5 9.5 5 14 6.5 85 3.2 1.4 9.6 5. 5 9.5 6.2 1.2 1.2 1.2 1.2 1.2 1.2 1.2 1.2 1.2 1	7		10		11	n		4	46		5.0		.3	2		83	1		43	10		145	
15.0 35 17 19 45 1.2 215 17 19 15 2.3 465 232 56 2.3 862 7. 150 3.5 4.4 9.6 5.5 2.5 465 232 56 2.3 862 7. 150 3.4 50 13 38 6.3 164 28 74 0.5 0.2 294 178 44 1.2 552 7. 150 3.5 150 3.5 150 5. 150 3.5	0	c	8.9		0.9	N C		4	200		8.7			Ņ.		170	i		0 40	NE		919	
564 8.6 7.5 3.1 6.7 6.0 35 4.4 9.6 5 7.5 66 31 3 .5 105 6. 3.4 50 13 38 6.3 164 28 74 0.5 0.2 294 178 44 1.2 552 7. 15.0 35 17 19 45 307 61 35 0.7 1.2 425 0.58 270 18 1.2 686 7. 64 31 65 23 41 - 2 256 33 28 81 45 85 27 415 2 294 25 1.4 786 7. 15.0 35 17 19 45 33 28 81 45 88 20 415 2 296 1.4 786 7. 15.0 35 17 19 45 38 32 86 43 77 2 488 66 304 32 1.4 786 7. 16.5 21 76 20 41 2.6 273 85 35 88 2 47 2 294 25 1.4 786 7. 16.5 21 76 20 41 3.9 300 103 50 .8 .2 497 2 288 42 1.6 770 8.			14		62	19.0	80	* . 1	215		27	150	- 67	N IO		465	1 1		232	26		862	X 4
15.0 35 17 19 45 307 61 35 0.7 1.2 425 0.58 272 18 1.2 552 7. 15.0 35 17 65 21 76 25 32 8 6.3 164 1.2 552 7. 15.0 35 17 19 45 273 8 6 13 10.0 WASHITA RIVER AT FARM ROAD 2564 NEAR ALLISON 10.0 WASHITA RIVER AT FARM ROAD 2564 NEAR ALLISON 10.0 10.0 WASHITA RIVER AT FARM ROAD 2564 NEAR ALLISON 10.0 10.0 10.0 10.0 10.0 10.5 0.2 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 786 7. 1.4 785 7. 1.4	1967				7.5			1.0	35		4.4	0.00	5			99	;		31	62		105	
15.0 35 17 19							98	FARI	10000	1.70	SERVOIR	NEAR	DCONA										
15.0 35 17 19 45 307 61 35 0.7 1.2 425 0.58 270 18 1.2 686 7. 65 23 32 86 43 49 42 66 25 23 3.6 27 35 35 35 35 35 35 35 35 35 35 35 35 35	1967		3.4		20	13	38	6.3			28	7.4	9.5	- 4		294	;		178	44		552	7.3
15.0 35 17 19 45 307 61 35 0.7 1.2 425 0.58 270 18 1.2 686 7. .64 31 20 25 32 32 32 1.4 786 7. .4 65 23 256 85 35 .5 .0 415 256 46 786 77 .65 21 76 20 41 256 85 35 .5 .0 415 294 25 1.4 755 77 .61 25 35 328 32							100.	WASH		AT		6.1		ALLISC	N								
. 64 31 89 20 33 332 86 43 .7 .2 488 .65 3.1 4 789 7	16, 1965	15.0	35		17	19		15	307		61	35	0.7	(*)		425	0.58		270	18	1.2	686	7.
. 61 25 80 23 56 3.5 328 81 45 .8 .2 497 294 25 1.4 755 758 30 76 24 61 3.9 300 103 50 .8 .2 497 288 42 1.6 770 8.	7 1966	. 64	-		88	02		00	200		99	43				400	00.		986	46	7.7	736	
. 61 25 80 23 56 3.5 328 81 45 .8 .2 476 294 25 1.4 755 758 30 76 24 61 3.9 300 103 50 .8 .2 497 288 42 1.6 770 8.	17 1967	. 65	- 51		76	20	41	1.6			85	35	in	0.		415	1 1		272	48	1.1	899	1
.58 30 76 24 61 3.9 300 103 50 .8 .2 497 288 42 1.6 770 8.	21.	.61	25		80	23	56				81	4.5	8.	ç		476	1		294	25	1.4	755	7.
200 000 000 000 000 000 000 000 000 000	7	. 58	30		7.6	24	61				103	20	×	6		497	1		288	42	1.6	770	8.2

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

,	Н		7 . 7	7.9	7.0	7.2	7.3		6.3	7.0	7.1	7.4	6.5	7.8	6.7	7.6		8.3	8.8	8.1	0.0	8.9	8 8	7.7	8.4	8.0	8.2	80 0			8 8		6 7
Specific	duct- ance (micro- mhos at 25°C)		085	351	550	431	438		218	515	679	948	351	877	300	148		1980	1550	1580	372	1850	1730	1590	1350	1510	1100	513	1390	1500	1610	1	118
- So	ad- sorp- tion ratio		1 3	1.0	1.9	9	1.3		0.4	6.	1.0	1.8	1.0	1.9	7.	.3		4.3		3.0	1.3	5.2		4.4			2.8	1.3	4.4	3.4	6.5		4 0
co,	Non- car- bon- ate		c	12	0 ;	10	2 0		11	62	99	88	26	74	23	10		350	230	130	30	282	266	248	200	229	121	72	190	219	230		1.0
Hardness as CaCO,	Cal- clum, Mag- ne- stum		190	101	168	212	145		85	208	274	296	110	274	107	64		495	340	352	118	395	395	360	320	355	315	174	295	340	345		40
lids d)	Tons per day																	8430	9370	13770	9430	103200	14320	12680	9480	12180	1220	4730	4180	4100	6630		
Dissolved solids (calculated)	Tons per acre- foot								0.17	.45	. 57	ì	i	1 1	1 1	1		1.66		1.39	.34	1.60	1.50	1.37	1.15	1.27	.89	.43	1.28	1.32	1.43		01.0
Diss (CI	Milli- grams per liter		310	179	321	287	255		124	334	422	532	198	515	172	89		1220	980	1020	253	1180	1100	1010	848	936	657	314	945	974	1050		20
	Bo- ron (B)																																
	N1- trate (NO ₂)		0.2	0.	oj c	5,0	£1 00		1.0	0,	0.0	Ç.	21.0	0 10	1.8	2.0		2.0	1:1	66	2.3	. 23	1.0	0 +	1		ŀ	L	1 1	11	11		0 0
	Fluo- ride (F)	ЮГРИ	0.5	ın o	i, a		9.	YTA	0.3	Si c	. r.	e.	٠. د	, 0,	4.5	4.	LY.																0
	Chloride (Cl)	NEAR RANDOLPH	23	16	24	12	24	NEAR CHICOTA	5.5	28	36	127	39	110	4.9	4.5	ARTHUR CITY	400	310	310	38	372	335	308	110	295	175	70	260	290 182	322	AR PARIS	0 9
	Sulfate (SO.)	CREEK	27	4	46	32	5.0	CREEK	22	82	93	77	46	91	32	9.2	RIVER AT A	258	192	218	43	262	32	225	195	210	122	140	195	215	225	CROOK NEAR	10
500	bon- ate (CO ₂)	D'ARC						SANDERS									RED RI															LAKE	
Bi-	car- bon- ate (HCO ₃)	BOIS	268	109	254	247	174 218	04. 8/	90	178	254	254	65	244	102	73	05.	176	134	144	108	138	158	136	120	154	236	124	128	152	140	106.	25
Do		103.				2.6	2.4 4.4	10	3.4	•					3.2		1																9.9
	(Na)		41	24	90	20	36		8.1	55	38	73	23	72	5.0	4.7		222	188	198	33	237	30	193	175	187	116	39	175	191	235		6.3
Mag	ne- stum (Mg)		1		*		3.1				7.6		6.6	23	2.3			51	34	26	6.8	30	32	31		1	ŀ	11	I	11	11		1.3
}	Cal- ctum (Ca)		72	36	200	81	53		30	73	16	80	33	72	34	22		114	81	98	36	109	105	94		1	F	1 1	-{	11	1 1		14
	Iron (Fe)																																
	Silica (SiO ₂)		8.2	2.9	9.4	10	5.6		7.5	14	112	17	15		8.6	5.1																	3.2
	Discharge (cfs)					39.0	.45		3.9	1.6	5	7.	6.2	.23	9.71	390		2560	3540	5000 c4820	13800	32400	4820 10800	4650	24200	4820	689	5580	1640	1560	2340 2180		
	of collection		1965	1966	1967	1	126		29, 1961	6, 1962	May 25	29, 1965	14, 1966	25, 1967	Apr. 7	11		1, 1961	4, 1962	21	: :						_	26	21	June 18	t. 9		Mar. 18. 1960
			Nov.	Mar.	Ann.	May 1	July 12 Sept. 6.		Nov.	Feb.	May	Nov.	Mar.	Feb.	Apr.	July		Nov.	Jan.	Feb.	Apr.	June	Aug.	Oct.	Nov.	Jan.	Feb.	Mar.	May	June July	Aug. Sept.		Mar.

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

		s in mill														Dis	solved so	olids	Hardi as Ca	ness	So-	Specific con-	,
	Date of collection	Discharge (cfs)	Silica (SiQ ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	Bi- car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)		Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	pН
			_					107	. BI	G PINI	CREEK N	VEAR MANC	HESTE	R									
			. 0		22	5.3	9	0	70		36	20	0.2			158	0.21		79	22	1.0	255	
	25, 1962	2.0	9.6		23 20	4.4		8	42		47	17	. 2			150	.10		68 35	34 19	. 9	218 122	
	27, 1963	500	5.8		10	2.5	8.1	2.8	20		25	10	.1	- 5		75	.10		- 50				-
								10)8. P	ECAN I	BAYOU NE	AR CLARKS	VILLE									0.07	
	17, 1967	0.2	6.1		24	2.7	15	4.8	50		36	21	0.0			135			71 51	30 15		237 181	
	25	6.1	1.1		16	2.8		5.0	44		23	16 15	. 1			98 138			88	2		250	6
pr.	8	616	6.1		29 5.	3.7 8 1.5		2.9	105		7.2	5.4	. 0			44			21	4	. 4	72	6
pr.	27					40 3000		1	09. F	ED RI	VER NEAR	NEW BOST	ON										
_					101	30	225		122		249	358		1.3	3	1110	1.51		375	275			
	1, 1960				106	33	215		140		249	350		8		1130	1.54		400 320	286 230			
	23				94	21	184		110		187	308		1.2		849 972	1.15		380	262			
	31, 1961				110	26	205		114		219 99	340 145		1.6		493	. 67		194	100	3.0	822	1 8
ar.	. 11				55	14	96					250		1.9		796	1.08		288	188	4.1	1280) (
pr	. 12				82	20	159		122 72		172 29	36		2.6		180	.24		9.0	31			
	10				27 54	5.5 13	98		90		110	150		2.		524	.71		190				
	y 26				122	28	228		188		242	360		1.:		1200	1.63		420 360	100000			
	t. 1 t. 20				95	30	227		116		240	360		1.	1	1070	1.46						
-					98	31	226		134		240	355		1.		1090	1.48		370 375				
	t. 27				101	30	203		120		225	342		1.		996 834	1.35		288				2
	. 17				79	22	176		108		185 205	275 385		12	0	1180	1.60		420				
	. 1				115	32	232 187		168 122		200	318			8	1000	1.36		350	250	4.4	1 158	0
)ec	. 6				88	32						265		1.		884	1.20		322	206	4.0	1400	
Jan	. 4, 1962				85	27	163 98		142		178	150		*0		544	.74		196				
	. 2				58 24	13			76		33	49				166	. 23		84				7
	2				28	3			84		28	26		2.		198	. 27		86 108				
	2				26		27		100		37	32			8	238	. 32						
					111	31	237		136		255	385			0	1220	1.66		405 380				
	e 19				101		210		150		238	330			7	1110	1.51		100				
	t. 11				31				102		28	25 275		1.	0	907	1.23		315				
	. 10				***	-			124		192 155	220		-		742	1.01		270	163	3 3.1	8 117	0
VOV	8					-			130						27	543	.74		216	113	3 2.	9 879	
Dec	5				00.00	-	- 96		126		111 116	146 158			_	593	.81		242				
	. 3, 1963					-			148		191	260			-	868	1.18		326				
Jan							- 104		208		124	174		-	-	687	.93		300				
	27					-	40.00		88		31	32		-	-	208	-28		100				
	. 27						- 144		148		155	240			-	802			308				
	24				- 22		- 171		148		205	270			-	922			344				
	y 22						- 192		142		225	295				992			148				
	ly 18					-	- 64		92		40	120 320		-		380 1030			370		5 4.		
	g. 12					-	- 204		152		228	320			-	1.500							_

See footnotes at end of table.

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas -- Continued

	рн
Specific	duct- ance (micro- mhos at 25°C)
8:	ad- orp- tion atto
aCO ₃	Non- car- bon- ate
Hard as C	Cal- clum, Mag- ne- slum
solids ted)	Tons per day
ssolved	Tons per acre- foot
ď	Milli- grams per liter
	Bo- ron (B)
	N1- trate (NO ₃)
	Fluo- Ni- B ride trate ri (F) (NO ₃) (;
	Chloride (C1)
	Sulfate (SO.)
	bon- ate (CO ₃)
	car- bon- ate (HCO ₃)
ç	tas- slum (K)
	Sodium (Na)
	ne- sium (Mg)
	ctum (Ca)
	(Fe)
	Silica (SiO ₂)
	Discharge Silica Iron Cal- (cfs) (SiO ₂) (Fe) (Ca)
i	Date of collection
	Dissolved solids Hardness Societated) as CaCO ₃ Societated

May 23, 1961	2.3	5.7 0.00	6.4	2.4	11	2.1	35	3.8	9.0		1.0	55
Oct. 23	9.	11	7.5	2.5		13	33	6.0	12		2.8	72
Nov. 30	4.4	11	4.5	1.5		14	25	8.8	12		8.	9
Mar. 16, 1962	18.2	8.3	3.5	1:1	8.0	.7	16	7.2	7.5		0.	44
May 26	. 5	8.6	8.0	2.4		13	44	2.8	14		0.	72
July 7	.5	9.1	8.0	2.4		12	40	4.4	12	2	ı.	69
July 31	1.6	8.9	6.0	1.4		14	33	11	8.0		ır.	99
Sept. 18	6.	12	7.2	1.8	13	2.6	36	6.8	10		.2	7.0

6.38 4.38 6.28

108 143 118 73 131 129 129

1...00 0....

0 64 64 49 60 50

26 28 28 17 17 13 30 30 21 25

0.10 .09 .09 .09 .09

Includes the equivalent of any carbonate (CO3) present.

b Field estimate. c Mean discharge.

Table 7, -- Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma

Column C									BI-						Dis	Dissolved solids	solids	Hardness as CaCO,	CO,	·\$	Specific con-	
March Marc		Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- ctum (Ca)	Mag- ne- stum (Mg)					(SO.)	Chloride (Cl)	Fluo- ride (F)	N1- trate (NO ₃)	 grams per liter	Tons per acre- foot	Tons per day	Cal- clum, Mag- ne- stum	17.63	dium ad- Borp- tion ratio	duct- ance (micro- mhos at 25°C)	
1999 1999							Α.	SALT	FORK		VER NEA	R VINSON	-	3								
15 15 15 15 15 15 15 15	21.				220	56	134		70	0.0	719	185	0.6	4.4	1350	1.84		780	722	63.0	1880	80 0
11 11 11 11 11 11 11 1	26.				230 290 460	92	136 146 161		130 86 64	0000	727 1050	202 215 260	9.7	40.01 0.00.00	1400 1970 2850	2.68 2.68		800 1100	730		2380 3230	7.80
1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1,	pr. 28				112	49	100		152	0	402	108	1	2.8	950	1.29		480	356	63	1270	8.1
1	ay 16				464 404 440 320	122 93 93 51	269 201 180 181		204 124 180 138	0000	1600 1360 1310 969	292 220 255 200	0.111	0.777	3020 2540 2550 1890	3.45 3.45 2.57		1660 1390 1480 1010	1490 1290 1330 897	0,0,0,0	3440 2820 2940 2270	1.00.1
11 12 13 14 15 15 15 15 15 15 15	ct. 18				332 344 296	20 73 78 83	49 191 200 166		232 148 166 70	00000	237 1090 11120 1050	49 205 220 205	1111		645 2110 2180 2020	2.87 2.96 2.75		400 1130 1180 1080	210 1010 1040 1020	-00000	2530 2350 2350	28.72
26. 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	pr. 19 ay 17. une 21. uly 12.				412 160 388 232 352	105 73 78 44	205 205 191 157 107		150 402 124 128	00000	1440 644 1210 670	200 82 192 135	11111	4 4 4 4 4 4 4 5	2640 1410 2370 1340 2050	3.59 1.92 3.22 1.82		1460 700 1290 760	1340 370 1190 655	965-1-	2870 1900 2620 1710	8.7.8
7. September 1. Se	ept. 26. ct. 24. ec. 5.				476 504	100	161 188 177 196 173		116 120 142 60 132		1510 1620 1180 1160 968	200 215 212 225 190	11111	1.9 0.1 2.9 2.9	2660 2800 2240 2120 1860	3.62 3.05 2.88 2.53		1680 1680 1260 1150	1580 1580 1140 1100	-0000	2860 3070 2630 2520 2280	
947 488 158 355 118 0 1690 555 0.5 3300 4.49 1870 1770 4080 -9, 1360 170 488 158 355 118 0 1690 555 0.5 3300 4.49 1870 1770 4080 -9, 136 170 18 184 272 2.0 3080 4.19 56 1970 1870 1350 -30 173 83 108 687 93 1.9 1.0 2020 1520 1010 350 -30 173 83 175 141 1290 232 1.0 2020 1460 1340 1520	ar. 7 ar. 25 pr. 25 ay 23. une 27.				111111	111111	193 203 204 109		96 92 62 166 122 140		1250 1380 1370 772 626 1020	240 220 220 220 295 170	111111	1.4	2420 2530 2550 1680 1370 2020	3.29 3.44 3.47 2.28 1.86 2.75		1300 1360 1360 855 725 1180	1220 1280 1310 719 625 1070	999999	2850 2860 2870 2160 1710 2480	
947 — 488 158 355 118 0 1690 555 0.5 3300 4.49 — 1870 1770 — 4080 555 1360 555 0.5 3300 4.49 — 1870 1770 — 4080 551 136 134 485 100 3.0 3.0 3.0 3.0 3.0 3.0 572 462 — 1350 572 573 573 573 573 573 573 573 573 573 573							В.	SALT	FORK			MANGUM,	OKLA.									
1360 170 36 86 134 485 100 3.0 946 1.29 3470 572 462 1350 947 6.68 568 135 198 123 1840 272 2.0 3080 4.19 56 1970 1870 1350 -30 1733 243 35 18 16 110 22 1.0 1.63 5780 754 666 1520 -1 432 152 141 1290 232 1.0 2270 3.09 1460 1340 1460 1340 1460 1340 1460 1410 1460 1340 1520 1460 1410 1520 1520 1520 1520 1520 1460 1340 1520 1540	Aug. 11-20, 1947				488	158	355		118	0	1690	555			3300	4.49	1	1870	1770	1	4080	1
mum., zul. 20, 1947 6.68 568 135 198 123 1840 272 2.0 3080 4.19 56 1970 1870 3500 mmm, rammin, ra	Minimum, Oct. 1-9, 12, 14, 1946 ater year 1948	1360			170	36	86		134			100			946	1.29	3470		462	1	1350	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Maximum, Apr. 21-30, 1947 Minimum, June 21-22,29-30	173			563	135	198		123		1840	272		8	3980	4.19	56	1970	1870		3500	1 1
	7, 1949 7eb. 10, 1950 9ec. 2 7eb. 6, 1951				350 432 459 401 488	83 92 97 111	175 152 169 162 199		136 141 161 167 137		1110 1290 1410 1240 1570	232 232 220 195 255		8 P 4 9 9	2020 2270 2610 2360 2850	3.09	11111		1100 1340 1410 1220 1560		2570 2870 2920 2750 3330	

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Table 7.--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

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	Нф
Specific	duum duct- ad- ance 1 sorp-(micro- tion mhos at ratio 25°C)
ò.	ad- sorp- tion ratio
	Non- car- bon-
Hardness as CaCO,	Cal- clum, Mag- ne- stum
solids	Tons per day
Dissolved solids	Tons per acre- foot
DIS	Milli- grams per litter
	Bo- ron (B)
	Fluo- Ni- Ba ride trate ro (P) (NO ₃) (I
	Fluo- ride (F)
	Chloride (C1)
	(SO4)
į	bon- ate (CO ₃)
Bi-	car- bon- ate (HCO ₃)
ģ	tas- stum (K)
	Sodium (Na)
Mag	ne- sium (Mg)
	ctum (Ca)
	(Fe)
	Silica (SiO ₂)
	Discharge (cfs)
i	Date of collection

B. SALT FORK RED RIVER AT MANGUM, OKLA .-- Continued

8.0	7.8	7.0	7.1	7.6	8.1	× ×	1	0 10	2.5	. 00	7.8	90	7.7	7.6	8	8.0	8.1	7.7	7.8	8 9	0 1	1.1	2.0	1.6	2.8	0	7.9	7.8	7.9		9.6	7.7		2 0	0.0	2 3	7.9	7.3	7.5	7.8	7.6
1480	3250	1560	1090	2830	1740	2090	2000	2540	2580	2740	1660	2270	1970	2920	1120	2570	2590	2610	2300	3010	0010	3000	2330	1470	2850	9590	2070	2910	3500	3100	2910	1940		4400	9010	1500	2890	2750	2100	2650	2480
1.1	2.0	6.6	1.	1.4	5.0	7 7	0	9.0	- 0	2.7	5	2.1	1.6	2.0	1.1	2.4	2.5	2.0	53	2.5	7 . 0	2.0	3.	4.0	1.8	2.9	2.1	2.5	3.0	6.1	000	2.6		4 5	0.0	8.0	3.3	3.3	3.1	3.6	3.6
534	1560	480	443	1480	652	926	1600	1840	0040	1180	580	978	810	1480	390	1160	1100	1190	906	1430	0000	1490	407	165	1390	1100	828	1270	1770	000	022	828		1			868		599	850	746
645	1660	069	620	1600	760	1020	1780	1940	1090	1280	665	1070	925	1560	490	1280	1240	1320	1060	1370					1480		895	1320	1860	0001	1320	950		1-	•					990	
11	į	H	1	1.32	108	1890	914	17.7	526	168	47.2	6,75	42.1	87.8	4300	621	490	486	433	106	201	200	200	1.57	46.3	300	338	221	21.7	190	221	130		80.4	194	434	1490	224	632	95.5	531
1.66	1	1-1	- }									2.65	2.12	3.69	1.15	3.13	3.13	3.10	2.60	3.75	20.00	20.00	1 59	1.58	3.43	2.91	2.27	3.48	4.38	2 40	000	2.19							2.04	2.83	2.60
1220	;	П	1	2440	1330	1720	2820	3280	1950	2400	1250	1950	1560	2710	842	2300	2300	2280	1910	2630	2220	2260	1190	1160	2520	2140	1670	2560	3220	2500	1460	1610		2480	2190	866	2210	2240	1500	2080	1910
1 1	1	1 1	ŀ	0.42	-	1	1	1	. 41	1	. 24	İ	į	.47	1	1	į	.30	1	1	70			1	3	i	1	i	1 1		1	1		0.41	1	ì	.23	. 52	.26	. 53	.47
0.8	1		ļ	2 . 2	2.5	2 2 2	3.7	9.		Ci.	9.	2	7.	1.0	3.7	7.7	2.4	1.6		0.4	0	0.0	10	0 00	2.5	1.3	2.8	1	6.1			1		1		5.8	0	1.0	1.8	1.0	1.8
1 1	1	1 1	Į	0.0	0 4	.7	1.0	6	1	ŀ	1.	1	i	9.	1	1	t	9 -	į	1 1	U	2 1		1	1	1	1	į	1 1	1	1	1	OKLA.							1.0	
212	252	210	18	881	102	205	260	252	182	210	152	178	165	210	78	180	205	210	061	415	200	195	06	106	222	210	155	260	280	280	150	245	CARTER,	200	388	205	350	310	252	305	295
238	1	1 1	1	1440	846	933	1580	1870	1020	1320	619	1030	784	1510	387	1200	1190	1200	1 1 1 1 1 1	1170	1550	1130	530	560	1360	1140	887	1300	1590	1280	720	850	RIVER NEAR	1060	919	371	893	1030	635	947 856	839
			1	00	0.0	0	0	0	0	0	0	О	0	0	0,10	0	0	0 0	0 0	00	c	0.0	0	0	0	0	0	0	0.0	0	0	0	RED RI	0	0	2	0	0	0	00	0
136	123	116	216	144	144	114	220	124	118	118	104	112	140	86	118	701	174	100	1.40	182	144	120	212	132	112	118	82	64	104	128	180	148	FORK F	232	212	178	368	154	172	170	200
2000	100	79	7.8																														NORTH	150	-00		27.0				2.0
-		-		125	149	156	192	213	150	224	129	1.59	110	180	200	120	199	160	666	268	187	152	84	7.9	158	175	146	208	180	236	124	185		359	279	147	260	797	194	242	251
8 8 8	0.0	102	2	6.5	64	99	125	142	7.1	64	45	56	09	88	25	90	7.1	7.5	2 0	112	122	76	39	57	E	1	1	1		1	1	1					121			96	
196	110	216	220	224	276	300	508	544	320	408	192	336	272	180	140	200	380	304	180	364	144	364	204	174	į.	1	í		1	1		1		132	270	42	270	07:	104	250	808
11		i	1	1 1	-	1	00.0			1		1	1		1 1		!	1 1	1	1	1	1	1			1			1	i	1	ţ								11	
1 1		1	1 3	10	;	i i	20	į.	14	1 !	17	1	1:	7.7	1 1		1 2	10	1	1	16		1	1	1	1			1	1	1	1		14		1	21	1.1	17	22	1.1
32.8	7060	5.7	2160	45	300	P407	120	2.0	100	26	14	11,	010	27	100	0 0	7.0	84	30	15	1.4	37	66	.5	6.8	52	12	1000	63.6	b28	584	30		b12	32.9	161	250	ò	156	204	103
16, 1952	3 1955	2	0000	21	14, 1960	. 26	25	18	5	June 20	TZ	July 19	13	17	14	1000	19 1991	13	13	Apr. 18		20	2	July 31,	13	Dec 4	30, 1962	30	Apr. 25	May 22	27	24		Oct. 28, 1959	0		Feb. 25, 1960		2	Jan. 17, 1961	
May 19. Jan. 16 Feb. 10		July	Oct.		-	÷	Feb.		2	91	I y	13	bt.	٠.	1 3						-	10.	Ly	1,5		7				22	1e 2	ot.			7	: 1 :		2	7 T		

See footnotes at end of table.

Table 7,--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

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See footnotes at end of table.

Table 7.--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

- u	pH pH
	duct- ance (micro- mhos at 25°C)
S.	ad- ad- Borp- tion ratto
LCO3	Non- ad- car- gorp- (m bon- ratio mb
Hardness as CaCO,	Cal- cfum, Mag- ne- slum
ollds	Tons per day
Dissolved solids	Tons per acre- foot
DIE	Milli- grams per liter
	Bo- ron (B)
	Ni- trate (NO ₃)
	Fluo- ride (F)
	Chloride Fluo- Ni- E ride trate r (Cl) (F) (NO ₂) (
	Sulfate (SO ₄)
į	bon- ate (CO ₂)
Bi-	car- bon- ate (HCO ₂)
Do	tas- sium (K)
	Sodfum (Na)
Mag	stum (Mg)
	ctum (Ca)
	Iron (Fe)
	Silica (SiO ₂)
	Discharge (cfs)
į	od collection

F. WASHITA RIVER NEAR DURWOOD, OKLA.

Water year 1951																				
Maximum.																				
Jan. 21-31, 1951.	436	ŧ	153	60	63	368	1	337	79	i	1.4	0		200	1100	000	200		4.000	
Minimum, May 2-3.	7850	1	20	17	17	153	00	56	2.1			0.0		000	2000	070	361	!	1330	
Weighted average.	1916	1	78	27	29	185	1	159	37	1	. 8	4 4	478	.65	2470	306	154	1	440	-
Water year 1952															1				200	
Maximum,																				
Dec. 11-20,1951.	233	15	136	53			2	337	9.5		1 3	0 24 0		6.4	2.7.4	0 8 8	000		0000	
Minimum, Nov. 2	1060	1	11	3.3	7.1	22	1	23	4.5	-			70	101	200	900	305	1	1290	~ .
Weighted average.	629	ŀ	7.5	29	43		1	169	46		0	7		0.7	200	1000	2 1	I	45	
Water year 1953									2		4			00.	200	306	101	!	736	
Maximum.																				
June 1-2, 1953	148	!	136	49	103	137	1	431	1.4.4		0 1	1.0		4.0	400		000			
Minimum, May 13	3410	1	32	4.6	12	111	6	17			0.0	2		05.	422	150	978	6.1	1410	~
Weighted average.	518	1	65	19	35	1.62	1	119	40.3	1 1	3.5	- 6	390	2 2 2 2	545	99	100	0.0	232	~
Water year 1954														3	0.10	0.62	100	1.0	282	
Maximum,																				
Sept. 11-14, 1954.	16.2		124	65	123	190	2	432	160		1.6	10		43	AG	277	410	0.0	1000	- 1
Minimum, June 8	7980	1	40	5.8	7.7	110	4	33	9.5		0 0	-		66	2450	200	010	2.7	1020	W -
Weighted average.	1258	ţ	63	16	26	153	1	101	36	1	2.5	+ 67	350	48	1190	222	0 0	ç a	545	~
Water water 1000												8				2	000	. 0	01.0	

1 8 1

7.2

88.1

7.8

1.1 2.5

8.4 8:1

7.8

1.58

2,5 3.9

1.1

1 1

1 11

Water year 1956
Maximum,
July 30, 1956...
Minimum, Oct.1-4,
7-10, 1955...
Weighted average.

8.5 7.9

7.7

8.1

3.9 2.6

5.4

7.9

22.

8.1

1.8 . 6

1.32 .15

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Water year 1955
Maximum,
May 5-10, 1955..
Minimum,
Oct. 1, 1954...
Weighted average.

		œ	00				00		-				- 0	0	
XX	2	1830	268	465			1090	410	787			1500	0801	2770	011
	7	3.7	3	9.			1.3	4	1.0			0 1	011		0.1
		309	14	68			258	20	150			200	000	000	007
100		450	121	200			465	160	337			200	000	220	900
100		594	29450	2850			912	1130	1320			299	2420	0010	010
0,		1,59	.23	.41			1.12	30	.71			1 65	06	7.0	4
200		1170	169	303			824	666	522			1910	0000	531	100
* * *		2.3	2.5	3.2			1.2	1.4	1.9			0.0	10	8 6	
		1	1	1			1	1	1			1			
200		300	8.8	26			06	26	62			100	12	49	
		569	18	2.0			263	42	149			650	86	198	
		4	1	1			10	1	1			;	9	;	
		164	130	162			232	152	229			120	132	175	
		182	8,5	20			99	21	43			86	1.5	41	
		49	7.5	10			49	11	31			63	11	28	
		100	36	50		10000	106	46	84			178	99	87	
		1	ŀ	1			1	1	1			I	1	1	
		188	64550	0000		100	410	1880	934			224	4533	640	
	Water year 1957 Maximum.	Jan. 23, 1957	May 17-20	" FRIEG AVELAGO.	Water year 1958	Maximum,	June 16-20,1938.	Aug. 10-13	Weighted average.	Water year 1959	Maximum,	Sept. 14-17,1959	Minimum, June 1-3.	Weighted average.	

Table 7.--Chemical Analyses of mater From streams at selected sites in the new areas basan, Calabour Continued in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.) Table 7. -- Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

	Hď
Specific con-	ad- Borp- (micro- tion mhos at ratio 25°C)
8	ad- iorp- tion ratio
	Yon- car- bon-
Hardness as CaCO,	Cal- ctum, Mag- ne- stum
olids	Tons per day
Dissolved solids	Rilli- Tons grams per per acre- liter foot
and .	
	Bo- ron (B)
	N1- trate (NO ₂)
	Fluo- ride (F)
	Chloride Fluo- Ni- (C1) (F) (NO ₂)
	Sulfate (SO ₄)
	Car- bon- ate (CO ₅)
Bi-	car- bon- ate (HCO ₃)
	Po- tas- sium (K)
	Sodium (Na)
8	Mag- ne- slum (Mg)
	Cal- clum (Ca)
	(Fe)
	Silica (SiO ₂)
	Discharge (SiO ₂)
	Date of collection

	s			(9 _{tw})		Ì	(HCO)	2					liter	r foot	day	ne- stum	ate n		ratto 25°C)		
					э 4	WASHIT	A RIVER	NEAR	DURWOOI	WASHITA RIVER NEAR DURWOOD, OKLAContinued	Cont	inued									i
Water year 1960 Maximum, Dec. 1-10, 1959.	447	1	188	73		77	336	1	509	92	1	3.2	1200	1.63	1450	770	494	1.2	1560	8.0	
Minimum, May 19-20, 1960. Weighted average.	17200	11	46 94	11 32		18	142	9	45	20	1.1	2.2	229 572	.31	10630	162 366	36	9. 6	376	4.1	
Water year 1961 Maximum, Mar.17-24, 1961.	953	9.0	114	#	89	2.6	248	N	250	110	0.3	1.2	1140	1.09	2930	455	248	1.4	1140	8.3	-
Minimum, Dec. 9, 1960 Weighted average.	3040	12	47	9.8	***	27	144	0 2	37	38	ι,	5.0	262	.36	2150	158 348	36	0.1	421 800	8. 1	
Water year 1962 Maximum, Feb. 11-20, 1962 Minimum, June 19.	9760	11	170	10	0.64	91 22 43	296 132	408	495 46 213	98 22 49	111	111	1190 221 581	.30	1600 5820 2110	705 140 363	456 32 195	1.5	1500 345 840	8 8	
Water year 1963 Maximum, July 23-28, 1963			1	1	ĭ	109	142	0	670	130	1	1	1450		316	760	9 644	1.7	1730	8.	
Minimum, Oct. 28-31, 1962 Weighted average.	5510	11	47 95	12 37		15	122	00	63 269	23	1.1	11	260 661	.90	3870	166 390	5 66	1.2	412 938	8 8	
Water year 1964 Maximum, May 1-3, 1964 Minimum, Aug. 18. Weighted average.	254 722 340	111	111	111	A	105 17 42	124 124 144	œ o o	695 37 207	138 15 54	111	111	1440 190 552	1.96	988 370 507	805 124 324	690	1.6	1760 328 777	8 8 8 4 6 8	
						G. CL	EAR BOC	GY CRI	SEK NEA	CLEAR BOGGY CREEK NEAR CANEY, OKLA	OKLA.										

	-	400		0.0	010	0	10		0 0	973	0 27	102	200	96	7.0	461
19 1961	139	95	1.3	233	212	0	17		0.2	213	0.0	404	000	2		
	1,65	5.1	2.1	39	214	0	21		7.	324	.44	56.9	212	37	1.2	573
	200			26	25	0	20		-	274	.37	203	162	34	1.2	468
	D214	99	0.7	20	001	0	2 0			000	0	000	2000	011		071
2 1962	102	83	26	56	320	0	30		0.	183	90.	133	350	00		100
4	258	1	;	26	206	4	31		1	285	.39	198	196	21	œ	467
0	194			39	140	0	33		1	261	.35	87.4	144	30	1.4	457
	000			0	9.4	0	13		1	110	.15	202	88	7	4.	208
16 20	000			5.6	190	0	11		1	388	. 53	14.7	208	52	1.7	899
	h919		1	11	166	0	11		1	210	. 29	120	146	10	7.	332
1	64	1	1	8.5	100	0	8.6		!	138	.19	23.8	88	9	7.	211
100	134	1	1	25	194	÷	25	37		290	.39	105	190	24	8.	463
	879	1	-	18	154	2	20		1 1	229	.31	543	152	22	9.	370
9 1963	173	1		36	176	0	34		*	301	.41	140	186	42	1.1	530
31	7.8	1	1	20	204	0	3.4		1 1	378	.51	9.62	220	53	1.5	699
200	15	1	-	57	170	2	33		1	357	. 48	54.0	196	23	1,8	643

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Table 7.--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

(Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

					Mag		Do.	Bi-	Con						Dis	solved s	olids	Hard as Ca		So-	Specific	
	Date of collection	Discharge (cfs)	Silica (SiQ ₂)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (C1)		Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	
						G. CL	EAR BO	OGGY CI	REEK N	EAR CAN	EY, OKLA	-Cont	inue	ı								
	26, 1963	107		22		30		196	0	30	47				290	0.39	83.8	192	31	1.0	504	8
	23	68						-	===						309	. 42	56.7	194	37	. 9	504	8
	21	58		26	22	53		136	4	30	86				319	. 43	50	156	38	1.8	537	- 8
	18	26 14		27 30	24 28	63		144	4	22	112				347	. 47	24.4	168	43	2.1	605	- 8
				30	28	61		166	4	17	116				358	. 49	13.5	190	47	1.9	649	8
	12	21		38	28	64		190	4	17	122				395	. 54	22.4	212	50	1.9	697	8
	. 9	3.5		54	28	66		228	8	15	124				443	.60	4.19	248	48	1.8	7.57	- 8
	5	8.6				W					52									-	631	- 5
	18	9.9									58					-04.00				50.00	658	
	30	9.2			07-04	62		262	4	23	140				480	. 65	11.9	308	86	1.5	878	8
	6, 1964	10		10.00	100.00	28		270	0	27	65				379	. 52	10.2	280	59	.7	639	
	20	11									67						10.2				527	
	28	8.6				-					82					int on					689	
	6	20			-	38		196	4	21	76				322	. 44	17.4	214	47	1.1	563	8
eb.	12	17		-	2000	-					81										648	6
eb.	17	14				-					80											
	24	11						7.7	22	-	137									99.40	675	
	2	11				46		162	4	35	95				200	==	10.0	010			858	100
	16	52						102		33	46				366	. 50	10.9	210	70	1.4	644	8
	20	666				11		108	0	21	21				192	26	245	120			507	
0.31	24	0.0							- 100						132	.26	345	116	27	. 4	284	7
	31	90 31				-		00.00			30				~~						422	
	6					1.0					80										612	
	14	1650 89				16		176	0	20	27				233	.32	1040	168	24	. 5	393	- 8
	21	50				-		200 000			34							401,000			470	
								(40.00)			86				44-46	Am. Am.					693	
	1	34									64								-		624	
	7	29				32		260	O	33	80				445	.61	34.8	292	79	. 8	709	8
	11	2670		H	-	13		136	0	15	18				159	.22	1150	124	12	. 5	293	- 8
	15	326						-			17					***	PF 90				341	
	22	68						-			26									400.00	442	
	28	50		-	200	-		27.00			37						-				497	
	10	28			-	24		242	0	31	40				328	. 45	24.8	234	35	. 7	530	8
	18	2700								7.0	7.2						21.0				129	0
	19	3380		-	A 100	3.2		60	0	7.0	6.4				74	.10	675	58	9	. 2	123	7
uly	13	6.1				25		212	0	20	43				292	.40	4.81	200	26	. 8	496	8
uly	28	1.1		~~		27		216	0	21	60											
	5	. 2						210		21	70				308	. 42	.91	224	47	- 8	552	8
	19	. 1		-		37		222	0	23	72				337			226	4.4	1 1	589	-
	25	5.1									103				331	. 46	. 09	226	44	1.1	606	7
	. 3	29									21										656	
	. 14	3.9																			301	-
	. 23	2940		300.00		91.40					30										391	35
	. 29	622				10		0.0			6.0										202	
che		022				10		98	0	10	20				148	.20	249	96	16	.4	246	7.

See footnotes at end of table.

Table 7. -- Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahomu--Continued

2	bH DH
Specifi con-	ance (micro- mhos at 25°C)
- os	ad- Borp- tion ratio
	Non- car- bon- ate
Hardness as CaCO ₃	Cal- cium, Mag- ne- stum
Dissolved solids	Tons per day
solved s	Tons per acre- foot
Dis	Milli- T grams per per ac
	Bo- ron (B)
	NI- trate (NO ₂)
	Fluo- ride (F)
	Chloride Fluo- Ni- BG (Ci) (F) (NO.) (F)
	Sulfate (SO.4)
	Car- bon- ate (CO ₃)
H.	car- bon- ate (HCO ₃)
	Po- tas- sium (K)
	Sodium (Na)
	Mag- ne- stum (Mg)
	Cal- cfum (Ca)
	Silica Iron (SiO ₂) (Fe)
	Discharge (S) (cfs)
	Date of collection

KIAMICHI RIVER NEAR BELZONI, OKLA.

1001	h 294		4.2	3.2	3.4	22	0	7.2	4.3	0.0	44	99.0	38.5	24		65	
10, 1901	1000			1	7 6	24	0	6.6	6.1	1.7	63	60.	22.6	20		.7	
	100				0 00	16		7.0	4.6	œ	43	. 90	322	15		9.	
4 4000	000000		1 7		0.0	16	0	8.2	3.51	7.	42	90.	236	14	_	6.	
Anr. 2, 1902	2910	1 1	2.2	1.6	5.3	16	0	6.2	2.6	. 7	5.1	.07	401	12		7.	49 7.3
	1540		4.6		6.2	24	0	6.2	3.0	9 .	56	. 08	233	1.7	0	.7	
	264			9	6.0	22	0	6.6	3.6	57	44	90.	31.4	1.7	0	. 6	
ne 20	2 2		6.4	1 0	4.6	36	0	5.2	6.2	1.0	61	. 08	.87	24	0	.8	
	990			6 6	. 00	26	0	6.0	5.4	1.0	51	.07	32.5	18	0	8.	
8	127		2.4	1.0	6.0	18	0	6.0	5.6	1	41	90 '	47.3	16	-	9.	
	1200		00	1.0	er er	16	0	6.2	5.6	1	46	90.	222	16	3	9.	
	200		000	1	0 9	16	0	6.4	6.6	1	44	90.	46.6	16	3	9.	
n. 4, 1995	200			if	200	20	0	7.8	7.4	1	41	90.	21.5	16	0	0.	
0. 31	700					2.4	0	9.5	8.6	;	20	.07	12.4	2.4	4	8.	
28	3050	17.0	4.0	1.5	0.0	16	0	3.8	6.3	1	51	.07	420	16	ಣ	9.	
	010		1	n.	6	9.4	0	7.0	7.7	3	46	90.	26.1	20	0	90	
23	633			0	6.7	24	0	5.8	6.4	1	47	90.	8.00	20	0	9.	
	7.3		7.9	1.0	6.2	22	0	7.4	7.1		21	.07	10.0	22		. 6	
1 19	30		1.0	1.9	3.0	91	0	5.4	4.4	1	52	. 07	4.21	18	21	ņ	
	9.4		3.2	1.9	8.8	20	О	4.0	4.2	1	49	. 07	1.24	16	0	i,	
			3.5	9 4	r.	20	0	4.3	6.0	1	48	.07	.64	18	2	ic.	

a Flow shown for maximum, minimum, and weighted average is mean discharge for the period. b Field estimate.